

## **FINAL REPORT**

# ASSESSMENT OF CURRENT PIPELINE FLUSHING AND DECOMMISSIONING REQUIREMENTS - RESEARCH AND FIELD TESTING

"Flushing Phase A"

## RFP# CBD SOL 1435-01-99-RP-31018

## March 25, 2001

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## ASSESSMENT OF CURRENT PIPELINE FLUSHING AND DECOMMISSIONING REQUIREMENTS - RESEARCH AND FIELD TESTING (Flushing Phase "A")

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Final Report Date: March 25, 2001

**1.0 INTRODUCTION** 

2.0 OBJECTIVES

**3.0 PROCEDURES** 

4.0 BACKGROUND AND ASSUMPTIONS

5.0 RESULTS AND OBSERVATIONS - 2820

6.0 RESULTS AND OBSERVATIONS - 2822

7.0 RESULTS AND OBSERVATIONS – 2823

8.0 RESULTS AND OBSERVATIONS – 2824

9.0 RESULTS AND OBSERVATIONS – 2826

10.0 RESULTS AND OBSERVATIONS - 11513

11.0 RECOMMENDATIONS AND CONCLUSIONS

12.0 APPENDIX – CORROSION INHIBITOR INFORMATION



## List of Figures

Figure 1 – Figure 2 – Figure 3 – Figure 4 – Figure 5 – Figure 6 – Figure 7 – Figure 8 – Figure 9 –	Pipeline Sample Locations Schematic – Segment 2820 Gas Composition (Major Constituents) – Segment 2820 Gas Composition (By Mol%) - Segment 2820 Flushwater Composition - Segment 2820 Chloride and Sulfate - Segment 2820 Iron Concentration - Segment 2820 Oil and Grease vs. Flush Volume - Segment 2820 Oil and Grease vs. Pipeline Flush Volume - Segment 2820 Pipeline Sample Locations Schematic - Segment 2822
Figure 10 –	Flushwater Composition - Segment 2822
Figure 11 –	Chloride and Sulfate - Segment 2822
Figure 12 –	Iron Concentration - Segment 2822
Figure 13 –	Oxygen Concentration - Segment 2822
Figure 14 –	Nitrogen Concentration - Segment 2822
Figure 15 –	Dissolved O2 and N2 - Segment 2822
Figure 16 –	Oil and Grease vs. Flush Volume - Segment 2822
Figure 17 –	Oil and Grease vs. Pipeline Flush Volume - Segment 2822
Figure 18 –	Pipeline Sample Locations Schematic - Segment 2823
Figure 19 –	Flushwater Composition - Segment 2823
Figure 20 –	Chloride and Sulfate - Segment 2823
Figure 21 –	Iron Concentration - Segment 2823
Figure 22 –	Pipeline Sample Locations Schematic - Segment 2824
Figure 23 –	Gas Composition (Major Constituents) - Segment 2824
Figure 24 –	Gas Composition (By Mol%) - Segment 2824
Figure 25 –	Flushwater Composition - Segment 2824
Figure 26 –	Chloride and Sulfate - Segment 2824
Figure 27 –	Iron Concentration - Segment 2824
Figure 28 –	Oil and Grease vs. Flush Volume - Segment 2824
Figure 29 –	Oil and Grease vs. Pipeline Flush Volume - Segment 2824
Figure 30 –	Pipeline Sample Locations Schematic - Segment 2826
Figure 31 –	Gas Composition (Major Constituents) - Segment 2826
Figure 32 –	Gas Composition (By Mol%) - Segment 2826
Figure 33 –	Flushwater Composition - Segment 2826
Figure 34 –	Chloride and Sulfate - Segment 2826
Figure 35 –	Iron Concentration - Segment 2826
Figure 36 –	Oil and Grease vs Pipeline Flush Volume - Segment 2826
Figure 37 –	Oil and Grease vs. Flush Volume - Segment 2826



## 1. Introduction

The abbreviated name for this project is "Flushing Phase A." The purpose of this study is to assist the MMS in assessing -- and if necessary, scoping and preparing -- regulations for the flushing, handling, and possible reactivation of out-of-service pipelines. This project focused on pipelines that have been taken out of service, but have not been flushed and filled with inhibited seawater. In keeping with this purpose, WINMAR has: reviewed current regulations for temporarily taking pipelines out-of-service lines, reviewed current practices for taking pipelines temporarily out-of-service, and reviewed practices, tools, and technologies for flushing and preserving out-of-service lines. WINMAR also assessed the effectiveness and risk/safety of the tools and practices, Finally, WINMAR performed field tests (offshore in-situ) to assess the condition of 5 out-of-service pipelines.

In a future project, already awarded to WINMAR Consulting, we will assess the condition of pipelines that have been flushed and filled with inhibited seawater. This future project is called "Flushing Phase B" to be completed in 2001-2002.

The project methodology for Flushing Phase A was carried out in a number of phases, as detailed below:

- 1) **Identification Phase**: The first step in this phase was a review of current regulations and practices for pipeline decommissioning and reuse -- temporary and permanent abandonments (MMS). This covered any existing regulations and/or recommended practice for out of service pipelines.
- 2) Interaction Phase: This phase was performed concurrently with Phase 1. Because Winmar has an excellent working relationship with the majority of the contractors in the Gulf of Mexico, we met with them to investigate pipeline decommissioning effectiveness, and the effects of time and the offshore environment on out-of-service pipelines. Contractors included:
- Platform and pipeline owners and operators
- Pipeline pigging and maintenance contractors
- Pipeline corrosion and corrosion inhibitor companies
- 3) Assessment Phase: The thrust of this phase was to assess how well outof-service pipelines fare in the marine environment - over time - for later use. Specifically, we assessed the risks to the environment, and health and safety of operations, for the different pipeline types and varying time the lines were out of service.

To aid in the assessment, a qualitative risk analysis was used to form a reuse matrix based on a number of factor. The factors used were: pipeline product, presence of



H2S, CO2, and of course age The matrix was used to compare the pipeline samples retrieved from offshore in order to grade them in condition.

This project assumed that external corrosion protection techniques were continued during the pipeline's temporary abandonment stage. This later proved to be a good assumption as the pipeline samples recovered showed little to no external corrosion.

4) **Data Gathering Phase:** This phase entailed gathering information during pipeline decommissioning, in order to gauge the effectiveness of the regulations/guidelines which were determined during the Assessment Phase.

Because Winmar decommissions pipelines which were formerly out-of-service, we had the opportunity to actually examine the pipelines in-situ, and assess their condition. Since we know the age of the pipelines tested, and when they were taken out of service, we were able to draw MANY valuable conclusions. Data acquired consisted of:

- Catching and sampling the fluids that were in the out-of-service pipeline. These fluids were sampled at pre-determined intervals, and analyzed for the presence of corrosion products (in the case of fluids) and corrosive properties (in the case of gas). CO2 and H2S was tested for at this time.
- Catching and sampling fluids during pipeline flushing. This test was performed on the pipelines during the actual decommissioning phase. The flushwater was sampled at pre-determined intervals and analyzed for the presence of hydrocarbons, corrosion products, oxygen, and chlorides and sulfates.
- 5) **Recommendation/Conclusion Phase:** At this stage, Winmar has compiled and presented recommendations for regulation of out-of-service pipelines. These recommendations were discussed with MMS pipeline specialists before being summarized and finalized in the report. WINMAR also targeted and recommended specific measures that can improve the safety and effectiveness of temporary abandonment/decommissioning and/or reuse of offshore pipelines.

**Definitions**: In order to avoid confusion, it is important to define "Out of Service" and "Abandoned" as the terms relate to pipelines. The definitions will also be included on future regulatory updates.

**Out-of-Service**: A pipeline that is out-or-service is still connected either at one end or at both ends, but it is not flowing. An out-of-service pipeline may or may not be filled with inhibited seawater. The out of service period begins when the line has not been flowed for 30 consecutive days. Taking a line out of service does not require MMS approval, however notification is required.



**Abandoned**: An abandoned pipeline has been cut at BOTH ends. The line has either been removed, or the ends of the pipeline plugged and buried in-place. Abandoning a pipeline requires MMS approval.



## 2. Objectives

The objectives of this project are many-fold, but to summarize:

1) Provide data to the MMS on the condition of various types of out-of-service pipelines through research and in-field testing. This data includes the composition of any product remaining in the pipeline, the composition of seawater/inhibitor in the pipeline (if present), and the composition of seawater used to flush the pipeline.

2) Assist the MMS in determining if the Out of Service (Shut-in for less than 1 year) pipeline regulations are adequate for ensuring pipeline safety and containment. This objective must be met for the various types of pipelines – treated/untreated, gas/oil/condensate, etc.

3) Assist the MMS in determining if the "Pickled" (Shut in greater than 2 but less than 5 years, flushed and filled with inhibited seawater) pipeline regulations are adequate for ensuring pipeline safety and containment. This objective must be met for the various types of pipelines – treated/untreated, gas/oil/condensate, etc.

4) Collect information through research and field testing to determine the effectiveness of various corrosion inhibitors for the "Pickled" pipelines. Determine if the generic requirement for use of "corrosion inhibitor" is adequate, too strict, or too lenient a term.

5) Gain a general understanding of condition of pipelines on the OCS in the Gulf of Mexico through the collection of out-of-service pipeline samples.



## 3. Procedures

This section of the report describes the field-testing portion of the project (Phase IV). Below is the detailed procedure that was supplied to the contractor prior to any offshore work/pipeline decommissioning.

## A. Offshore Procedures

**General:** Field trip to site will confirm location and work area available to flush pipeline. Brief Field Personnel on flushing procedure. Company procedures are to be incorporated into flush procedure. Confirm location and type of Pipeline End Flanges. Review contingency clean-up plans and fluid disposal with Field Foreman. Check flanged connection for integrity. Check for Check Valves.

- 1. Verify communication link is working between crews at both ends of the pipeline.
- 2. Verify that pipeline is LOCKED and TAGGED OUT and line has ZERO PRESSURE before removing pipeline-end flanges.
- 3. Check pipeline for check valves. Replace if pigs are not able to travel through valves.
- 4. Remove pipeline end flanges and install ANSI 600 Ball valves onto flange ends at both platforms. Close block valves.
- 5. Install all gauges/meters and verify both units have all openings closed and/or plugged.
- 6. Install fill line from pump to flushing head. This line to have an overflow by-pass to divert water overboard and a meter beyond the by-pass in order to know volume of water pumped into line. Flow direction to be controlled with block valves before meter and on overboard line.
- 7. Install pipe discharge line with meter from receiving end to storage/receiving tanks or to production process equipment.
- 8. Hook up Sampling Hose at receiving thread-o-let location.
- 9. Take first Gas Sample using Vacuum tube and Plastic Bag
- 10. Verify pipeline and discharge line at receiving end are open.
- 11. Check flow meter and zero.
- 12. Confirm Production Platform crew is ready to receive water. Open block valve Divert flow from overboard to flushing head using in-line block valves.
- 13. Check pressure gauges to ensure no built up in pressure is occurring at flushing site.
- 14. Check with receiving crew that flow has started.
- 15. Take second Gas Sample using Vacuum tube and Plastic Bag
- 16. Monitor pressure. Do not let pressure build up beyond 1000 PSI. Stop pumping if pressure starts to exceed 1440 PSI.
- 17. Take third Gas Sample at midpoint of Line. Take fourth sample before Flush Water arrives.



- 18. Once fluid returns, capture min. 2 fluid samples. One sample into Mineral Pattern Analysis Bottle and one into Oil and Grease Bottle. Take one more set of samples just before pumping ceases.
- 19. Label ALL sample bottles.
- 20. Open by-pass valve at Well Platform before shutting down pump and then closing block valve located before meter.
- 21. Check and bleed all pressure from fill line and pipeline. Verify zero pressure before removing any piping at either end of pipeline.
- 22. At Well Platform, disconnect pump and fill line. Re-confirm zero pressure and remove flushing head and block valve. Re-install blind flange initially removed from pipeline.
- 23. At Production Platform, remove discharge line. Re-confirm zero pressure and remove receiving hose and block valve. Re-install blind flange initially removed from pipeline.
- 24. Secure samples for shipment to Laboratory. Send field report copies to office.
- 25. De Mob equipment and personnel to shore base.

## **B.** Pictorial Presentation

This section provides a pictorial presentation of how the offshore field testing phase was performed.



Photo #1



The flowmeter reads in hundreds of gallons pumped. It was "zeroed" and calibrated prior to commencing work.



Photo #2 An assortment of flanges were kept on-hand to ensure a good fit-up to the pipeline.





Photo #3

This picture shows the workspread used, as well as one of the well protector platforms. The flushing pump is located on the jackup boat, and a hose connects the pump to the pipeline via a hose that runs across the gangway. Upon close observation, the central facility platform is visible in the background.





Photo #4

This photo shows the top-of-riser sample point at the central facility platform. This location was ideal for taking samples and was used when available. If it was impossible to hook up to the top of the riser (for example, if the riser was removed to the +10 level) then the sampling spool was used (see next photo).





Photo #5

This picture shows how gas samples were taken, to be tested for H2S. The plastic jar shown was filled with gas using the intrinsically safe pump. The length of stain tester was inserted into the plastic jar, and some gas was sucked into the length of stain test tube.





Photo #6

The sampling spool was fitted into the flushing hose – where two hoses were connected. This was done at the platform cellar deck level, between the riser and the water-receiving tank.





Photo #7

Gas Samples were taken using Tedlar bags. These sample bags are the best way to ensure that a good sample has been taken. One can be SURE that the bag is full, as opposed to a steel vacuum cylinder, where it is not obvious/foolproof.





Photo #8

MPA Jars come cleaned, sealed and certified. This photo shows three samples from pipeline 2820. To take a sample, jars are simply filled, and sealed.



Photo #9

The plastic Zero Head Space jars are used for taking samples which cannot have any atmospheric air in them. Once the jars are filled with liquid, they can be purged of air



and sealed. WINMAR used these jars to catch samples for oxygen and nitrogen testing.

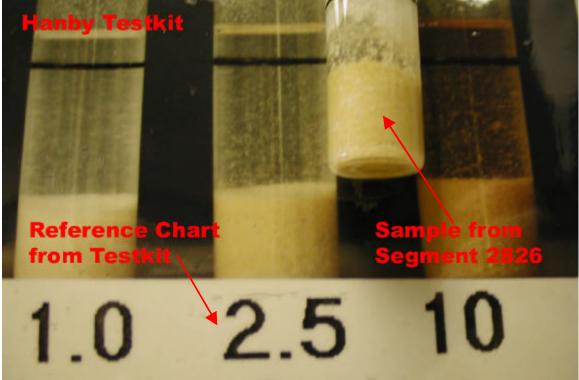


Photo #10

A Hanby Environmental Labs testkit was used as a "quick-check" in the field for the presence of oil and grease. The results from the Hanby kit were very close (to within 5 ppm) to the tests results from the lab.



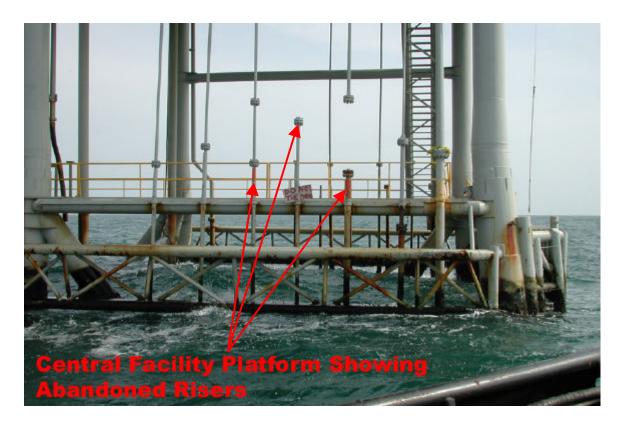


Photo #11



## 4. Background and Assumptions

The following sections summarize the results for the various samples taken. The results are compiled and displayed graphically in order to help interpret and analyze the data. For each pipeline tested, the results are organized into sections. The sections are listed below, along with any assumptions made during the data interpretation:

#### Sample/Locations Observations

The location of the samples was derived by analyzing the amount of fluid pumped at the time the sample was taken. The flowmeter was used to obtain this volume, and the internal pipeline diameter was used to convert this volume to a distance. This process assumes that the flow in the line is uniform, and that no multi-phase flow occurs. It also assumes that the pipeline internal diameter is the same throughout the line.

#### Gas Composition Observations

No assumptions were made. The data is plotted exactly the same as the lab results .

#### Flushwater Composition Observations

No assumptions were made. The data is plotted exactly the same as the lab results.

#### **Oil and Grease Observations**

No assumptions were made for this analysis. For comparison purposes, all of the oil and grease measurements were normalized, based on volume flushed divided by total pipeline volume. These normalized results were also all plotted on the same graph, for comparison of all the different oil and grease flushing profiles.

#### Pipe Cutout Observations

The 5 foot pipeline sections were removed and brought to shore for examination. it is important to consider that these samples may not be representative of each pipeline as a whole.

#### Reference and Baseline Material

Some reference material was used in the analysis and comparison of Natural Seawater (NSW). These charts and articles are included in this section. This reference material has an excellent description of the ions and elements present in seawater, and how they react with each other and with other ions/elements.

Constituent	<b>Chemical symbol</b>	Mole percent	
Nitrogen	N <sub>2</sub>	78.084	
Oxygen	0 <sub>2</sub>	20.947	
Argon	Ar	0.934	
Carbon dioxide	CO <sub>2</sub>	0.0350	
Neon	Ne	0.001818	
Helium	Не	0.000524	
Methane	CH <sub>4</sub>	0.00017	
Krypton	Kr	0.000114	
Hydrogen	H <sub>2</sub>	0.000053	
Nitrous oxide	N <sub>2</sub> O	0.000031	
Xenon	Xe	0.0000087	
Ozone*	O <sub>3</sub>	trace to 0.0008	
Carbon monoxide	СО	trace to 0.000025	
Sulfur dioxide	SO <sub>2</sub>	trace to 0.00001	
Nitrogen dioxide	NO <sub>2</sub>	trace to 0.000002	
Ammonia	NH <sub>3</sub>	trace to 0.0000003	

## Gaseous composition of dry air.

\* Low concentrations in troposphere; ozone maximum in the 30- to 40-km regime of the equatorial region.

Mackenzie, F.T. and J.A. Mackenzie (1995) Our changing planet. Prentice-Hall, Upper Saddle River, NJ, p 288-307.

(After Warneck, 1988; Anderson, 1989; Wayne, 1991.)

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# Approximately what percent of air (by volume) is made up of each of two most abundant elements?

Nitrogen	N <sub>2</sub>	78.084
Oxygen	02	20.946
Argon	Ar	0.934
Neon	Ne	0.0018
*Helium	He	0.000524
Methane	CH <sub>4</sub>	0.0002
Krypton	Kr	0.000114
*Hydrogen	H <sub>2</sub>	0.00005
Nitrous oxide	N <sub>2</sub> O	0.00005
Xenon	Xe	0.0000087

The composition of air, by volume, is as follows:

The two most abundant elements in the universe, marked above with asterisks, are Hydrogen (75%) and Helium (25%).

#### Respondents: Serge, DA

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## TEXT AND DIAGRAMS BY RANDY HOLMES FARLEY

# **Understanding Seawater**

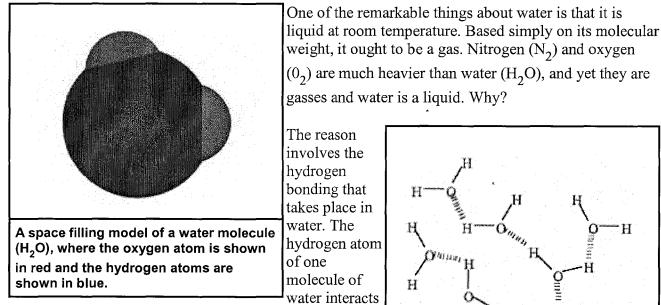
The chemistry of marine aquaria is a complex subject and one that is not easily described in a short article. Previous articles on marine chemistry in Aquarium Frontiers authored by Craig Bingman have dealt with selected topics of interest to marine aquarists. In particular, these articles have focused on the biochemistry taking place in aquaria. In this article I will endeavor to provide an understanding of seawater itself, rather than how the components are used by the tank inhabitants.

Do you have an opinion on the issues raised in this article? Join in the discussion by going to: Understanding Seawater.

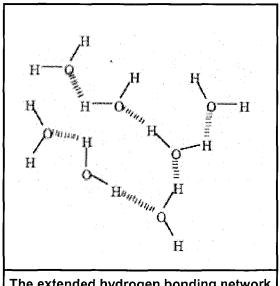
## What's In Seawater?

## Major species

Seawater has been found to contain virtually every chemical element, although some of them are found in very small concentrations. Water is, of course, the most abundant molecule, comprising about 97 percent of seawater. Water itself is far more complicated than is generally recognized and has been an active area of chemical research for more than a hundred years.



strongly with the oxygen atom of a nearby water molecule. This interaction is much weaker than the bond between atoms within a single water molecule, but it is strong enough to make the water molecules "prefer" to be surrounded by each other, rather than floating around individually, as they would in a gas. Hydrogen bonding is best viewed as a fleeting interaction between water molecules that lasts only a tiny fraction of a second before

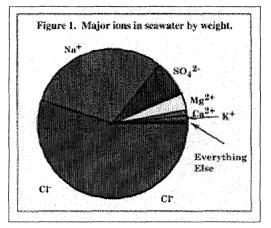


The extended hydrogen bonding network in water. Hydrogen bonds are indicated in red.

breaking. Once broken, however, they quickly reform, perhaps to a different water molecule. On balance, each water molecule is bonded to one or two other water molecules almost all of the time.

## **Major ions**

Most of the remaining constituents of seawater are inorganic ions. The major components of seawater — all ions present at greater than 1 part per million (ppm) or 1 milligram per liter (mg/L) — are shown in *Figure 1* and *Table I*. A different definition of major ions based on the numbers of ions present, rather than the weight of those ions, has a slightly different list, with lithium being added. Together, these ions account for 99.9 percent of the dissolved solutes in seawater.



It is clear from *Figure 1* that seawater contains mostly table salt (sodium and chloride). In fact, sodium and chloride comprise 86 percent of the ions present in seawater, by weight.

One important point about these concentrations: they are correct for typical seawater, which contains about 35 parts of salt by weight per thousand parts of seawater (35 ppt). This seawater has a specific

gravity of around 1.027, so it may be higher than is maintained in many marine aquaria. As the salinity of seawater is varied, these concentrations move up and down together. Consequently, if an aquarium contains water with a specific gravity of 1.023, the salinity is about 30 ppt and all of the concentrations in *Table I* are reduced by about 14 percent.

A logical question to ask is why do we not hear much discussion about chloride, sulfate or sodium levels in marine aquaria, if they are among the most abundant ions? The answer is that while they are very important, their abundance makes it difficult for them to become significantly depleted or enriched without altering the salinity. Of course, one could start out with a salt mix that did not contain the correct proportions, but assuming one starts out correctly, there isn't any normal activity in a marine aquarium that will significantly change the levels of these ions (without changing salinity).

All of these major ions are essentially unchanged in concentration at different locations in the ocean, except as salinity changes move them all up or down together. Ions that do not change concentration from place to place are referred to as "conservative type" ions, a description that also applies to some of the minor and trace elements that are discussed below.

I have also included organics on this list, though they traditionally are not considered a major specie. As will be discussed below, organics are important in seawater, but are poorly understood.

	TABLE I Major Species in Seawater					
	Species	Concentration milligrams per liter (mg/L)				
_	Cl <sup>-</sup> (chloride)	19,000				
	Na <sup>+</sup> (sodium)	10,500				
$\leq$	$SO_4^{2-}$ (sulfate)	2700				
	Mg <sup>2+</sup> (magnesium)	1280				
	Ca <sup>2+</sup> (calcium)	412				
7	Major Species Species $Cl^{-}(chloride)$ $Na^{+}(sodium)$ $SO_{4}^{2-}(sulfate)$ $Mg^{2+}$ (magnesium) $Ca^{2+}(calcium)$ $K^{+}(potassium)$ $HCO_{3}^{-}$ (bicarbonate) $Br^{-}(bromide)$ $CO_{3}^{2-}$ control (carbonate) $Br^{-}(strontium)$ $B(OH)_{3} + B$ $(OH)_{4}^{-}(borate)$ $F^{-}(fluoride)$ Organics Everything else	399				
	HCO <sub>3</sub> -	110				
	(bicarbonate)					
	Br <sup>-</sup> (bromide)	67				
	CO <sub>3</sub> <sup>2-</sup>	20				
e	(carbonate)					
	Sr <sup>2+</sup> (strontium)	7.9				
	$B(OH)_3 + B$	5 (as Boron)				
	(OH) <sub>4</sub> <sup>-</sup> (borate)					
	F⁻ (fluoride)	1.3				
	Organics	1 to 2				
ı	combined (except dissolved	Less than 1				

### **Minor ions**

There are various definitions, of which ions in seawater constitute the "minor ions." By some definitions, the list of constituents is rather long. *Table II* shows just a few of the constituents of seawater that are often labeled as minor ions. The more abundant of these are sometimes lumped with the major ions (such as lithium), while the least abundant (such as iron) are often lumped in with trace elements. Ions in this category often vary significantly with location in the ocean. That is primarily because many of them are tightly linked to biological activity. These ions can be locally depleted if biological activity is high enough. Ions that vary in this fashion are referred to as "nutrient type" ions, because they are consumed by one or more types of organism.

### **Trace elements**

There is much discussion about trace elements in marine aquaria and for good reason. Most chemicals dissolved in seawater are classified as trace elements simply because there are so many ions and molecules present at very low concentrations. In many cases, these ions are quite variable in concentration from place to place and also as a function of depth. Anyone wishing to view extensive lists of these ions is advised to check out one of the references given at the end of this article.

Many of these trace elements are metals. While people typically view dissolved heavy metals as toxic, a great many of them are essential for organisms. Their toxicity is primarily related to their concentration: a happy medium is essential, where enough of each of these metals is present for life to exist, but not so much is present as to be toxic.

	TAB Some of the Minor and	
	Species	Concentration milligrams per liter (mg/L)
	Li <sup>+</sup> (lithium)	0.17
	Rb <sup>+</sup> (rubidium)	0.12
	$H_2PO_4^{-} + HPO_4^{-2} +$	0.0 to 0.3
1	PO <sub>4</sub> <sup>3-</sup> (phosphate)	
	$IO_3^-$ (iodate)	0.03 to 0.06
	I <sup>-</sup> (iodide)	0 to 0.03
	Ba <sup>+</sup> (barium)	0.004 to 0.02
,	Al <sup>3+</sup> (aluminum)	0.00014 to 0.001
	$Fe^{2+} + Fe^{3+}$ (iron)	0.000006 to 0.00014
	$Zn^{2+}$ (zinc)	0.000003 to 0.0006

A perfect example is copper. It is present in natural seawater at about 0.25 parts per billion (ppb), which is about a thousand times less than the toxic levels often used to kill microorganisms in the treatment of sick marine fish. It is, however, absolutely necessary for many animals to have copper available to them to survive.

Some of the most important trace elements to marine aquarists are those involved in the nitrogen cycle (ammonia/nitrite/nitrate). These are discussed in detail below.

### Organics

The nature of organic molecules is certainly the most complicated aspect of seawater chemistry. Organics comprise about 2 ppm of seawater. Of this 2 ppm, the majority is in the form of dissolved organic carbon (DOC). DOC includes all fully dissolved organic compounds and any particulates that are small enough to pass through a 0.45-micron ( $\mu$ m) glass fiber filter. Strictly speaking then, it is not all fully dissolved. Any organic particles greater than 0.45  $\mu$ m are called particulate organic carbon (POC). The POC is about a factor of 10 lower in concentration than DOC and is composed of living

and dead organisms, as well as assemblies of organic molecules.

DOC is an incredibly complicated mixture of molecules that represents billions of years of biological waste products from uncounted numbers of different organisms, combined with reactions catalyzed by light, heat, inorganic catalysts (metals), biological processes, and many other factors. It includes carbohydrates (20 to 35 percent of the total), humic substances (10 to 30 percent of the total), amino acids and proteins (2 to 3 percent), hydrocarbons (less than 1 percent), carboxylic acids (1 percent) and steroids (trace).

There is also a great deal of uncharacterized organic material. In fact, the study of seawater organics is an active area of research. Additionally, the summation of all dissolved organics in the ocean is a pool of carbon larger than carbon dioxide in the atmosphere, so it cannot be ignored by those looking at the planetary carbon cycle. In addition to carbon, these organics contain significant amounts of oxygen, nitrogen, phosphorus, and sulfur.

It is probably also safe to say that most, if not all, closed marine systems have higher organic levels than the ocean, although hard numbers are difficult to come by. The desire to reduce these organic levels is one of the reasons for the popularity of skimmers with marine aquaria.

## What Forms Do Ions Take In Seawater?

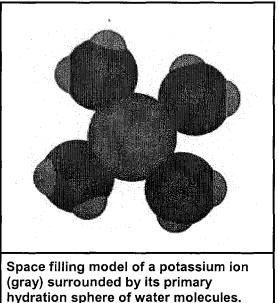
In the previous sections I have described what ions are present in seawater, but I have not presented the forms they typically take. Contrary to popular belief, many of these ions are attached to each other in solution and do not act as completely individual species. This tendency to form ion pairs in solution is much more prevalent for some ions (e.g.,  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $CO_3^{2-}$ , F<sup>-</sup>, OH<sup>-</sup>) than it is for some

others (e.g.,  $Na^+$ ,  $K^+$ ,  $Cl^-$ ,  $Br^-$ ). In general, the tendency to form ion pairs is higher for ions with a higher net charge. In the next few sections, I will present an overview of some of these interactions and why they are important.

## Simple ions

The simplest positively charged ions in solution are sodium (Na<sup>+</sup>) and potassium (K<sup>+</sup>). They are primarily free ions, with a shell of three to four tightly bound water molecules attached to them. This is known as the "primary hydration sphere." These water molecules are fairly tightly bound, but are rapidly exchanged with other water molecules from the bulk solution (at a rate of about a billion exchanges per second for each ion!). Beyond this first shell are another 10 to 20 water molecules that are less tightly bound, but that are still strongly influenced by the metal ion. These types of hydrating water molecules are present for all ions in solution and won't be mentioned further for each ion in turn.

A small proportion of both sodium and potassium (about 5 percent) exists as ion pairs with sulfate, forming  $NaSO_4^-$  and  $KSO_4^-$ . This type of ion pair is best viewed



as a temporary association between the two ions and may only last for a very small fraction of a second before the ions move apart. Nevertheless, this type of association can have very important implications for the behavior of these ions, as will be shown below. Ions forming such pairs actually "touch" each other. That is, most or all of the hydrating water molecules that are in between them have been temporarily removed. This removal of the intervening water molecules is the primary distinction between ion pairs and ions that are simply near each other.

The simplest negatively charged ions, chloride (Cl<sup>-</sup>) and bromide (Br<sup>-</sup>), form few ion pairs in solution. They are primarily present in the form of hydrated free ions, with two and one tightly bound water molecules, respectively.

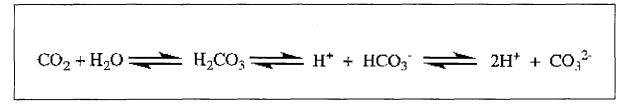
## Carbonate

One of the more complex interactions, and one that is very important for marine reefkeepers, involves carbonate  $(CO_3^{2^-})$ . Carbonate is primarily ion paired in solution, with only about 15 percent of it actually present as free  $CO_3^{2^-}$  at any given point in time. This fact is very important to the maintenance of calcium and alkalinity levels in aquaria, because it is the free carbonate concentration that "wants" to precipitate with calcium as calcium carbonate (CaCO<sub>3</sub>). If the free carbonate levels rise too much, the calcium levels will drop due to CaCO<sub>3</sub> precipitation.

So, what is carbonate ion paired with? Primarily magnesium, forming soluble  $MgCO_3$ . This is the reason why magnesium levels are so important in marine aquaria for maintenance of simultaneously high levels of alkalinity and calcium. If magnesium is too low, more carbonate will be in the free form and will "want" to precipitate as calcium carbonate.

Carbonate is also ion paired to sodium and calcium, forming soluble  $NaCO_3^-$  and  $CaCO_3^-$ , respectively. The soluble calcium ion pair sounds odd, but it is essentially one individual molecule of  $CaCO_3^-$  that is soluble in water: it is not precipitated out of the solution. The fact that carbonate is also ion paired by sodium is one of the reasons that salinity has an impact on the amount of calcium and alkalinity that can be maintained in solution: lower salinity means lower sodium, which means more free carbonate and a greater likelihood of precipitation of  $CaCO_3^-$ .

Ion pairing has another large effect on carbonate that is more subtle. In water, carbon dioxide hydrates to form  $H_2CO_3$ , which can then break up (ionize) into protons (H<sup>+</sup>), bicarbonate (HCO<sub>3</sub><sup>-</sup>) and carbonate CO<sub>3</sub><sup>-2-</sup>).

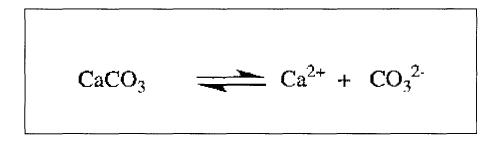


When  $CO_2$  is added to water, the system will come to equilibrium with specific concentrations of each of the species shown above. By LeChatelier's principle, if one takes away something from one

side of the equilibrium, the equilibrium will shift in that direction. For example, if carbonate is removed from the system, then each of the reactions shown will proceed to the right, effectively replacing some of the carbonate that was removed.

Importantly, that is exactly the effect that takes place in seawater when carbonate is "removed" by forming ion pairs. It is only the "free" concentration of these species that determines the position of the chemical equilibrium, so carbonate in the form of an ion pair does not "count," and the equilibrium shifts strongly to the right. If one then counts carbonate in all forms (free and ion paired) it is found to be far higher in seawater than in freshwater at the same pH and ion pairing is the primary reason.

The exact same effect can be seen in the solubility of  $CaCO_3$ .



In this case, if  $CaCO_3$  is added to water, it breaks apart into  $Ca^{2+}$  and  $CO_2^{2-}$ . Eventually, an equilibrium is reached where no more  $CaCO_3$  will dissolve. However, if some of the carbonate is removed by ion pairing (and some of the  $Ca^{2+}$  as well), then additional  $CaCO_3$  can dissolve to replace those that were "lost." This is the primary reason that  $CaCO_3$  is approximately 15 times more soluble in seawater than in freshwater.

### Calcium, magnesium and strontium

Calcium, magnesium and strontium are primarily present in the free form, hydrated by six to eight tightly bound water molecules. A small percentage (about 15 percent) is present as an ion pair with sulfate. Much smaller percentages are present as ion pairs with carbonate and bicarbonate. Importantly, while these complexes involve only a small percentage of the total calcium and magnesium, they involve a large portion of the total carbonate (which is possible because there is so much calcium and magnesium compared to carbonate).

### Sulfate

As mentioned above, sulfate forms ionic interactions with most positively charged species in seawater. In fact, more than half of it is in the form of an ion pair, with  $NaSO_4^-$  and  $MgSO_4$  dominating.

## Phosphate

Phosphate in marine aquaria is of tremendous importance because it is often a limiting nutrient for algae growth. In seawater, the amount of phosphate present is typically quite low (usually less than 0.1 ppm) and often varies significantly from location to location. In many marine aquaria, however,

the phosphate concentration can be significantly higher (up to several ppm).

The ability to export phosphate from marine aquaria has been the topic of lengthy discussion and is the object of numerous commercial products. The nature of the inorganic phosphate present in marine aquaria, however, is certainly more complicated than traditionally credited.

Inorganic phosphate can exist in a number of forms, in a manner analogous to carbonate.

$$H_3PO_4 = H^+ + H_2PO_4 = 2H^+ + HPO_4^{2-} = 3H^+ + PO_4^{3-}$$

Ignoring ion pairing and complex formation for the moment, phosphate is primarily found in the  $HPO_4^{2-}$  and  $PO_4^{3-}$  forms in seawater. This is quite different than in freshwater at the same pH, where the  $H_2PO_4^{-2}$  and  $HPO_4^{-2}$  forms predominate. *Table III* shows the forms of phosphate present in seawater at a pH of 8.1.

To a large extent, the high proportion of phosphate present in the **TABLE III Speciation of Phosphate in**  $PO_4^{3-}$  form in seawater is due to ion pairing, just as in the case of Seawater carbonate. These various phosphate species pair extensively with Form Percentage of total in magnesium and calcium in seawater.  $PO_4^{3-}$  is nearly completely ion seawater (at pH 8.0) paired (96 percent), while only 44 percent of  $HPO_4^{2-}$  is paired. This is H<sub>2</sub>PO<sub>4</sub> trace what causes the shift in the equilibrium to more of the  $PO_4^{3-}$  form in H<sub>2</sub>PO<sub>4</sub>-0.5 percent seawater compared to freshwater (just as it does for carbonate).  $HPO_4^{2-}$ 79.2 percent

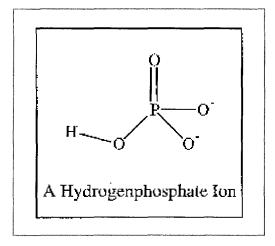
Phosphorus is

organics. While natural seawater

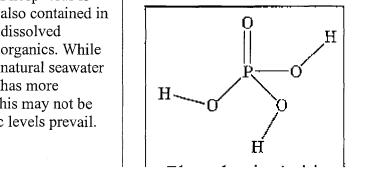
dissolved

has more

Additionally, phosphate will interact with certain



inorganic phosphate than organic forms, this may not be true in aquaria where much higher organic levels prevail.



PO\_4<sup>3-</sup>

ions in a manner that is much stronger than simple ion pairs. Phosphate can, for example, complex with a number of positively charged species, including both metals (e.g., iron) and organics. These interactions further serve to

reduce the concentration of free phosphate.

20.4 percent

### Metals

The metals, in particular, are strongly ion paired in solution. Copper primarily forms soluble  $CuCO_3$ , iron

forms soluble  $Fe(OH)_3$  and silicon (not strictly a metal) forms  $(Si(OH)_4)$ . Some of the other metals

that are biologically important (e.g., zinc, molybdenum, manganese, cobalt) form a wide variety of ion pairs with different ions in solution. In some cases, the number of different species that form is extensive. *Table IV* shows the speciation of copper in seawater at a pH of 8.1.

In recent years, however, it has become more and more apparent that certain metals are largely complexed to organic materials, even in natural saltwater where the level of organics is low. In a marine aquarium, the level of organics can be higher than in the ocean, so such complexes are even more likely to form.

In addition to complexation of metals to the widespread organics present in the oceans (e.g., humic acids), there is also the possibility of complexation to specific organics that were made exclusively for that purpose. For many microorganisms, metals such as iron are limiting nutrients for growth and these creatures have designed systems to bring iron to them.

cuCO<sub>3</sub> Cu(CO<sub>3</sub>) $_2^{2-}$ the possibility exclusively for as iron are designed Cu(OH)<sup>+</sup> Cu(OH)<sub>2</sub> CuSO<sub>4</sub> CuHCO<sub>3</sub><sup>+</sup> pounds called

Bacteria and fungi, for example, release organic compounds called

siderophores into the environment. They are large organic molecules with a very high affinity for iron. The released siderophores eventually encounter an iron atom and bind very strongly to it. The organisms themselves have enzymes in their outer membranes that interact strongly with siderophores that contain iron, and transport them into the cell. Consequently, the siderophores can be viewed as collection devices for iron.

Of course, many of the siderophores released into the ocean are not quickly reabsorbed by the microorganisms and remain in solution. In a closed marine aquarium with a large population of microorganisms, one would expect that such molecules would be present in solution. Consequently, many metals in solution may be bound by such molecules.

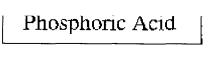
Additionally, many aquarists intentionally add complexing agents in the various supplements they add to their aquaria. These include EDTA and citrate, which are two common forms for adding iron. These will equilibrate with other metals already in the tank and the tank will then contain a variety of metals complexed to these organics.

### Nitrogen compounds

The primary nitrogen compound in seawater is nitrogen gas  $(N_2)$ . It is present at about 11 ppm at 25

degrees Celsius (77 degrees Fahrenheit), although its solubility is a strong function of temperature, with nearly twice as much dissolving in near freezing seawater. Nitrogen gas is present at a higher concentration than any other dissolved gas, with oxygen  $(0_2)$  at 7 ppm, argon (Ar) at 0.4 ppm and all others at sub-ppb levels (not including carbon dioxide, which is primarily ionized in seawater).

There are certain organic and inorganic forms of nitrogen at concentrations lower than nitrogen gas. The organic forms are poorly defined, but include such molecules as proteins.



**TABLE IV** 

Speciation of Copper in

Seawater

Copper form Percentage of total

73.8

14.2

4.9

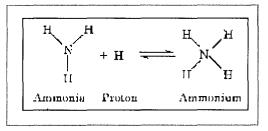
3.9

2.2

1.0

0.1

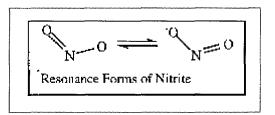
The inorganic forms are much more familiar to aquarists as components of the nitrogen cycle. The concentrations of these components in seawater are highly variable. In natural seawater, ammonia  $(NH_3)$  ranges in concentration from 0.02 to 8 ppm (as ammonia), nitrite  $(NO_2^-)$  ranges from 0.005 to 0.2 ppm (as nitrite) and nitrate  $(NO_3^-)$  ranges from 0.06 to 30 ppm (as nitrate). These values vary by location, depth and time of year. Other inorganic forms present at much lower concentration include hydroxylamine  $(NH_2OH)$ , nitrous oxide  $(N_2O)$ , and hyponitrite  $(N_2O_2^{-2})$ .



Ammonia exists in two forms in seawater. The primary form is ammonium  $(NH_4^+)$ , which accounts for about 95 percent of the total in seawater at a pH of 8.1. The secondary form is free ammonia  $(NH_3)$ , which accounts for the remaining 5 percent. These proportions vary strongly with pH and the free ammonia form rises as pH rises, to about 50 percent of the total at a pH of 9.5.

The toxicity of ammonia towards fish has been found to depend upon pH, with some researchers observing lower toxicity at lower pH. It has been suggested that this relationship between toxicity and pH is due to the proportion of ammonia in each form at a given pH. While these ideas seem to have been accepted by many in the aquarium hobby, the exact cause of this relationship is unclear and is beyond the scope of this article. This topic is discussed in more detail in *Captive Seawater Fishes* (Spotte 1992).

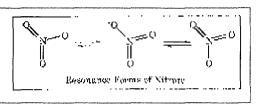
Nitrite and nitrate are both interesting molecules in that they exist in a number of resonance forms. If one draws a simple structure for these molecules it appears that the oxygen atoms are not all exactly the same, with one carrying a negative charge, while the others do not. Experimentally, however, this has not been found to be the case: all oxygen atoms are exactly equivalent.



How can this be? Resonance forms are a simple way of thinking about this, with the various forms interconverting extremely rapidly. The only thing required to convert one form to another is to move electrons around within the ion, so it can happen essentially instantly. In reality, the electrons are spread around these ions in such a way that each oxygen on average carries a partial negative charge (-3/4) in the case of nitrite; -1/3 in the case of nitrate).

### Iodine

Iodine seems to get an amazingly disproportionate amount of discussion with respect to marine aquaria and much of it is incorrect. The reasons for this are many, but are primarily related to its chemical and biochemical complexity. In fact,

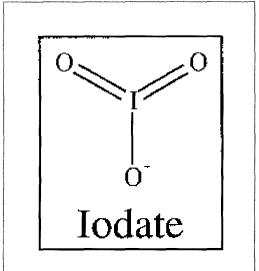


its chemical complexity is far greater than many aquarists are aware.

Iodine takes two primary forms in seawater: iodide ( $I^{-}$ ) and iodate ( $IO_{3}^{-}$ ). The often quoted value for

the total concentration of iodine in seawater (0.06 ppm) is reasonably accurate, although the value varies significantly. This value, however, is a combination of both iodide and iodate. It is not correct to state that seawater contains 0.06 ppm of iodide. The value for iodide is more typically around 0.01 ppm or less, although it is sometimes as high as 0.03 ppm and sometimes as low as 0.002 ppm. The remainder is iodate.

Additionally, the interconversion between iodide and iodate in seawater is very slow. This reaction is believed to be mediated in a number of ways, including catalysis by light and microorganisms. It is probably safe to say, however, that the two are not in equilibrium in marine aquaria. One effect of this lack of equilibrium is that dosing one type does not necessarily give you any of the other type.



It is not well known which forms are used by which organisms, so I will not comment on the necessity of maintaining specific levels of iodide or iodate. There is good evidence, however, that iodide is rapidly depleted in marine aquaria, although it is not well established where it goes. Conversion of iodide to iodate has been observed in aquaria, but this may not represent a significant sink. Iodate itself is much slower to become depleted from marine aquaria and can build up to toxic levels if it is being actively dosed.

An additional complication is that some aquarists dose a third form of iodine:  $I_2$ . Lugol's solution, for example, is a combination of iodide and iodine. When iodine (as  $I_2$ ) is

added to seawater, it quickly reacts to form other iodine

species that probably end up as both iodide and iodate in marine tanks.

## Conclusion

There are, of course, many other details of seawater chemistry that may be of interest to marine aquarists. This article is only a first pass at understanding the chemistry behind what is happening in our tanks.

For those wanting a more in depth exposure to marine chemistry, I recommend two books: *Captive Seawater Fishes. Science and Technology* by Stephen Spotte (Wiley-Interscience, New York. Pp. 942.) and *Chemical Oceanography, Second Edition* by Frank J. Millero (CRC Press, Boca Raton, FL. Pp. 469.).

The Spotte book is excellent, with sections directed specifically toward aquarium chemistry. It covers chemistry from the standpoint of aquarium keeping, rather than understanding of the natural ocean. It is also practically oriented, rather than directed toward a deep chemical understanding of phenomena.

The Millero book will only be of interest to those who are undaunted by chemical reactions and jargon. It is, however, the best marine chemistry book I have encountered. It gives a tremendous amount of detail about natural marine systems, but has no discussion about aquaria. Most of the chemical data in this paper was pulled from this book.

Previous "Biochemistry of Reef Aquariums" columns in Aquarium Frontiers magazine have also

http://www.animalnetwork.com/fish2/aqfm/1998/july/features/1/default.asp

dealt with selected topics of interest to marine aquarists, especially the column on "Ion Pairing, Buffer Perturbation and Phosphate Export in Marine Aquariums" (Bingman, C. 1996. *Aquarium Frontiers* 3[1]:10-17).

HOME Table Of Contents FEATURE

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\*

# Detailed composition of seawater

at 3.5% salinity							
Element	At.weight	ppm	Element	At.weight	ppm		
Hydrogen H2O	1.00797	110,000	Molybdenum Mo	0.09594	0.01		
Oxygen H2O	15.9994	883,000	Ruthenium Ru	101.07	0.0000		
Sodium NaCl	22.9898	10,800	Rhodium Rh	102.905			
Chlorine NaCl	35.453	19,400	Palladium Pd	106.4	.		
Magnesium Mg	24.312	1,290	Argentum (silver) Ag	107.870	0.0002		
Sulfur S	32.064	904	Cadmium Cd	112.4	0.0001		
Potassium K	39.102	392	Indium In	114.82			
Calcium Ca	10.08	411	Stannum (tin) Sn	118.69	3000.0		
Bromine Br	79.909	67.3	Antimony Sb	121.75	0.0003		
Helium He	4.0026	0.0000072	Tellurium Te	127.6			
Lithium Li	6.939	0.170	Iodine I	166.904	0.064		
Beryllium Be	9.0133	0.000006	Xenon Xe	131.30	0.0000		
Boron B	10.811	4.450	Cesium Cs	132.905	0.0003		
Carbon C	12.011	28.0	Barium Ba	137.34	0.021		
Nitrogen ion	14.007	15.5	Lanthanum La	138.91	0.0000		
Fluorine F	18.998	13	Cerium Ce	140.12	0.0000		
Neon Ne	20.183	0.00012	Praesodymium Pr	140.907	0.0000		
Aluminium Al	26.982	0.001	Neodymium Nd	144.24	0.0000		
Silicon Si	28.086	2.9	Samarium Sm	150.35	0.0000		
Phosphorus P	30.974	0.088	Europium Eu	151.96	0.0000		
Argon Ar	39.948	0.450	Gadolinium Gd	157.25	0.0000		
Scandium Sc	44.956	<0.00004	Terbium Tb	158.924	0.0000		
Titanium Ti	47.90	0.001	Dysprosium Dy	162.50	0.0000		
Vanadium V	50.942	0.0019	Holmium Ho	164.930	0.0000		
Chromium Cr	51.996	0.0002	Erbium Er	167.26	0.0000		
Manganese Mn	54.938	0.0004	Thulium Tm	168.934	0.0000		
Ferrum (Iron) Fe	55.847	0.0034	Ytterbium Yb	173.04	0.0000		
Cobalt Co	58.933	0.00039	Lutetium Lu	174.97	0.0000		
Nickel Ni	58.71	0.0066	Hafnium Hf	178.49	<0.000		
Copper Cu	63.54	0.0009	Tantalum Ta	180.948	<0.000		
Zinc Zn	65.37	0.005	Tungsten W	183.85	<0.000		
Gallium Ga	69.72	0.00003	Rhenium Re	186.2	0.0000		
Germanium Ge	72.59	0.00006	Osmium Os	190.2			
Arsenic As	74.922	0.0026	Iridium Ir	192.2			
Selenium Se	78.96	0.0009	Platinum Pt	195.09			
Krypton Kr	83.80	0.00021	Aurum (gold) Au	196.967	0.0000		
Rubidium Rb	85.47	0.120	Mercury Hg	200.59	0.0001		
Strontium Sr	87.62	8.1	Thallium Tl	204.37			
Yttrium Y	88.905	0.000013	Lead Pb	207.19	0.0000		
Zirconium Zr	91.22	0.000026	Bismuth Bi	208.980	0.0000		
Niobium Nb	92.906	0.000015	Thorium Th	232.04	0.0000		
			Uranium U	238.03	0.0033		
			Plutonimu Pu	(244)	·		
	NI-fal		n = ma/litro = 0.001a/ka				

Note! ppm= parts per million = mg/litre = 0.001g/kg.



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## Dissolved Oxygen Table (mg/l)

				Salinity	(ppt)	·····		
Temperature ⁰C	0	5	10	15	20	25	30	35
0	14.6	14.11	13.64	13.18	12.74	12.31	11.90	11.50
2	13.81	13.36	12.91	12.49	12.07	11.67	11.29	10.91
4	13.09	12.67	12.25	11.85	11.47	11.09	10.73	10.38
6	12.44	12.04	11.65	11.27	10.91	10.56	10.22	9.89
8	11.83	11.46	11.09	10.74	10.40	10.07	9.75	9.44
10	11.28	10.92	10.58	10.25	9.93	9.62	9.32	9.03
12	10.77	10.43	10.11	9.80	9.50	9.21	8.92	8.65
14	10.29	9.98	9.68	9.38	9.10	8.82	8.55	8.29
16	9.86	9.56	9.28	9.00	8.73	8.47	8.21	7.97
18	9.45	9.17	8.90	8.64	8.38	8.14	7.90	7.66
20	9.08	8.81	8.56	8.31	8.06	7.83	7.60	7.38
22	8.73	8.48	8.23	8.00	7.77	7.54	7.33	7.12
24	8.4	8.16	7.93	7.71	7.49	7.28	7.07	6.87
26	8.09	7.87	7.65	7.44	7.23	7.03	6.83	6.64
28	7.81	7.59	7.38	7.18	6.98	6.79	6.61	6.42
30	7.54	7.33	7.14	6.94	6.75	<u>6.</u> 57	6.39	6.22

7204



## 5. Results and Observations - 2820

### a. Sample/Locations Observations

Gas samples were taken at the top of the riser before the blind flange was removed. Samples were taken when the odor of natural gas was present. All bolts and flange seals were intact and did not indicate any leakage. H2S length of stain samples were taken at this time as well. Sample 2820-B obtained just ahead of the flush water - had some water in it. The testing lab indicated that this water would not affect the gas samples integrity.

Water samples were taken at the same location. Water samples seemed uniform, and representative of the flush fluid stream.

A five foot (5') sample of the pipeline was removed, which included the tubeturn to pipeline weld.

### b. Gas Composition Observations

The results of the gas analysis are plotted and summarized in the results section. Atmospheric air composition is also plotted for reference purposes.

Four gas samples were taken. Two of the samples (2820-1 and 2820-A) were taken at the same time and location. Sample 1 was taken using a steel vacuum tube and Sample A was taken using a Tedlar bag. According to the lab, the Tedlar bags are the preferred sampling method, as when they are full, one can be sure they contain a sample, whereas with the tubes, there is no indication that a sample was taken. It is important to note that the composition of the two samples is different. The sample taken with the vacuum tube contains more methane and less nitrogen than the sample taken with the Tedlar bag. Sample 2820-B, which is estimated to originate 2670' from the well protector platform is lower in methane and higher in nitrogen than the other two samples. It has almost the same amount of oxygen and nitrogen as the atmosphere. This indicates that it is probably a mixture of natural gas, and atmospheric air. This would follow from the fact that end of the pipeline was opened in order to connect the flushing pump. At that time, it would have been possible to introduce air into the line.

## c. Flushwater Composition Observations

The flushwater composition for segment 2820 is plotted in the results section. Natural Seawater composition is also plotted for comparison purposes. The ions/elements plotted are: Alkalinity (CO3), Barium, Calcium, Iron, Magnesium, and Potassium.



Because of their high values (in PPM), Chlorides and Sulfates are plotted on a separate chart.

For the flushwater, the mineral pattern relative to NSW is summarized below:

Alkalinity (bicarb) – Higher Barium – Higher/Same Calcium – Lower/Equal (First sample much lower) Iron – Higher Magnesium – Lower/Equal (First sample much lower) Potassium – Lower/Equal (First sample much lower) Chloride – Lower Sulfate – Lower/Higher (First sample much lower)

The first sample, containing the most hydrocarbons was MUCH lower than NSW in almost all elements/ions tested for.

The iron content is plotted as a separate graph in order to focus on these values. The first sample had a very high iron concentration of 91.3 ppm (ppm also equals milligrams/liter). This concentration is over 26,000 times greater than NSW. Observations from the field could explain this very high concentration. The sample was taken at the very front of the flushwater "slug." This slug picked up metal debris, as can be evidenced in the photographs. This debris included metal particles which were picked up from the pipe wall. The sampling procedure "dissolved" these metal particles and recorded them as a concentration value. The following two samples were lower in concentration, but still much higher than NSW values.





Photo #12 - Mineral Pattern Analysis Samples - 2820

The ions/elements to focus on from this analysis are those found in steel corrosion products: FeO2, FeS. The samples showed higher than NSW concentrations of both Fe and S, indicating that corrosion has taken place, however, it is difficult to derive specific corrosion features from this data.

# d. Oil and Grease Observations

Samples taken at the end of the flushing operation had no detectable oil and grease concentration. The detection limit is 2.5 PPM. The last sample was taken when approximately 1.75x the pipeline volume had been flushed. The graph shows a very rapid drop in oil and grease concentration, with the non-detectable limit appearing to be reached at 1.5x flush volume.

# e. Pipe Cutout Observations

A five foot horizontal section of pipe was retrieved from near the base of the platform and includes pipe on both sides of the weld connecting the pipeline to the riser/tubeturn.

This sample showed only light surface rust and had some debris in the 5-7 o'clock position of the line, indicating that there may have been some standing fluid in the



pipeline for some time. This area did not show any significant metal loss, but has a buildup or caking of silt/sand.

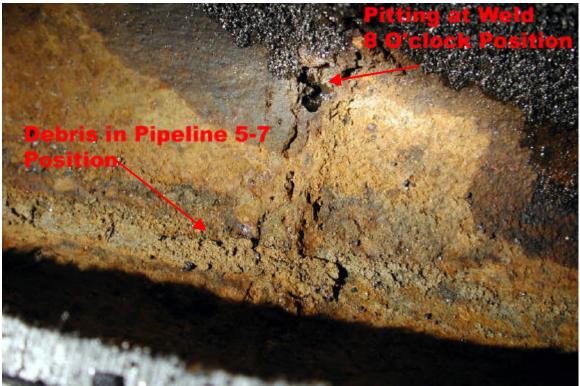


Photo #13

An important feature to note is that this sample did have a deep pit in the weld. The pit is clearly shown in the sample photos. The depth of the pit appears to be approximately 0.5\*t. This defect is in the 10 o'clock position in the pipe, so it was not in the "wet" section of the pipe. Based on the shape of the defect, it appears to be a corrosion feature, and not caused by erosion (due to sand or other abrasives in the gas). This looks like an Microbial Induced Corrosion pit.





Photos #14 and #15 - Sample Photos from Line 2820



Photo #15



# SHELL OFFSHORE INC. HI-135-1

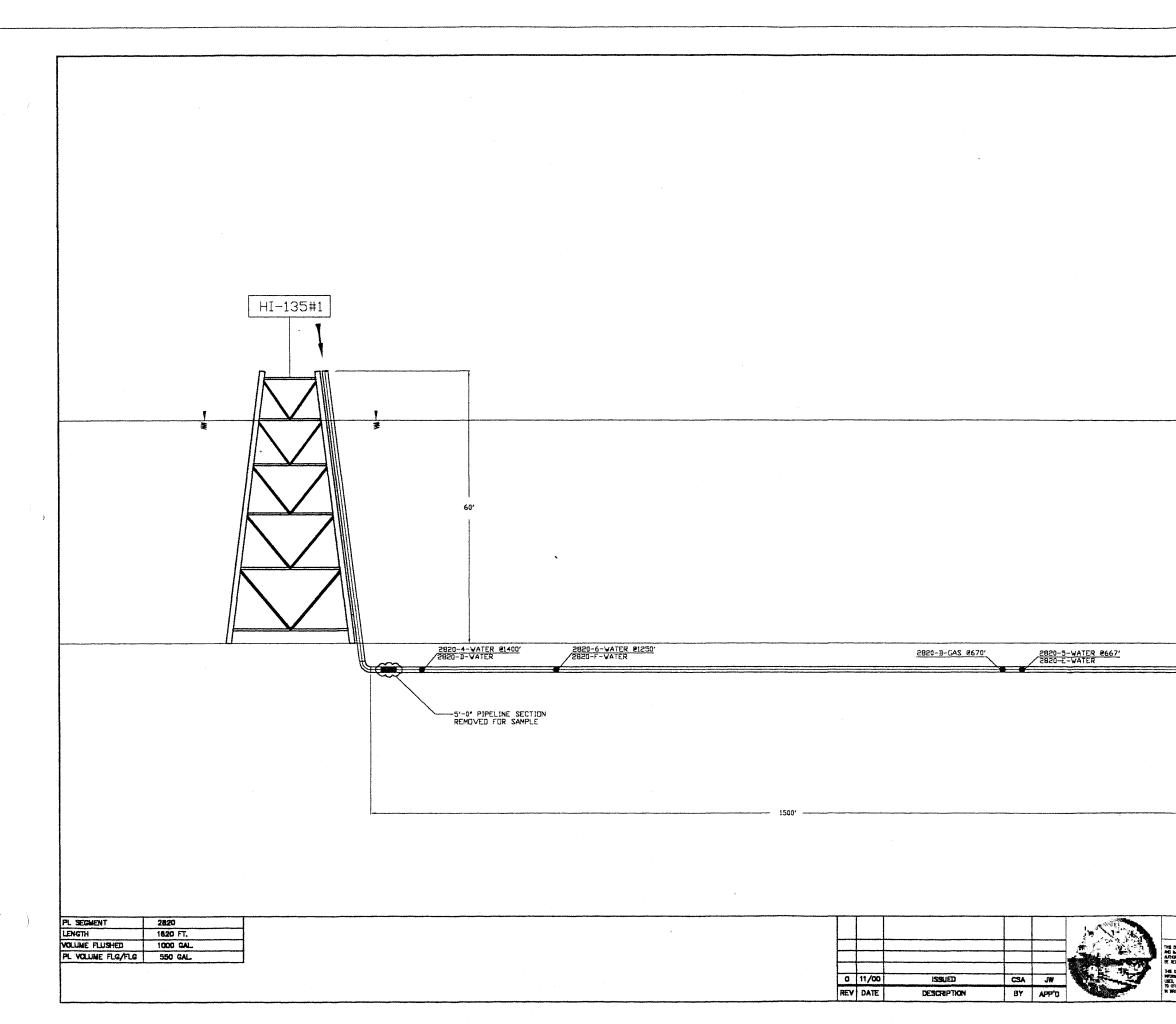
			Р	LATFOR	M			
MMS Gen	eral	0	DS Gener	al	MMS Lo	ocation	MMS Fa	cility
Water	49 feet	Function	WP		Lease	741 feet	Helideck	Yes
Major	No	Piles	NA		Complex	10025	Quarters	None
Decks	1	Slots	3		Longitude	-94.119	Generator	No
Slots	3	ODS ID	738		Latitude	29.259	Cranes	NA
Wells	2	Previous 1	NA		Х	3,555,877'	Gas	Yes
Flare	No	Previous 2	NA		Y	550,662'	Oil	No
Installed	01 1964	Previous 3	NA		To Shore	25 miles	Comp	No
Revised	12 1998	Previous 4	NA		N-S feet	S 4182'	8 hour	No
Removed	NA	Notes	NA		E-W feet	W 881'	24 hour	No
			PIF	PELINES	MMS			
Segment	2820							
Origin	HI-135-1							
Terminus	HI-136-A							
O.D.	3"							
Length	1,500'							
Product	BLKG							
Status	ACT							
Installed	NA							
Abandon	NA							
Revised	Aug-94							
Operator	SHELL O	FFSHORE INC	-					
			۷	VELLS N	MMS			
API Well ID	Well	Spud	Revised	Status	MD	Bot Lease	Sur Long	Sur Lat
427080004600	1	03 1964	05 1964	COM	10,777'	741	-94.119	29.259
427080004700	2	10 1964	05 1986	ST	8,678'	741	-94.112	29.260
427080004701	2	05 1986	07 1986	COM	9,450'	741	-94.115	29.260
427080004800	3	11 1964	08 1985	PA	9,663'	741	-94.104	29.265
427080004900	4	11 1964	06 1990	PA	11,483'	741	-94.116	29.269
427080005000	5	04 1965	08 1985	PA	9,006'	741	-94.089	29.262
427080006100	6	06 1965	06 1965	PA	9,085'	741	-94.089	29.271
	8	07 1967	08 1967	ST	9,452'	741	-94.103	29.260
427080007300	9	08 1967	08 1967	COM	9,020'	741	-94.106	29.260
	0		1		40.004	741	04.075	29.262
427080007301	11	04 1986	01 1996	ST	10,824'	741	-94.075	29.202
427080007300 427080007301 427084023300 427084023301		04 1986 01 1996	01 1996 04 1996	ST COM	10,824' 11,033'	741	-94.075 -94.075	29.202



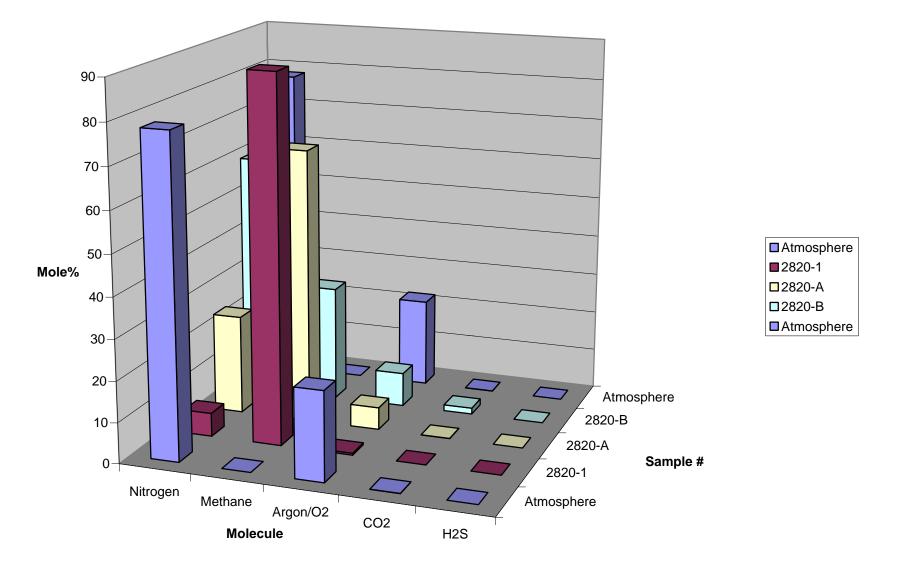
### PIPELINE FLUSHING AND SAMPLING RECORD

I. Pipeline Information	
MMS Segment No.	2820
Date:	10/24/2000
Pipeline Origination	
Area	High Island
Block	135
Platform	#1
Lease	OCS-G-0741
Pipeline Destination	
Ārea	High Island
Block	136
Platform	A
Lease	OCS-G-0742
Pipeline Size (in)	3
Pipelines Length (ft)	1,500
Pipeline Volume (bbls)	13
· · · · ·	
II. Flushing Information	
Flushing Information	
Volume Flushed	1000g
Flow Rate (GPM)	100
Pigged Used	No
Type of Pig	No
Size of Pig	No
Clean Returns	Yes
Inhibitor	
Chemical Inhibitor Used	
Type of Chemical	
Quantity of Chemical	
Origination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Destination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Comments:	
Company Representative	
Signature	

III. Sampling Data - Tracking									
Sample Location									
Platform:	HI 136A								
Pipeline Sampling Site:		Top of Riser Bleed Valve							
Flushing Start Time:	15:00	15:00							
Flushing Start Time: Gas Samples	Sample ID	Sample Date	Vol. Flushed (g)	H2S (PPM)					
Vacuum Tubes									
	2820-1	10/24/2000	0						
Plastic Bags									
	2820-A	10/24/2000	0						
	2820-B	10/24/2000	250						
Water Samples	Sample ID	Sample Date	Vol. Flushed (g)	Notes					
Vineral Pattern Analysis									
	2820-4	10/24/2000							
	2820-5	10/24/2000							
	2820-6	10/24/2000	1,000						
Oil and Grease Analysis	0000 D								
	2820-D	10/24/2000							
	2820-E	10/24/2000	800						
	2820-F	10/24/2000	1,000						
Comments:									
	2820-B had s	ome water in the gas	s sample.						
Company Representative	2020 2 1144 0	onio nator in tro gao	, earlipier						



HI-	-136A				
THAR STATES		ROJECT# WM			
VIE IS THE PROFERTY OF THINKE CONSULTING SERVICES, NC., NO LAY NOT IS: USED FOR MAY REPOSE UNLESS UNDERED: IN UTIONS, MOI (VOID COMMOL ING). R. RELINIED TO MINUTE COMMULTING BETWEEK, NG., RES DOCUMENT COMMUNE CONFIDENTIAL PROFILETY	PIPEL	INE SAMPLE L		36.4	
HIS DOCULED!! CONTINUE CONTINUE AND THE CONTINUE AND SHILL NOT BE CONTINUE AND SHILL NOT BE CONTINUE AND AND CONTINUE AND	CSA	DATE 11/00	GH ISLAND 1. DVG. NOL WM0070E	REV.	
	······		4	لمتتحم	



# Figure 2 - Gas Composition (Major Constituents) - 2820

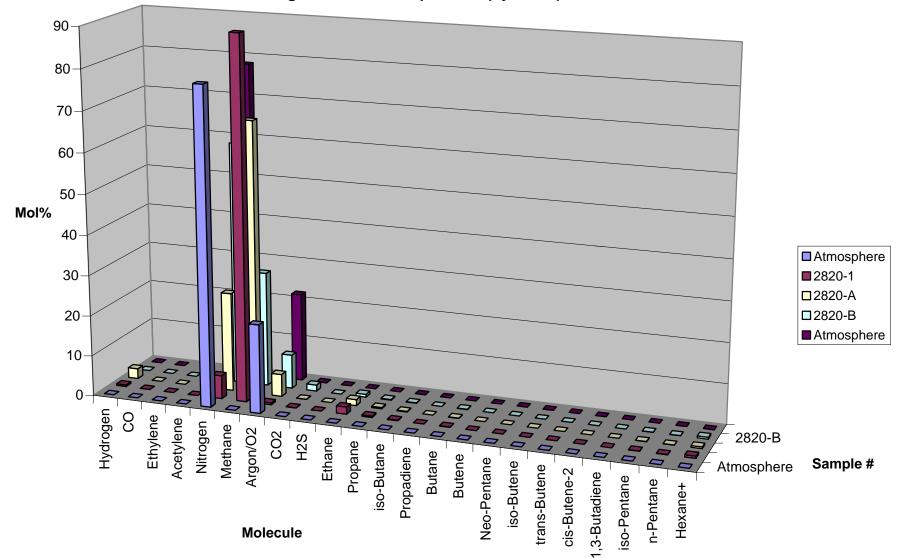
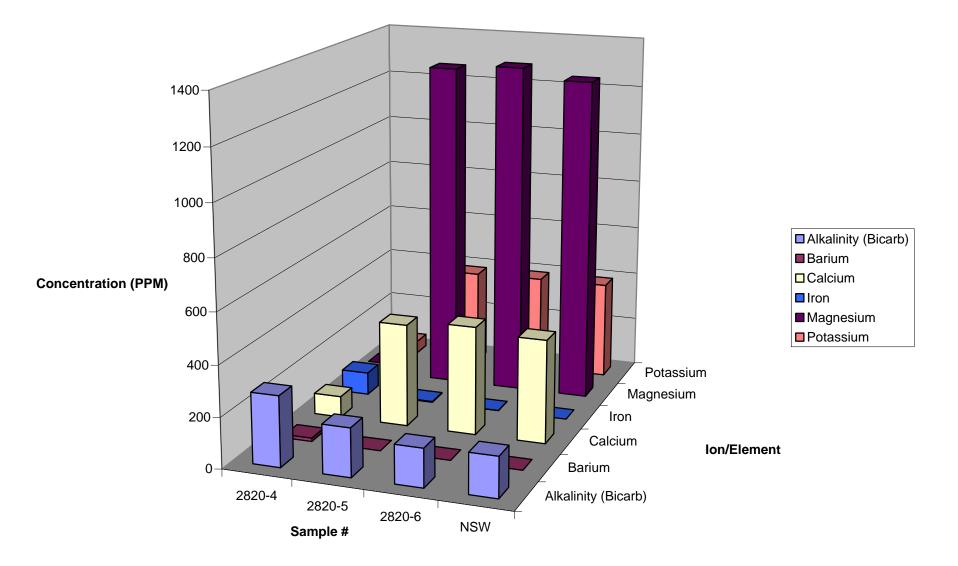
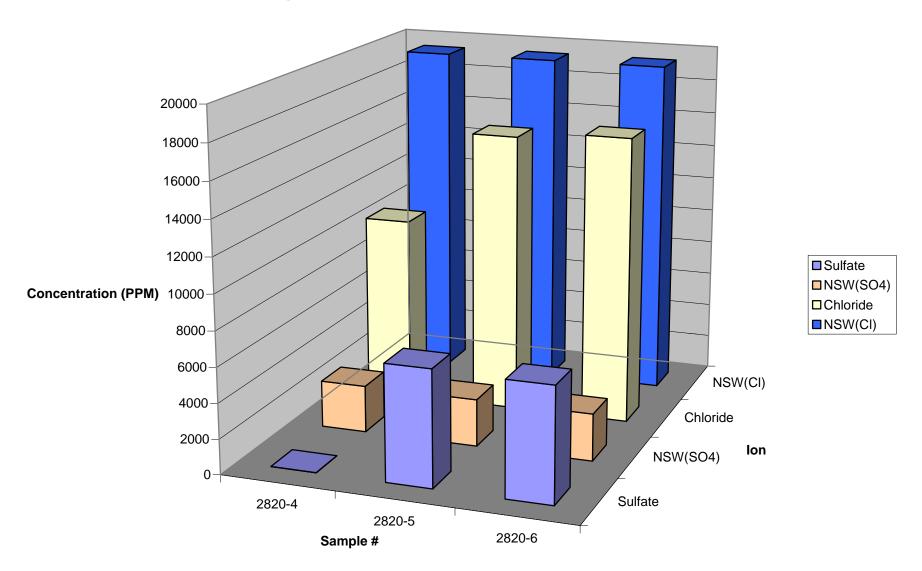


Figure 3 - Gas Composition (by Mol%) - 2820

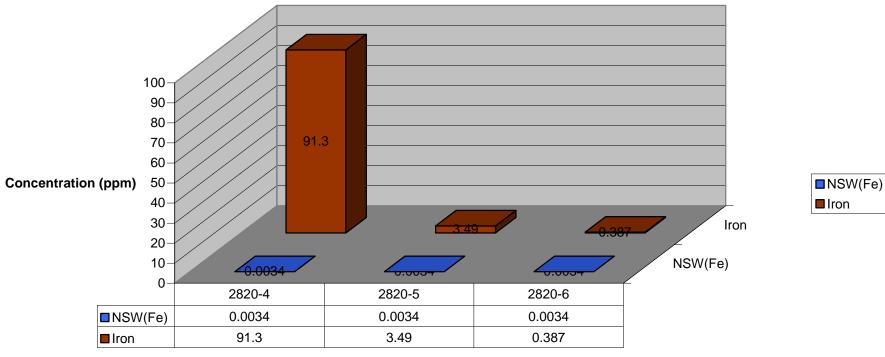
# Figure 4 - Flushwater Composition- 2820



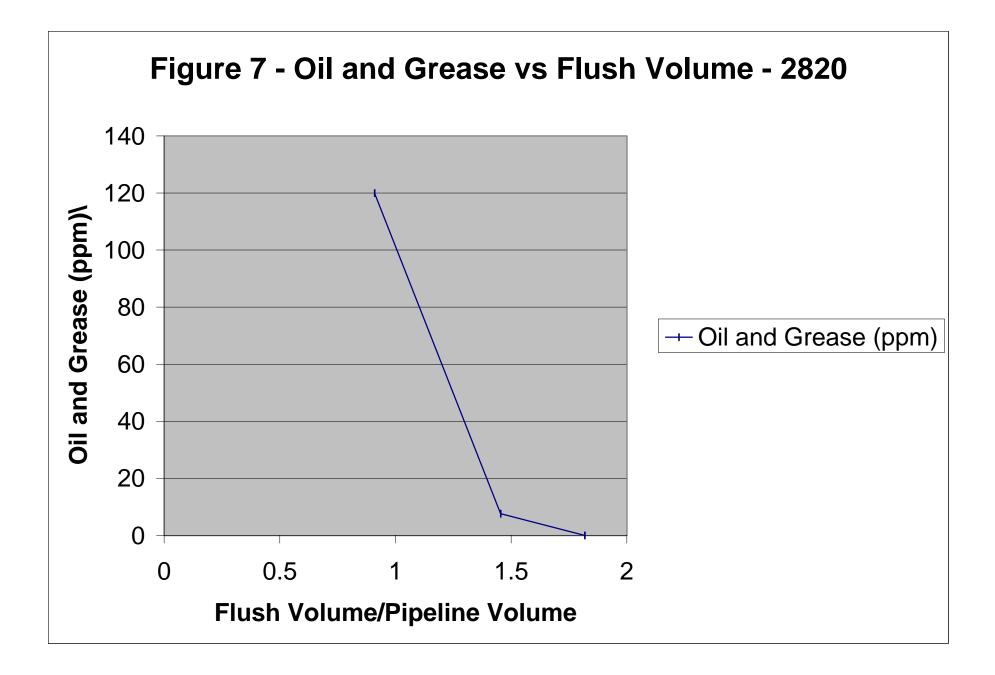


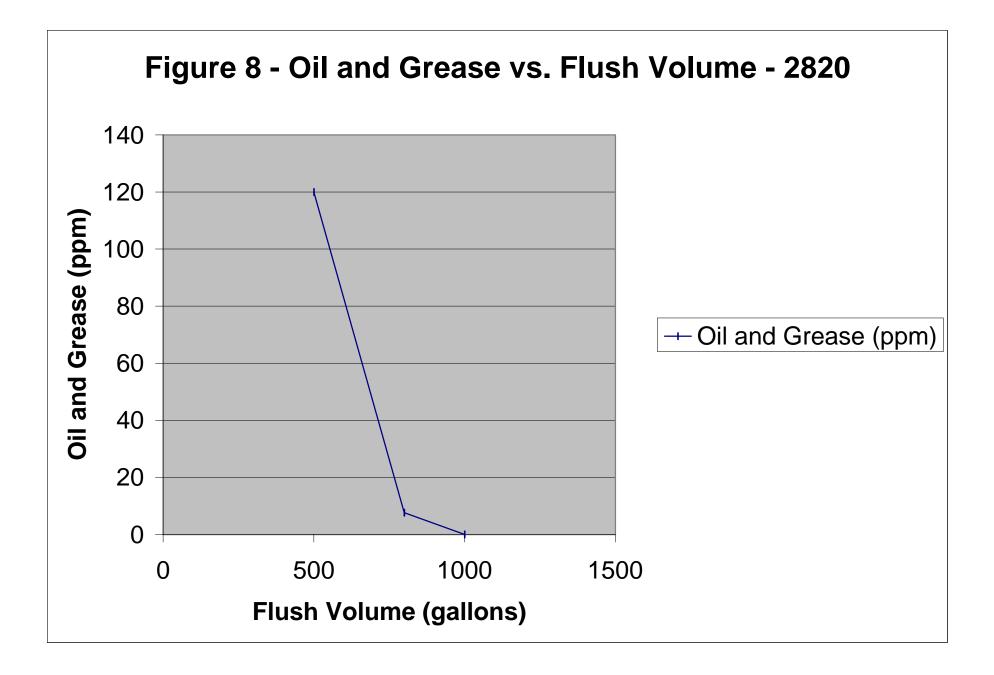
# Figure 5 - Chloride and Sulfate - 2820

Figure 6 - Iron Concentration - 2820



Sample #







PAGE 1 OF 3

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CERTIFICATE OF ANALYSIS NUMBER : 110003-008A

Sample ID.: 2820-1 : 10/26/00 MW0070

For : Winmar Attn: James Wiseman : 5700 NW Central Dr. Suit 150 : Houston, Texas 77092

TCD Analysis:

COMPONENTS	SAMPLE MOL %
Hydrogen	1,445
Carbon Dioxide	0.028
Ethylene	0.000
Ethane	1.802
Acetylene	0.000
Argon/Oxygen	0.469
Nitrogen	5.783
Methane	89.273
Carbon Monoxide	0.000

UnNormalized, Mol%	:	96.292		
Specific Gravity (Air = 1.000 @ 60F)	:	0.6054		
•		Net		Gross
BTU / ft3	:	886.0		982.8 Dry
( @ 14.65 & 60F)		870.6	Wet	965.7 Wet

SOUTHERN PETROLEUM LABORATORIES, INC.

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Fred C. DeAngelo



PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-008A

Sample ID.: 2820-1 : 10/26/00 MW0070

For : Winmar

Attn: James Wiseman : 5700 NW Central Dr. Suit 150 : Houston, Texas 77092

FID Analysis:

\_\_\_\_\_

	SAMPLE	
COMPONENTS	MOL %	
Hexanes Plus	0.530	
Methane	89.273	
Ethane/Ethylene	1.802 /	0.000
Propane	0.340	
Propylene	0.000	
iso-Butane	0.136	
Propadiene	0.000	
n-Butane	0.088	
Butene-1	0.000	
Neo-Pentane	None Detected	
iso-Butene	0.000	
trans-Butene-2	0.000	
cis-Butene-2	0.000	
1,3-Butadiene	0.000	
iso-Pentane	0.058	
n-Pentane	0.048	

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PAGE 3 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-008A

Sample ID.: 2820-1 : 10/26/00 MW0070

For : Winmar

Attn: James Wiseman : 5700 NW Central Dr. Suit 150 : Houston, Texas 77092

Completed Analysis:

Component	MOL 8			WT%		
Hydrogen	1.445			0.166		
Carbon Dioxide	0.028			0.070		
Carbon Monoxide	0.000			0.000		
Ethylene	0.000			0.000		
Acetylene/Propylene	0.000	1	0.000	0.000	1	0.00
Argon/Oxygen	0,469	1		0.855	'	
Nitrogen	5.783			9.245		
Methane	89.273			81.681		
Ethane	1.802			3.091		
Propane	0.340			0.856		
iso-Butane	0.136			0.450		
Propadiene	0.000			0.000		
n-Butane	0.088			0.291		
Butene-1	0.000			0.000		
Neo-Pentane	0.000			0.000		
iso-Butene	0.000			0.000		
trans-Butene-2	0.000			0.000		
cis-Butene-2	0.000			0.000		
1,3-Butadiene	0.000			0.000		
iso-Pentane	0.058			0.241		
n-Pentane	0.048			0.198		
Hexane Plus	0.530			2.857		
	100.000			100.000		

#### SOUTHERN PETROLEUM LABORATORIES, INC.

Fred C. DeAngelo

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PAGE 1 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-001A

Sample ID.: 2820-A : 10/26/00 WM0070

For : Winmar Attn: James Wiseman : 5700 NW Central Dr. Suite 150 : Houston, Texas 77092

TCD Analysis:

COMPONENTS	SAMPLE MOL %
Hydrogen	0.000
Carbon Dioxide	0.039
Ethylene	0.000
Ethane	1.510
Acetylene	0.000
Argon/Oxygen	5.529
Nitrogen	24.476
Methane	67.437
Carbon Monoxide	0.000

UnNormalized, Mol% : 93.771

Specific Gravity : (Air = 1.000 @ 60F)	0.7120	
	Net	Gross
BTU / ft3 : (@14.65 & 60F)	671.8 Dry 660.1 Wet	744.8 Dry 731.9 Wet

SOUTHERN PETROLEUM LABORATORIES, INC.

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Fred C. DeAngelo



PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-001A

Sample ID.: 2820-A : 10/26/00 WM0070

For : Winmar

- Attn: James Wiseman
  - : 5700 NW Central Dr. Suite 150
    - : Houston, Texas 77092

FID Analysis:

	SAMPLE	
COMPONENTS	MOL %	
Hexanes Plus	0.410	
Methane	67.436	
Ethane/Ethylene	1.510 /	0.000
Propane	0.300	
Propylene	0.000	
iso-Butane	0.125	
Propadiene	0.000	
n-Butane	0.079	
Butene-1	0.000	
Neo-Pentane	None Detected	
iso-Butene	0.000	
trans-Butene-2	0.000	
cis-Butene-2	0.000	
1,3-Butadiene	0.000	
iso-Pentane	0.052	
n-Pentane	0.044	

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#### PAGE 3 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 110003-001A

- Sample ID.: 2820-A : 10/26/00 WM0070
- For : Winmar
- Attn: James Wiseman
  - : 5700 NW Central Dr. Suite 150
  - : Houston, Texas 77092

Completed Analysis:

Component	MOL %			WT%		
Hydrogen	0.000			0.000		
Carbon Dioxide	0.039			0.083		
Carbon Monoxide	0.000			0.000		
Ethylene	0.000			0.000		
Acetylene/Propylene	0.000	1	0.000	0.000	1	0.000
Argon/Oxygen	5.529	•		8.577	•	
Nitrogen	24.476			33.260		
Methane	67.436			52.449		
Ethane	1.510			2.201		
Propane	0.300			0.642		
iso-Butane	0.125			0.351		
Propadiene	0.000			0.000		
n-Butane	0.079			0.223		
Butene-1	0.000			0.000		
Neo-Pentane	0.000			0.000		
iso-Butene	0.000			0.000		
trans-Butene-2	0.000			0.000		
cis-Butene-2	0.000			0.000		
1,3-Butadiene	0.000			0.000		
iso-Pentane	0.052			0.181		
n-Pentane	0.044			0.152		
Hexane Plus	0.410			1.881		
	100.000			100.000		

#### SOUTHERN PETROLEUM LABORATORIES, INC.

Fried C. DeAngelo



#### PAGE 1 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-002A

Sample ID.: 2820-B : 10/26/00 WM0070

For : Winmar Attn: James Wiseman : 5700 NW Central Dr. Suite 150 : Houston, Texas 77092

TCD Analysis:

\_\_\_\_\_

COMPONENTS	SAMPLE MOL %
	0.000
Hydrogen	0.000
Carbon Dioxide	1.517
Ethylene	0.000
Ethane	0.694
Acetylene	0.000
Argon/Oxygen	8.403
Nitrogen	60.044
Methane	28.448
Carbon Monoxide	0.000

UnNormalized, Mol% :	88.386		
Specific Gravity : (Air = 1.000 @ 60F)	0.8848		
•	Net		Gross
BTU / ft3 : (@14.65 & 60F)	303.6 298.3	-	336.1 Dry 330.2 Wet

SOUTHERN PETROLEUM LABORATORIES, INC.

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Fréd C. DeAngelo

11-30-00;14:18 ;SPL HOUSTON



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-002A

Sample ID.: 2820-B : 10/26/00 WM0070

For : Winmar

- Attn: James Wiseman
  - : 5700 NW Central Dr. Suite 150
    - : Houston, Texas 77092

FID Analysis:

from

	SAMPLE	
COMPONENTS	MOL %	
Hexanes Plus	0.462	
Methane	28.448	
Ethane/Ethylene	0.694 / 0.000	0
Propane	0.172	
Propylene	0.000	
iso-Butane	0.092	
Propadiene	0.000	
n-Butane	0.068	
Butene-1	0.000	
Neo-Pentane	None Detected	
iso-Butene	0.000	
trans-Butene-2	0.000	
cis-Butene-2	0.000	
1,3-Butadiene	0.000	
iso-Pentane	0.055	
n-Pentane	0.046	

SOUTHERN PETROLEUM LABORATORIES, INC.

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Fred C. DeAngelo



#### PAGE 3 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-002A

- Sample ID.: 2820-B : 10/26/00 WM0070
- For : Winmar
- Attn: James Wiseman : 5700 NW Central Dr. Suite 150 : Houston, Texas 77092
- Completed Analysis:

Component	MOL %		WT8	
	~~~~~~~~			
Hydrogen	0.000		0.000	
Carbon Dioxide	1.517		2.605	
Carbon Monoxide	0.000		0.000	
Ethylene	0.000		0.000	
Acetylene/Propylene	0.000	/	0.000 0.000 / 0.000	)
Argon/Oxygen	8.403		10.488	
Nitrogen	60.044		65.646	
Methane	28.448		17.802	
Ethane	0.694		0.814	
Propane	0.172		0.296	
iso-Butane	0.092		0.209	
Propadiene	0.000		0.000	
n-Butane	0.068		0.154	
Butene-1	0.000		0.000	
Neo-Pentane	0.000		0.000	
iso-Butene	0.000		0.000	
trans-Butene-2	0.000		0.000	
cis-Butene-2	0.000		0.000	
1,3-Butadiene	0.000		0.000	
iso-Pentane	0.055		0.156	
n-Pentane	0.045		0.127	
Hexane Plus	0.462		1.705	
	100.000		100.000	

#### SOUTHERN PETROLEUM LABORATORIES, INC.

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Fred C. DeAngelo



Client Sample ID 2820-D		Co	lected:		SPL Sample ID: 001	00898-06
		Site	: WN	10070		
Analyses/Method	Rəsult	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
OIL & GREASE, TOTAL RECOVI	ERABLE		MCL	E413.1	Units: mg/L	
Oil & Grease, Total Recoverable	120	2.0		1 E	11/06/00 9:00	461303

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MJ - Matrix Interforence 713 6608975;



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, YEXAS 77054 (713) 660-0901

Client Sample ID 2820-E		Col	lected:		SPL Sample ID: 0	0100898-07
		Site	a: WMO	)70		
Analyses/Method	Result	Rep.Limit	C	II. Factor QUAL	Date Analyzed Anal	yst Seq. #
OIL & GREASE, TOTAL RECOVE	RABLE	• ,	MCL	E413.1	Units: mg/L	
Oil & Grease, Total Recoverable	7.6	2.0		1	17/06/00 9:00	461305

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:58:28 AM



Client Sample ID 2820-F		Col	lected:			SPL Sample ID: 001	00898-08
		Site	: WM	0070			
Analyses/Method	Result	Rep.Limit		Dil. Factor QU	AL	Date Analyzed Analys	t Seq. #
OIL & GREASE, TOTAL RECOV	ERABLE		MCL	E413	.1	Units: mg/L	
Oil & Grease, Total Recoverable	ND	2.0	•	1		11/06/00 9:00	461306

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution

MI - Matrix Interference



Client Sample ID 282	0-4	Col	ected:		SPL Sample ID: 00100896-		
		Site	9: <b>HI</b> 1	135/136			
Analyses/Method	Resul	it Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #	
ALKALINITY, BICARE Alkalinity, Bicarbonate	ONATE 27	9 2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	461528	
ALKALINITY, CARBO Alkalinity, Carbonate	NATE NI	<b>)</b> <u>2</u>	MCL	<b>M2320 B</b>	Units: mg/L 11/01/00 14:00 SN	460250	
CHLORIDE, TOTAL Chloride	10600	0 250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	458319	
METALS BY METHOD	60108, TOTAL		MCL	SW6010B	Units: mg/L		
Barium	13.1			1	11/10/00 22:17 E_B	471285	
Calcium	84.4			1	11/10/00 22:17 E_B	471285	
Iron	91.	· · · · · · · · · · · · ·		1	11/10/00 22:17 E_B	471285	
Magnesium	38.			10	11/13/00 16:22 JM	471771	
Potassium	51.	5 20		10	11/13/00 16:22 JM	47177	
•	1001110C-471285						
Prep Method	Prep Date	Prep Initials					
SW3010A	11/01/2000 8:30	MR					
Run ID/Sed #: TJ/ Prep Method	001113B-471771	Preo Initials	i				
SW3010A	11/01/2000 8:30	MR					
				ni ing ning ning ning ning ning ning ni			
PH	7	5 <b>0</b> .10	MCL.	E150.1	Units: pH Units 11/01/00 16:00 EC	458741	
рH	7,			- 1 1		+307+1	
RESISTANCE @ 25 C Resistance	N		MCL	<b>120.1</b>	Units: Mohms/cm 11/03/00 9:15 C_V	461403	
SPECIFIC GRAVITY		anna an tar a san tar Tar a san tar a san ta	MCL	ASTM D-1429	Units: Specific Gr	avity @	
Specific Gravity	1.01	4 0		1	11/06/00 11:00 C_V	462235	
SULFATE, TOTAL Sulfate			MCL	<b>E375.4</b> 1	Units: mg/L 11/01/00 10:00 SN	458831	
TOTAL DISSOLVED S Total Dissolved Solids,		0 10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471972	
TOTAL SODIUM, CAL Total Sodium, Calculate		0 10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471989	
TOTAL SUSPENDED Suspended Solids (Res Filterable)	and the second	8 8	MCL	E160.2 2	Units: mg/L 11/03/00 15:45 EC	462121	

Qualifiers:

and a submanifier of the second second

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits
- J Estimated Value between MDL and PQL

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>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution

MI - Matrix Interference

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Client Sample ID 282	20-5		Col	liected:	10/27/00	SPL Sample ID: 00100	896-08
			Site	ə: Hi1	35/136		
Analyses/Method	R	esult	Rep.Limit		Dil-Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICAR	BONATE			MCL	M2320 B	Units: mg/L	
Alkalinity, Bicarbonate		192	2		1	11/01/00 14:00 SN	46152
ALKALINITY, CARBO	NATE			MCL	M2320 B	Units: mg/L	
Alkalinity, Carbonate		ND	2		1	11/01/00 14:00 SN	46025
CHLORIDE, TOTAL				MCL	E325.3	Units: mg/L	
Chloride	1	6100	250		250	11/08/00 11:00 CV	46832
		÷				n an	
METALS BY METHOD Barium		0,216	0.005	MCL	SW6010B	Units: mg/L 11/10/00 22:32 E_B	47128
Calcium		409	0,1		1	11/10/00 22:32 E_B	47128
Iron		3.49	0.02		1	11/10/00 22:32 E_B	47128
Magnesium	•	1310	1	•••	10	11/13/00 16:26 JM	47177
Potassium		385	20		10	11/13/00 16:26 JM	47177
/	A_001110C-471286	<u>.</u>		1			
Prep Method	Prep Date		Prep Initials				
SW3010A	11/01/2000 8:30		MR				
Prep Method	A_001113B-471772 Prep Date	<b>i</b>	Prep Initials	1			
SW3010A	11/01/2000 8:30		MR				
<b>D</b> U			· · · · ·				· <del>-</del> ··-
Р <b>Н</b> рН	•	7.8	0.10	MCL	E150.1	Units: pH Units 11/01/00 16:00 EC	45874
. ·		1.0		ene i se e			10014
RESISTANCE @ 25 C				MCL	120.1	Units: Mohms/cm	
Resistance	• • • • • • • • • • • • • • • • • • •	ND	0.00100	<u>.</u> .	<b>1</b>	11/03/00 9:15 C_V	46140
SPECIFIC GRAVITY				MCL	ASTM D-1429	Units: Specific Grav	ity @
Specific Gravity		1.032	0		1	11/06/00 11:00 C_V	46223
SULFATE, TOTAL				MCL	E375.4	Units: mg/L	
Sulfate		6600	1000		1000	11/01/00 10:00 SN	45883
TOTAL DISSOLVED S				MCL	TDS-MINERAL	Units: mg/L	
Total Dissolved Solids		5500	10		1	11/13/00 18:00 ES	47197
			• • •••			·····	
TOTAL SODIUM, CAL Total Sodium, Calculate		0500	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	47199
	PM 1		1V				4/135
TOTAL SUSPENDED	•••••••••••••••••••••••••••••••••••••••		· ·	MCL	E160.2	Units: mg/L	
Suspended Solids (Res	idue,Non-	789	4		1	11/02/00 15:00 EC	461982
Filterable)					• • • • • • • • • •	· · · · • • • • • • • • • • • • • • • •	<b>-</b> ·

# Qualifiers:

ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

. . . . ..



Cilent Sample ID 2820	)-6	Co	llected:	10/27/00	SPL Sample ID: (	0100896-09
		Sit	e: Hill	135/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Anal	
ALKALINITY, BICARB Alkalinity, Bicarbonate	<b>ONATE</b> 152	2	MCL	<b>M2320 B</b>	Units: mg/L 11/01/00 14:00 SN	46153
ALKALINITY, CARBOI Alkalinity, Carbonate	ND ND	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	460253
CHLORIDE, TOTAL Chloride	16500	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468322
METALS BY METHOD	60108, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	0.0693	0.005		1	11/10/00 22:36 E_B	471287
Calcium	431	0.1		1	11/10/00 22:36 E_B	471287
Iron	0.387	0.02		1	11/10/00 22:36 E_B	471287
Magnesium	1330	1		10	11/13/00 16:31 JM	471773
Potassium	390	20		10	11/13/00 16:31 JM	471773
Run ID/Seq #: TJA						
Prep Method	Prep Date	Preo Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA						
Prep Method	Prep Date	Prep Initials				
SW301DA	11/01/2000 8:30	MR	İ			
PH	······································		MCL	E150.1	Units: pH Unit	
pH	8.1	0.10		1	11/01/00 16:00 EC	458744
RESISTANCE @ 25 C Resistance	0.015	0.00100	MCL	1 <u>20.1</u>	Units: Mohms 11/03/00 9:15 C_V	cm 461407
	nan na an an ann an an an an an an an an		MACI	ASTM D-1429		Canada a
SPECIFIC GRAVITY Specific Gravity	1.016	0	MCL	A31NI D-1428	Units: Specific 11/06/00 11:00 C_V	462237
SULFATE, TOTAL Sulfate	6500	1000	MCL	E375.4 1000	Units: mg/L 11/01/00 10:00 SN	458834
TOTAL DISSOLVED S	•••	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471974
TOTAL SODIUM, CAL			MCL	TDS-MINERAL	Units: mg/L	
Total Sodium, Calculater	d 10600	10		1	11/13/00 18:00 ES	471991
				*** * * * * * * * *	•• •	
TOTAL SUSPENDED Suspended Solids (Resident		4	MCL	E160.2	Units: mg/L 11/02/00 15:00 EC	461983

Qualifiers:

......

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Diarik
- \* Surrogate Recovery Outside Advisable QC Limits
- J Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference



# 6. Results and Observations – 2822

# a. Sample Locations Observations

This pipeline was flushed and filled with seawater in October 1994. It was reflushed and abandoned in place in October 2000. Samples were taken of the entrained water that had been sitting in the pipe for 6 years, as well as the volume of flushwater that was run through the pipe. The sampling location for this pipeline was at the "sampling spoolpiece," which was connected at the platform cellar deck level.

A five foot (5') sample of the pipeline was removed, which included the tubeturn to pipeline weld.

# b. Gas Composition Observations

This was a gas/condensate line, however, there were no gas samples taken, as the line was filled completely with seawater.

# c. Flushwater Composition Observations

# i. Mineral Pattern Analysis

Samples A-E were taken from the standing water in the pipeline, while F-H were taken from the flush water. These discreet sample types are evident in the plotted data.

For the water that stood in the pipeline for 6 years, the mineral pattern relative to NSW is summarized below:

Alkalinity (bicarb) – Lower Barium – Equal Calcium – Lower Iron – Higher Magnesium – Lower Potassium – Similar/Equal Chloride – Lower Sulfate – Lower

For the flushwater, the mineral pattern relative to NSW is summarized below:

Alkalinity (bicarb) – Equal Barium – Equal Calcium – Equal



Iron – Higher Magnesium – Equal Potassium – Similar/Equal Chloride – Lower/Equal Sulfate – Equal

The plot of the Iron Concentration is of particular interest for this pipeline. As the standing water was pushed out of the pipeline, the sample concentrations rose sharply and linearly until all of the entrained water was pushed out. The highest value reached was 76 .1 PPM. The flushwater showed iron concentrations in the range of 0.4 - 1.29 PPM, which is much higher than natural seawater concentrations, but nothing like the levels from the entrained water. These flushwater concentrations were lower than for other pipelines, perhaps because many of the loose iron particles and all of the dissolved iron had already been pushed out of the line.

# ii. Nitrogen and Oxygen Concentration

These values were tested in order to determine whether corrosion inhibitor was present in the line, and if it was, its effectiveness. In all samples, dissolved nitrogen levels were much lower than NSW. This indicated that an Amine based corrosion inhibitor was not present. It was interesting to note that nitrogen levels never reached NSW level, not even in the flushwater.



Photo #16

Dissolved oxygen levels were slightly lower than NSW in the entrained water. Interestingly, dissolved oxygen concentrations were higher than NSW levels in the flushwater. This data shows that an oxygen scavenger was not used in this pipeline when it was flushed and filled the first time.

# d. Oil and Grease Observations

The first oil and grease sample was taken from the standing water in the pipeline, and the two following samples were taken from the flushwater. Of interest is the fact that oil



and grease was present in the first sample at 270 PPM. These levels dropped away almost immediately during flushing, and was non-detectable by the time the pipeline was flushed 1.5 times.

# e. Pipe Cutout Observations

It is clear from the photographs of sample segment 2822, that there have been significant changes to the pipe wall since it was filled with seawater. Based on the linear features of the debris in the pipe, it is evident that there was an air/water interface at some point in time after the line was filled. The pipe walls are coated with debris, the majority of which does not appear to be a corrosion product. One possibility is that seafloor mud was sucked into the pump intake and pumped into the pipe during initial flushing - a likely scenario in shallow water, with very turbid conditions. This is an important consideration for flushing and filling out-of-service pipelines. Pipelines that are required to be filled with inhibited seawater may be negatively impacted by the addition of these sediments. Better procedures may be required to ensure that sediments are not introduced during flushing and filling operations.



Photo #17

The line also showed significant metal loss corrosion. The weld at the tubeturn to pipeline connection also showed deep pitting corrosion.





Photos #18 and #19 - Pipeline Sample Photos - 2822





# SHELL OFFSHORE INC. HI-135-2

PLATFORM									
MMS Gene	eral	0	DS Genera	al	MMS Lo	cation	MMS Fa	MMS Facility	
Water	50 feet	Function	WP		Lease	741 feet	Helideck	Yes	
Major	No	Piles	NA		Complex	10015	Quarters	None	
Decks	1	Slots	3		Longitude	-94.112	Generator	No	
Slots	3	ODS ID	739		Latitude	29.260	Cranes	NA	
Wells	2	Previous 1	NA		Х	3,558,137'	Gas	Yes	
Flare	No	Previous 2	NA		Y	551,391'	Oil	Yes	
Installed	01 1964	Previous 3	NA		To Shore	25 miles	Comp	No	
Revised	05 1998	Previous 4	NA		N-S feet	S 4911'	8 hour	No	
Removed	NA	Notes	NA		E-W feet	W 3141'	24 hour	No	
			PIF	PELINES	MMS				
Segment	2821		2822		2823				
Origin	HI-135-2		HI-135-2		HI-135-2		#N/A		
Terminus	HI-136-A		HI-136-A		HI-136-A		#N/A		
O.D.	3"		4"		4"		#N/A		
Length	4,000'		4,000'		4,000'		#N/A		
Product	BLKG		BLKG		BLKG		#N/A		
Status	ACT		OUT		PABN		#N/A		
Installed	NA		NA		NA		#N/A		
Abandon	NA		NA		NA		#N/A		
Revised	Aug-94		Oct-94		Aug-94		#N/A		
Operator	SHELL O	FFSHORE INC	SHELL C	FFSHORE	SHELL OFF	SHORE INC.	#N/A		
			V	VELLS N	IMS				
API Well ID	Well	Spud	Revised	Status	MD	Bot Lease	Sur Long	Sur Lat	
427080004600	1	03 1964	05 1964	COM	10,777'	741	-94.119	29.259	
427080004700	2	10 1964	05 1986	ST	8,678'	741	-94.112	29.260	
427080004701	2	05 1986	07 1986	COM	9,450'	741	-94.115	29.260	
427080004800	3	11 1964	08 1985	PA	9,663'	741	-94.104	29.265	
427080004900	4	11 1964	06 1990	PA	11,483'	741	-94.116	29.269	
427080005000	5	04 1965	08 1985	PA	9,006'	741	-94.089	29.262	
427080006100	6	06 1965	06 1965	PA	9,085'	741	-94.089	29.271	
427080007300	8	07 1967	08 1967	ST	9,452'	741	-94.103	29.260	
427080007301	9	08 1967	08 1967	COM	9,020'	741	-94.106	29.260	
427084023300	11	04 1986	01 1996	ST	10,824'	741	-94.075	29.262	
427084023301	11	01 1996	04 1996	COM	11,033'	741	-94.075	29.262	
427084023400	9	08 1986	09 1986	TA	10,230'	742	-94.132	29.259	

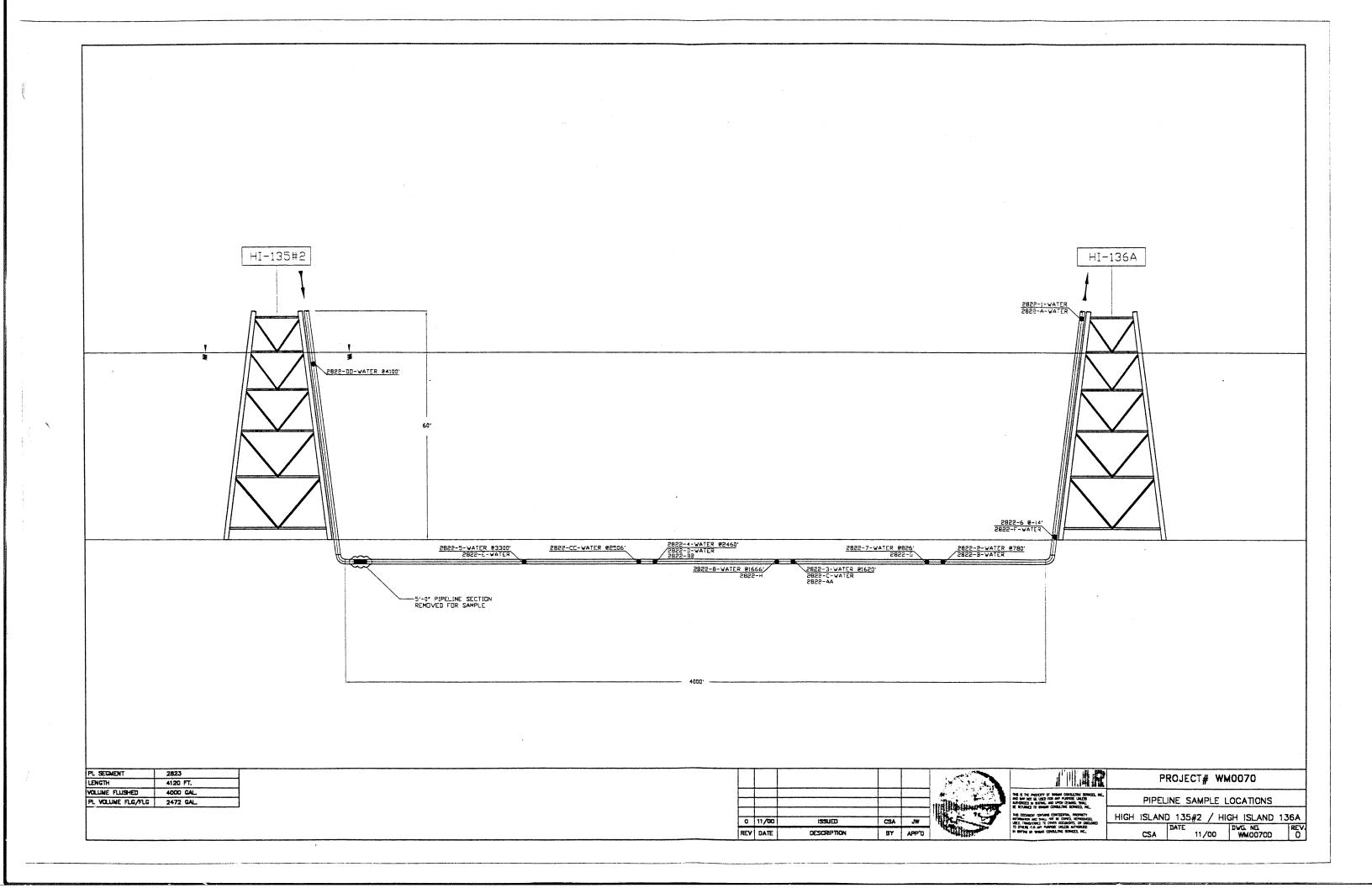


### PIPELINE FLUSHING AND SAMPLING RECORD

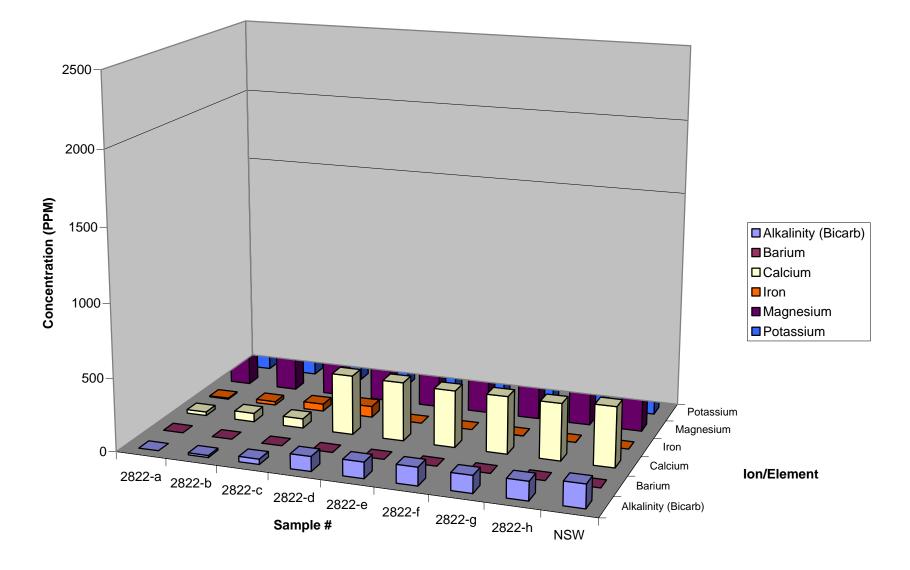
Dinalina Information	
I. Pipeline Information	2022
MMS Segment No.	2822 10/27/2000
Date: Pipeline Origination	
	Llich Island
Area	High Island
Block	135
Platform	#2 OCS-G-0741
Lease Displing Destinction	065-6-0741
Pipeline Destination	Llich Jolond
Area	High Island
Block	136
Platform	A 000 0 0742
Lease	OCS-G-0742
Pipeline Size (in)	4
Pipelines Length (ft)	4,000
Pipeline Volume (bbls)	62
II. Eluching Information	
II. Flushing Information	
Flushing Information Volume Flushed	4000-
	4000g
Flow Rate (GPM)	100
Pigged Used	No
Type of Pig	No
Size of Pig	
Clean Returns	Yes
Inhibitor	
Chemical Inhibitor Used	
Type of Chemical	
Quantity of Chemical	
Origination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Destination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Comments:	
Company Representative	
Cianatura	
Signature	



Sample Location	ormation			
Platform:	Hi 136A			
Pipeline Sampling Site:	Hose Connec	tion w/ Sampling Spo	ool	
Flushing Start Time:	13:00			
Water Šamples	Sample ID	Notes		
Plastic Bottles - Zero Head Space	2822-1	Sample Date 10/27/2000	Vol. Flushed (g) 0	
·	2822-2	10/27/2000	500	
	2822-3	10/27/2000	1,000	
	2822-4	10/27/2000	1,500	
	2822-5	10/27/2000	200	
	2822-6	10/27/2000	2,500	
	2822-7	10/27/2000	3,000	
	2822-8	10/27/2000	3,500	
Water Samples	Sample ID	Sample Date	Vol. Flushed (g)	Notes
Vineral Pattern Analysis	2822-A	10/27/2000	0	
	2822-B	10/27/2000	500	
	2822-C	10/27/2000	1,000	
	2822-D	10/27/2000	1,500	
	2822-E	10/27/2000	2,000	
	2822-F	10/27/2000	2,500	
	2822-G	10/27/2000	3,000	
	2822-H	10/27/2000	3,500	
Oil and Grease Analysis	2822-AA	10/27/2000		
	2822-BB	10/27/2000	1,500	
	2822-CC	10/27/2000	4,000	
	2822-DD	10/27/2000	5,000	
Comments: Company Representative	plugged, and	added flange for flus	ded over riser top, no f shing.	lange. We cold-cu
	James Wiser	1011		
	1			



# Figure 10 - Flushwater Composition - 2822



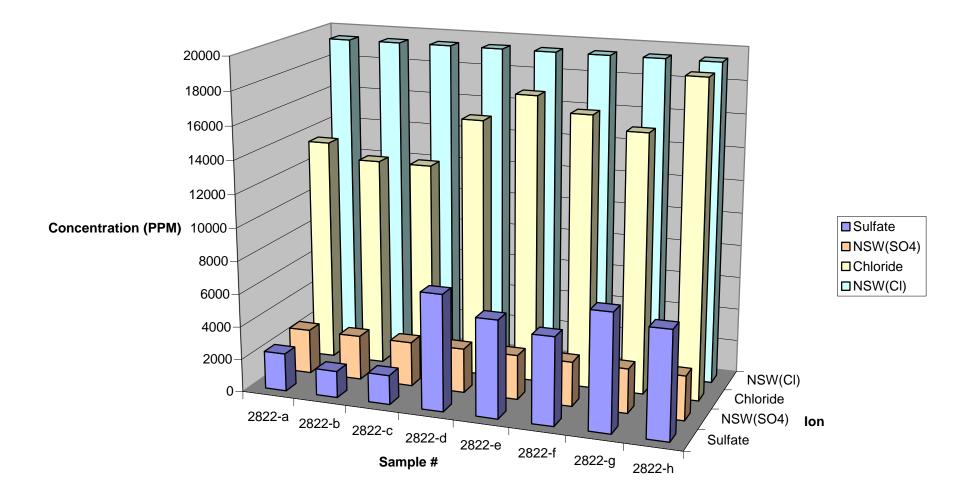


Figure 12 - Iron Concentration - 2822

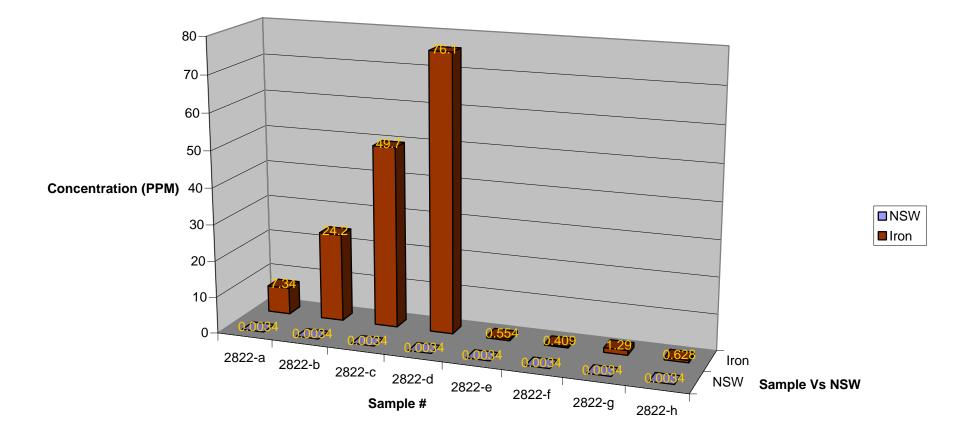


Figure 13 - Oxygen Concentration - 2822

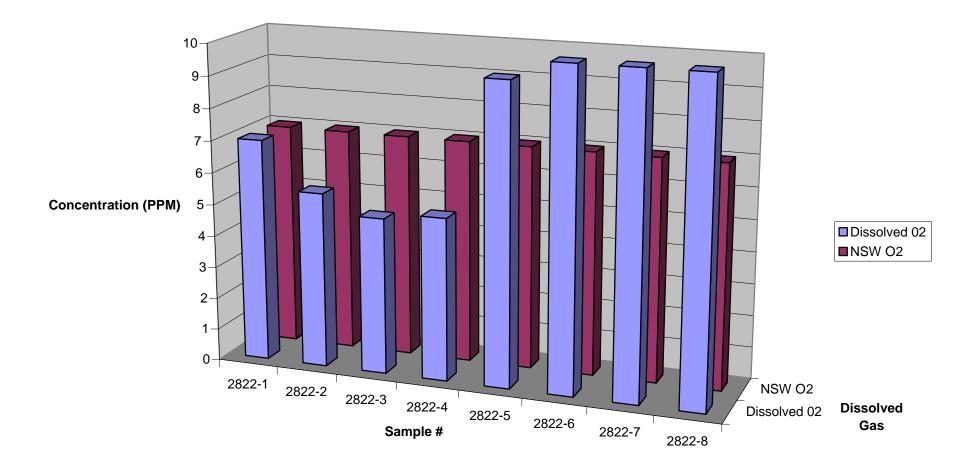


Figure 14 - Nitrogen Concentration - 2822

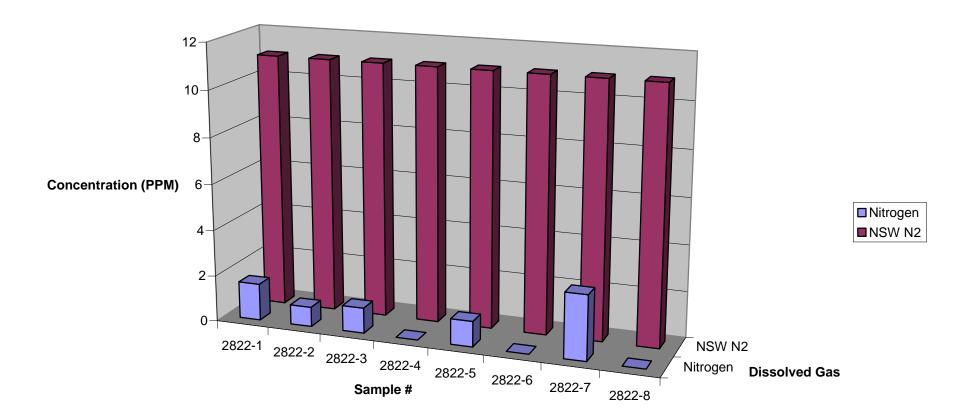
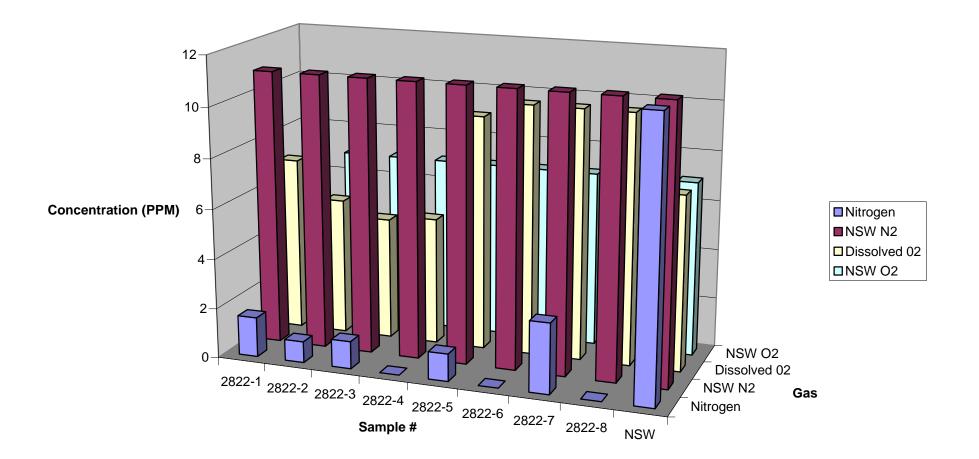
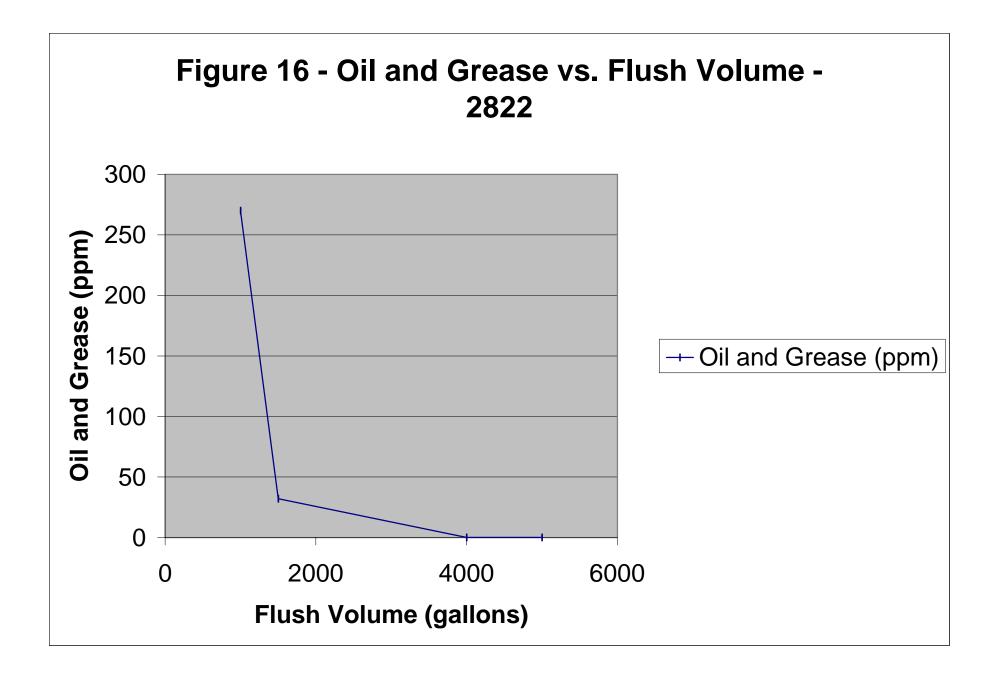
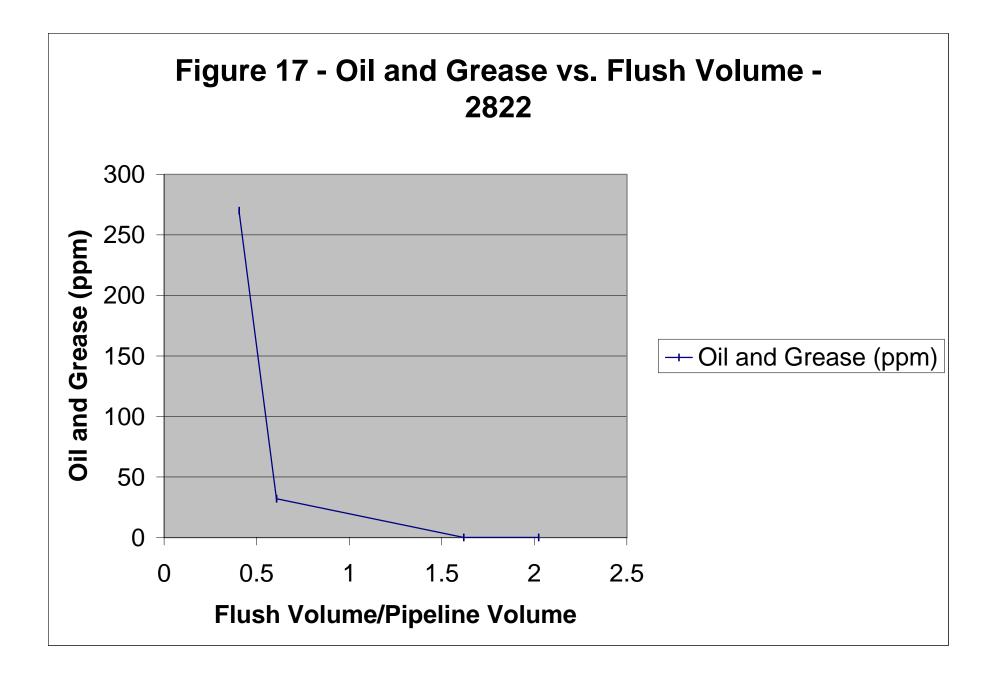


Figure 15 - Dissolved O2 and N2 - 2822







713 6608975;



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 650-0901

Client Sample ID 2822-AA		Collected: SPL Sample ID: 00				100898-09	
Site: WM0070							
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #	
OIL & GREASE, TOTAL RECOV	ERABLE		MCL	E413.1	Units: mg/L		
Oil & Grease, Total Recoverable	270	2.0		1 E	11/06/00 9:00	461307	

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:55:29 AM

713 6608975;



HOUSTON LABORATORY 8480 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 560-0901

Client Sample ID 2822-BB		Col	lected:		SPL Sample ID: 0010			
Site: WM0070								
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed A	nalyst Seq. #		
OIL & GREASE, TOTAL RECOV	ERABLE		MCL	E413.1	Units: mg/L			
Oil & Grease, Total Recoverable	32	2.0		1 E	11/06/00 9:00	46130		

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:56:29 AM



Client Sample ID 2822-CC		Collected:			SPL Sample ID: 0010	0898-11	
Site: WM0070							
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #	
OIL & GREASE, TOTAL RECOVE	ERABLE		MCL	E413.1	Units: mg/L		
Oil & Grease Total Recoverable	ND	2.0		1	11/06/00 9:00	461310	

Qualifiers;

. ..... ND/U - Not Detected at the Reporting Limit

.. .

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:56:29 AM



Client Sample ID 2822-DD		Collected: SPL Sample ID: 00					
Site: WM0070							
Analyses/Method	Result	Rep.Limit	· · · · · · · · · · · · · · · · · · ·	DII. Factor QUAL	Date Analyzed Analyst	Seq. #	
OIL & GREASE, TOTAL RECOV	ERABLE		MCL	E413.1	Units: mg/L		
Oil & Grease, Total Recoverable	ND	2.0	.,	1	17/06/00 9:00	461311	

Qualifiers:

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Blank \* - Surrogate Recovery Outside Advisable QC Limits
- J Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference



## Winmar Consulting Services

Certificate of Analysis Number:

### 00100895

<u>Report To:</u>	Winmar Consulting James Wiseman 5700 NW Central Dr Suite 150 Houston			<u>Project Name</u> <u>Site:</u> <u>Site Address</u>	HI 135/136 WM0070	
	TX 77092- ph: (713) 895-8240	fax: (713) 8:	95-8270	<u>PO Number:</u> <u>State:</u> <u>State Cert. N</u>	WM0070 a.:	
<u>Fax To:</u>	Winmar Consulting James Wiseman	Services fax : (713)	895-8270	Date Reporte	<u>.d:</u> 11/14/00	
Clien 2822-A 2822-B 2822-C 2822-C 2822-C 2822-E 2822-F 2822-F -H	nt Sample ID	Lab Sample ID 00100895-01 00100895-02 00100895-03 00100895-04 00100895-05 00100895-06 00100895-06 00100895-07 00100895-08	Matrix Water Water Water Water Water Water Water Water	Date Collected 10/28/00 10/28/00 10/28/00 10/28/00 10/28/00 10/28/00 10/28/00 10/28/00 10/28/00	Date Received 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM 10/30/00 12:05:00 PM	COCID

Neschich.Paul Senior Project Manager 11/20/00

Date

HOLD

Joel Grice Laboratory Director

Ted Yen Quality Assurance Officer



Client Sample ID 2822-A		Colle	cted:	10/28/00	SPL Sample ID: 0010	0895-01
		Site:	HI 1	35/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Scq. #
ALKALINITY, BICARBON Alkalinity. Bicarbonate	ATE ND	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	461501
ALKALINITY, CARBONA Alkalinity, Carbonate	<b>TE</b> 121	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460235
CHLORIDE, TOTAL Chloride	13600	500	MCL	<b>E325.3</b> 500	Units: mg/L 11/08/00 11:00 CV	468300
METALS BY METHOD 60	10B, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	0.0527	0.005		1	11/10/00 20:15 E_B	471265
Calcium	25.2	0.1		1	11/10/00 20:15 E_B	471265
Iron	7.34	0.02		1	11/10/00 20:15 E_B	471265
Magnesium	821	0.5		5	11/13/00 14:46 JM	471748
Potassium	388	2.		1	11/13/00 14:41 JM	471747
sector in the sector of the se	ep Date	Prep Initials				
SW3010A 11	/01/2000 8:30	MR				
PH pH	9.6	0.10	MCL.	E150.1	Units: pH Units 11/01/00 16:00 EC	458724
RESISTANCE @ 25 C Resistance	ND	0.00100	MCL	<b>120.1</b>	Units: Mohms/cm 11/03/00 9:15 C_V	461385
SPECIFIC GRAVITY Specific Gravity	1.02	· O	MCL	ASTM D-1429 1	Units: Specific Gr 11/06/00 11:00 C_V	avity@ 462220
SULFATE, TOTAL Sulfate	2300	250	MCL	<b>E375.4</b> 250	Units: mg/L 11/01/00 10:00 SN	458812
TOTAL DISSOLVED SOL Total Dissolved Solids, Calc		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471958
TOTAL SODIUM, CALCU Total Sodium, Calculated	<b>LATED</b> 8200	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471975
TOTAL SUSPENDED SOI Suspended Solids (Residue Filterable)		4	MCL	<b>E160.2</b>	Units: mg/L 11/02/00 15:00 EC	461 <del>9</del> 68

Qualifiers:

. .

... . . . ......

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Mothod Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

.

>MCL - Result Over Maximum Contamination Limit(MCL) D · Surrogate Recovery Unreportable due to Dilution

MI - Matrix Interference



Client Sample ID 2822	-B	Coli	ected:	10/28/00	SPL Sample ID: 00100	895-02
		Site:	HI 1	135/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARBO Atkalinity Bicarbonate	<b>DNATE</b> 14.1	2	MCL	<b>M2320 B</b> 1	Units: mg/L 11/01/00 14:00 SN	<b>46</b> 1504
ALKALINITY, CARBON Alkalinity, Carbonate	<b>IATE</b> 56.6	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	460236
CHLORIDE, TOTAL Chloride	12700	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468303
METALS BY METHOD	6010B. TOTAL	• • • • •	MCL	SW6010B	Units: mg/L	
Barium Calcium	0.0703 59.3	0.005		1	11/10/00 20:20 E_B 11/10/00 20:20 E_B	471266 471266
tron Magnesium	24.2 904	0.02 0.5		1	11/10/00 20:20 E_B 11/13/00 14:54 JM	471266 471750
Potassium	368	2		1	11/13/00 14:50 JM	471749
	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
<b>РН</b> pH	9.2	0.10	MCL	<b>E150.1</b>	Units: pH Units 11/01/00 16:00 EC	458726
RESISTANCE @ 25 C Resistance	0.0052	0.00100	MCL	120.1 1	Units: Mohms/cm 11/03/00 9:15 C_V	461387
SPECIFIC GRAVITY Specific Gravity	1.02	0	MCL	ASTM D-1429 1	Units: Specific Gra 11/06/00 11:00 C_V	vity @ 462221
SULFATE, TOTAL Sulfate	1620	250	MCL.	<b>E375.4</b> 250	Units: mg/L 11/01/00 10:00 SN	458815
TOTAL DISSOLVED SC Total Dissolved Solids, C	DLIDS	10	MCL	TDS-MINERAL 1	Units: mg/L 11/13/00 18:00 ES	471959
TOTAL SODIUM, CALC Total Sodium, Calculated		10	MCL	TDS-MINERAL	Units: mg/L. 11/13/00 18:00 ES	471976
TOTAL SUSPENDED S Suspended Solids (Resid Filterable)		4	MCL	<b>E160.2</b> 1	Units: mg/L 11/02/00 15:00 EC	461971

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Lim.ts

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unrepenable due to Dilution MJ - Matrix Interference



Client Sample ID 2822-A		Coll	ected:	10/28/00	SPL Sample ID:	00100	00895-01			
		Site	: HI	135/136 WM0070						
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #			
TOTAL SUSPENDED SOLIDS	417	MCL E160.2 Units: mg/L								
Suspended Solids (Residue,Non- Filterable)	117	4		1	11/02/00 15:00		461968			

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution Mi - Matrix Interference

11/15/00 11:20:38 AM



Client Sample ID 2822-B		Coll	ected:	10/28/00	SPL Sample ID:	0010	0895-02
		Site	: <b>Hi</b> 1	35/136 WM0070			
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
TOTAL SUSPENDED SOLIDS	•		MCL	E160.2	Units: mg	1L	
Suspended Solids (Residue,Non- Filterable)	208	4		1	11/02/00 15:00	ËC	461971

Qualifiors:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

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>MCL - Result Over Maximum Contamination Limit(MCL) D - Burroyate Recovery Unreportable due to Difution

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MI - Matrix Interference

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Client Sample ID 282	2+C	Coll	ected:	10/28/00	SPL Sample ID: 00100	895-03
		Site	: HJ 1	35/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARB		· ,	MCL	M2320 B	Units: mg/L	
Alkalinity, Bicarbonate	36.4	2		1	11/01/00 14:00 SN	46150
ALKALINITY, CARBOI	NATE		MCL	M2320 B	Units: mg/L	
Alkalinity, Carbonate	44.4	2	•••	1	11/01/00 14:00 SN	46023
CHLORIDE, TOTAL			MCL	E325.3	Units: mg/L	
Chloride	12700	250		250	11/08/00 11:00 CV	46830
METALS BY METHOD	6010B TOTAL		MCL	SW6010B	Units: mg/L	
Barium	0.0457	0.005		1	11/10/00 20:34 E_B	47126
Calcium	64.9	0.1		1	11/10/00 20:34 E_B	47126
Iron	49.7	0.02		1	11/10/00 20:34 E B	47126
Magnesium	930	0.5		5	11/13/00 15:12 JM	47175
Potassium	371	2		1	11/13/00 15:08 JM	47175
Run ID/Seq #: TJA	_001110C-471269					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA	_001113B-471754					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA		1				
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
PH			MCL	E150.1	Units: pH Units	
pH	8.8	0.10		1	11/01/00 16:00 EC	45872
RESISTANCE @ 25 C			MCL	120.1	Units: Mohms/cm	
Resistance	ND	0.00100		1	11/03/00 9:15 C_V	46138
SPECIFIC GRAVITY	,		MCL	ASTM D-1429	Units: Specific Gra	uitv (D)
Specific Gravity	1.012	0		1	11/06/00 11:00 C_V	46222
SULFATE, TOTAL		:	MCL	E375.4	Units: mg/L	· · ·
Sulfate	1780	250		250	11/01/00 10:00 SN	45881
TOTAL DISSOLVED S			MCL	TDS-MINERAL	Units: mg/L	··· ·
Total Dissolved Solids, (		10		1	11/13/00 18:00 ES	47196
TOTAL SODIUM, CAL		:	MCL	TDS-MINERAL	Units: mg/L	•

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Receivery Unreportable due to Dilution MI - Matrix Interference

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Client Sample ID 2822-C		Col	lected:	10/28/00	SPL Sample ID:	0010	0895-03
		Site	: HI	135/136 WM0070			
Analyses/Nethod	Result	Rep.Limit	2	Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
TOTAL SUSPENDED SOLIDS	÷.		MCL	E160.2	Units: mg/	L	••••
Suspended Solids (Residue, Non- Filterable)	424	8		2	11/02/00 15:00 E	EC	461972

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limil(MCL) D - Surrogate Recovery Unreportable due to Dilution Mi - Matrix Interference



Client Sample ID 2822-D		Colle	ected:	10/28/00	SPL Sample ID: 0010	0895-04
	· · · ·	Site:	HI 1	35/136 WM0070		
Analyses/Nethod	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARBONATE Alkalinity, Bicarbonate	101	2	WCL	<b>M2320 B</b>	Units: mg/L 11/01/00 14:00 SN	461506
ALKALINITY, CARBONATE Alkalinity, Carbonate	20.2	2	MCL	<b>M2320 B</b>	Units: mg/L 11/01/00 14:00 SN	460238
CHLORIDE, TOTAL Chioride	15700	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468305
METALS BY METHOD 6010B, T Banum	OTAL 0.0817	0.005	MCL	SW6010B	Units: mg/L 11/10/00 20:38 E_B	471270
Calcium Iron	407 76.1	0.1		1	11/10/00 20:38 E_B 11/10/00 20:38 E_B	471270 471270
Magnesium Polassium	2420 462	1 2		10 1	11/13/00 15:21 JM 11/13/00 15:16 JM	471757 471756
Run ID/Seq #: TJA_001110C           Preo Method         Preo Date           SW3010A         11/01/200           Run ID/Seq #: TJA_001113B-	0 8:30	Prep Initiats MR				
Prep Method         Prep Date           SW3010A         11/01/200           Run ID/Seg #: TJA_001113B	0 8:30	Prep Initials MR				
Prep Method Prep Date SW3010A 11/01/200		Prep Initials MR				
РН рН	8.4	0.10	MCL.	E150.1	<b>Units: pH Units</b> 11/01/00 16:00 EC	458728
RESISTANCE @ 25 C Resistance	0.0093	0.00100	MCL	<b>120.1</b>	Units: Mohms/cm 11/03/00 9:15 C_V	461389
SPECIFIC GRAVITY Specific Gravity	1.034	0	MCL	<b>ASTM D-1429</b> 1	Units: Specific Gr 11/06/00 11:00 C_V	avity@ 462223
SULFATE, TOTAL Sulfate	7040	1000	MCL	<b>E375.4</b>	Units: mg/L 11/01/00 10:00 SN	458817
TOTAL DISSOLVED SOLIDS Total Dissolved Solids, Calculated	34500	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471961
TOTAL SODIUM, CALCULATED Total Sodium, Calculated	8230 <sup>°°</sup>	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471978

Qualifiers:

ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)

D - Surragate Recovery Unreportable due to Dilution

MI - Matrix Interference



Client Sample ID 2822-D		Coll	ected:	10/28/00	SPL Sample ID:	0010	0895-04			
	Site: HI 135/136 WM0070									
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #			
TOTAL SUSPENDED SOLIDS	· • • • •	· _ ·	MCL	E160.2	Units: mg/					
Suspended Solids (Residue Non- Filterable)	786	8		2	11/02/00 15:00 E		461973			

Qualifiers:

ND/U - Not Detected at the Reporting Limit

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B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/15/00 11:20:42 AM



Client Sample ID 282	2-E	Co	llected:	10/28/00	SPL Sample ID: 0010	0895-05
		Site	e: HI 1	35/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARB	ONATE		MCL	M2320 B	Units: mg/L	
Alkalinity, Bicarbonate	111	2		1	11/01/00 14:00 SN	461507
ALKALINITY, CARBO Alkalinity, Carbonate	NATE 20.2	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460239
CHLORIDE, TOTAL			MCL	E325.3	Units: mg/L	<b></b>
Chloride	17400	250		250	11/08/00 11:00 CV	468306
· · · · ·	:			·		
METALS BY METHOD			MCL	SW6010B	Units: mg/L	474074
Banum	0.0137	0.005		<b>1</b>	11/10/00 20:51 E_B	471271
Calcium	402	0.1		1	11/10/00 20:51 E_B	471271
Iron	0.554	0.02		1	11/10/00 20:51 E_B	471271
Magnesium	1290	1		10	11/13/00 15:25 JM	471758
Polassium	373	20		10	11/13/00 15:25 JM	471758
	L_001110C-471271					
Prep Method	Prep Date	Prep Initials	Ì,			
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA	001113B-471758		1			
Prep Method	Prep Date	Prep Initials	i			
SW3010A	11/01/2000 8:30	MR	]			
PH			MCL	E150.1	Units: pH Units	.:
pH	8.4	0.10		1	11/01/00 16:00 EC	458729
	al de la companya de					::::
RESISTANCE @ 25 C		, <b></b>	MCL	120.1	Units: Mohms/cm	
Resistance	0.0083	0.00100	MCL	<b>120.1</b>	Units: Mohms/cm 11/03/00 9:15 C_V	461390
Resistance	0.0083	0.00100		1	11/03/00 9:15 C_V	461390
Resistance SPECIFIC GRAVITY			MCL MCL	120.1 1 ASTM D-1429	11/03/00 9:15 C_V Units: Specific Gra	461390 avity@
Resistance	0.0083	0.00100	MCL	1	11/03/00 9:15 C_V	461390
Resistance SPECIFIC GRAVITY				1	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L	461390 avity@ 462224
Resistance SPECIFIC GRAVITY Specific Gravity			MCL	1 AST <u>M D-142</u> 9 1	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V	461390 avity@ 462224 458818
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate	1.029	0	MCL MCL	1 ASTM D-1429 1 E375.4 500	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN	461390 avity@ 462224
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate TOTAL DISSOLVED S	1.029 5900	0 500	MCL	1 ASTM D-1429 1 E375.4	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN Units: mg/L	461390 avity @ 462224 458818
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate	1.029 5900	0	MCL MCL	1 ASTM D-1429 1 E375.4 500	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN	461390 avity@ 462224 458818
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate TOTAL DISSOLVED S	1.029 5900 OLIDS Calculated 36500	0 500	MCL MCL	1 ASTM D-1429 1 E375.4 500	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN Units: mg/L 11/13/00 18:00 ES Units: mg/L	461390 462224 458818 471962
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate TOTAL DISSOLVED S Total Dissolved Solids, (	1.029 5900 OLIDS Calculated 36500 CULATED	0 500	MCL MCL MCL	1 ASTM D-1429 1 E375.4 500 TDS-MINERAL 1	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN Units: mg/L 11/13/00 18:00 ES	461390 Avity @ 462224 458818 471962
Resistance SPECIFIC GRAVITY Specific Gravity SULFATE, TOTAL Sulfate TOTAL DISSOLVED S Total Dissolved Solids ( TOTAL SODIUM, CAL	1.029 5900 OLIDS Calculated 36500 CULATED Id 11000	0 500 10	MCL MCL MCL	1 ASTM D-1429 1 E375.4 500 TDS-MINERAL 1	11/03/00 9:15 C_V Units: Specific Gra 11/06/00 11:00 C_V Units: mg/L 11/01/00 10:00 SN Units: mg/L 11/13/00 18:00 ES Units: mg/L	461390 462224 458818 471962

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyle detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and POL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unrepartable due to Dilution MI - Matrix Interference

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11/15/00 11:20:43 AM



Client Sample ID 2822	2-F		Colle	ected:	10/28/00	SPL Sample ID: 001	00895-06
			Site:	HI 1	35/136 WM0070		
Analyses/Nethod	Ře:	suit	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analys	t Seq. #
ALKALINITY, BICARB Alkalinity, Bicarbonate		125	2	MCL	M2320 B 1	Units: mg/L. 11/01/00 14:00 SN	461509
ALKALINITY, CARBON Alkalinity, Carbonate		1.04	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460240
CHLORIDE, TOTAL Chloride	16	500	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468307
METALS BY METHOD	6010B, TOTAL		· · · · ·	MCL	SW6010B	Units: mg/L	• •
Barium	0.0	128	0.005		1	11/10/00 19:49 E_B	471259
Calcium		388	0.1		1	11/10/00 19:49 E_B	471259
tron	Q.	409	0.02		1	11/10/00 19:49 E_B	471259
Magnesium	1	350	1		10	11/13/00 14:24 JM	471743
Potassium		407	20		10	11/13/00 14:24 JM	471743
Run ID/Seq #: TJA	001110C-471259				•	sta o fini e i miname esperi	·····
Prep Method	Prep Date		Prep Initials				
SW3010A	11/01/2000 8:30		IMR				
Run ID/Seq #: TJA	_001113B-471743						
Prep Method	Prep Date		Prep Initials				
SW3010A	11/01/2000 8:30		MR				
РН pH		8.3	0.10	MCL	<b>E150.1</b>	Units: pH Units 11/01/00 16:00 EC	458730
RESISTANCE @ 25 C Resistance		ND	0.00100	MCL	<b>120</b> .1 1	Units: Mohms/cr 11/03/00 9:15 C_V	n 461392
SPECIFIC GRAVITY Specific Gravity	· · 1.	033	0	MCL	ASTM D-1429	Units: Specific G 11/06/00 11:00 C_V	i <b>ravity@</b> 462225
SULFATE, TOTAL Sulfate	5	300	500	MCL	<b>E375.4</b> 500	Units: mg/L 11/01/00 10:00 SN	458819
TOTAL DISSOLVED S				MCL	TDS-MINERAL	Units: mg/L	
Total Dissolved Solids. C		100	10		1	11/13/00 18:00 ES	471963
TOTAL SODIUM, CALC Total Sodium, Calculated		500	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471980
TOTAL SUSPENDED S Suspended Solids (Resid Filterable)		25	4	MCL	E160.2	Units: mg/L 11/02/00 15:00 EC	461975

Qualifiers:

ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution Mi - Matrix Interference



Client Sample ID 2822-G		Co	ected:	10/28/00	SPL Sample ID: 001	00895-07
		Site	e, Hit	135/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARBONATE	· · · · ·		MCL	M2320 B	Units: mg/L	
Alkalinity, Bicarbonale	119	2		1	11/01/00 14:00 SN	461513
ALKALINITY, CARBONATE		• : *	MCL	M2320 B	Units: mg/L	
Alkalinity, Carbonate	8.08	2		1	11/01/00 14:00 SN	460241
CHLORIDE, TOTAL		••	MCL	E325.3	Units: mg/L	
Chloride	15700	250	meu	250	11/08/00 11:00 CV	468308
METALS BY METHOD 6010B, TO	TAL		MCL	SW6010B	Units: mg/L	
Barium	0.0154	0.005		1	11/10/00 20:56 E_B	471272
Calcium	392	0.1		1	11/10/00 20:56 E_B	471272
Iron	1.29	0.02		1	11/10/00 20:56 E_B	471272
Magnesium	1330	1		10	11/13/00 15:29 JM	471759
Potassium	423	20		10	11/13/00 15:29 JM	471759
Prep Method         Prep Date           SW3010A         11/01/2000 8	3:30	Prep Initials				
PH			MCL	E150.1	Units: pH Units	. ,
ρH	8.4	0.10		1	11/01/00 16:00 EC	458731
RESISTANCE @ 25 C Resistance	0.043	0.00100	MCL	1 <b>20.1</b>	Units: Mohms/cm 11/03/00 9:15 C_V	n 461393
SPECIFIC GRAVITY	• .	. •	MCL	ASTM D-1429	Units: Specific G	ravitu Ø
Specific Gravity	1.029	0		1	11/06/00 11:00 C_V	462226
SULFATE, TOTAL			MCL	E375.4	Units: mg/L	
Sulfate	7100	1000		1000	11/01/00 10:00 SN	458820
TOTAL DISSOLVED SOLIDS Total Dissolved Solids, Calculated	35500	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471964
TOTAL SODIUM, CALCULATED Total Sodium, Calculated	10400	10	MCĻ	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471981
TOTAL SUSPENDED SOLIDS Suspended Solids (Residue,Non- Filterable)	34	.· :· · 4	MCL	<b>E160.2</b>	Units: mg/L 11/02/00 15:00 EC	461977

Filterable)

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
D - Surrogate Recovery Unreportable due to Dilution

MI - Matrix Interference

11/20/00 10:55:34 AM



Client Sample ID 2822	-H	Coll	ected:	10/28/00	SPL Sample ID: 001	00895-08
		Site	: <u>HI</u> 1	35/136 WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analys	•
ALKALINITY, BICARBO Alkalinity, Bicarbonate	<b>DNATE</b> 129	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	46151
ALKALINITY, CARBON Alkalinity, Carbonate	8.08	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	46024
CHLORIDE, TOTAL Chloride	19100	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	46831
METALS BY METHOD	6010B, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	0.0128	0.005		1	11/10/00 21:00 E_B	47127
Calcium	389	0.1		1	11/10/00 21:00 E_B	47127
Iron	0.628	0.02		1	11/10/00 21:00 E_B	47127
Magnesium	1310	1		10	11/13/00 15:34 JM	47176
Potassium	397	20		10	11/13/00 15:34 JM	47176
Run ID/Seg #: TJA	_001110G-471273				•••••	
Prep Method	Prep Date	Prep Initials				
	11/01/2000 8:30	MR				
Run ID/Seq #: TJA		<b>n</b> 1 1				
	Prep Date	Prep Initials			:	
SW3010A	11/01/2000 8:30	MR			, ;	
PH			MCL	E150.1	Units: pH Units	
<b>PH</b>	8.4	0.10		1	11/01/00 16:00 EC	458732
RESISTANCE @ 25 C Resistance	0.02	0.00100	MĊĽ.	<b>120.1</b>	Units: Mohms/cr 11/03/00 9:15 C_V	n 46139!
SPECIFIC GRAVITY Specific Gravity	1.025	0	MCL	ASTM D-1429	Units: Specific G 11/06/00 11:00 C_V	ravity @ 462227
SULFATE, TOTAL Sulfate	6500	500	MCL	<b>E375.4</b> 500	Units: mg/L 11/01/00 10:00 SN	458822
TOTAL DISSOLVED SC Total Dissolved Solids, C	· · · ·	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471965
TOTAL SODIUM, CALC Total Sodium, Calculated		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471983
TOTAL SUSPENDED S Suspended Solids (Resid Filterable)		4	MCL	<b>E160.2</b>	Units: mg/L 11/02/00 15:00 EC	461978

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/15/00 11:20:45 AM



Client Sample ID 2822-1		Cóll	ected:	SPL Sample ID	: 0010	0892-01	
		Site	: HI 135/	136 #WM0070	)		
Analyses/Method	Result	Rep.Limit	DI	Factor QUAL	Date Analyzed	Analy5t	Seq. #
DISSOLVED OXYGEN	·· · · · · · · · · · · · · · · · · · ·	·······	MGL	E360.1	Units: mg	Л.	
Oxygen, Dissolved	7	2	······································	1 100000	10/31/00 17:45 (	<u>_v</u>	457695
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	/L	
Nitrogen, Kjeldahl, Total	1.6	0.3		1		S	465077

Qualifiers:

ND/U - Not Detected at the Reporting Limit

 $\boldsymbol{\beta}$  - Analyte detected in the associated Method Blank

.......

\* - Surrogate Recovery Outside Advisable QC Limits J - Estimated Value between MDL and PQL >MCL - Result Over Maximum Contamination Limit(MCL) () • Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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11/9/00 11:02:22 AM

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Client Sample ID 2822-2		Coll	ected:	4.4 M (10 - 10	SPL Sample ID	: 00100	892-02
		Site	: HI 13	5/136 #WM0070	)		
Anaiyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg	/L	
Oxygen,Dissolved	5.5	2		1	10/31/00 17:45 0	<u>v_</u>	457698
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	/L	
Nitrogen, Kjeldahl, Total	0.85	0.3		1	11/06/00 11:30	JS	465078

Qualifiers:

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

. . .



Client Sample ID 2822-3	Collected:			SPL Sample ID	: 00100	00100892-03	
	_	Site:	HI 13	5/136 #WM007	Ó		
Analyses/Method	Result	Rop.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN		· · · · · · · · · · · · · · · · · · ·	MCL	E360.1	Units: mg	۲L	
Oxygen.Dissolved	4.9	2		1	10/31/00 17:45	¢_V	457700
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	/L	
Nitrogen, Kjeldahl, Total	1.1	0.3		1	11/06/00 11:30	JS	465079

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

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- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

a in the second 
11/9/00 11:02:23 AM

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Client Sample ID 2822-4		Collected:			SPL Sample ID	: 0010	0892-04
		Site	: HI 13	5/136 #WM0070	)		
Analyses/Method	Result	Rep.Limit	ť	II. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN		,	MCL	E360.1	Units: mg		
Oxygen, Dissolved	5.1	2		1	10/31/00 17:45	<u>_v</u>	457701
NITROGEN, KJELDAHL, TOTAL		•	MCL	E351.3	Units: mg	۷L	
Nitrogen, Kjeldahl, Total	ND	0.3		1		JS	465080

Qualifiers:

. . . . . . . . . .

ND/U - Not Detected at the Reporting Limit

6 - Analyte detected in the associated Method Diank

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\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference



Client Sample ID 2822-5		Colle	SPL Sample ID	: 00100892-05		
		Site:	HI 1	35/136 #WM0070		
Analyses/Method	Result	Rep.Limit		DII. Factor QUAL	Date Analyzed	Analyst Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg	/L
Oxygen, Dissolved	9.4	2		1	10/31/00 17:45	2_V 457702
NITROGEN, KJELDAHL, TOTAL	· · · ·		MCL	E351.3	Units: mg	ί.
Nitrogen,Kjeldahl,Total	1.1	0.3	• • •	1	11/06/00 11:30	JS 465082

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank \* - Surrogate Recovery Outside Advisable QC Limits

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J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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Client Sample ID 2822-6		Colle	cted:		SPL Sample ID	: 0010089	2-06
		Site:	HI 1	35/136 #WM007	מ		
Analyses/Method	Result	Rep.Llmit		Dil. Factor QUAL	Date Analyzed	Analyst S	ieq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg	<u>ب</u>	
Oxygen, Dissolved	10	2	• • •	1	10/31/00 17:45		57703
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	<u>/L</u>	
Nitrogen,Kjeldahl,Total	ND	0.3		1	11/06/00 11:30		65083

Qualifiers:

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Blank \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

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>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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Client Sample ID 2822-7	Collected:			SPL Sample ID	: 0010	00100892-07	
		Site:	HI 13	5/136 #WM0070			
Analyses/Method	Result	Rep.Limit		DII. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN			MCL E360.1		Units: mg/L		
Oxygen, Dissolved	10	2		1	10/31/00 17:45	<u>v_v</u>	457704
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	/L	· · · · · · ·
Nitrogen, Kjeldahl, Total	2.8	0.3	• •	1	11/06/00 11:30	JS	465084

Qualifiers:

- ND/U Not Detected at the Reporting Limit
- B Analyte detected in the associated Method Blank

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- \* Surrogate Recovery Outside Advisable QC Limits
- J Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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Client Sample ID 2822-8		Colle	SPL Sample ID	00100892-08		
		Site:	HI 13	5/136 #WM0070		
Analyses/Method	Result	Rep.Limit	۵	II. Factor QUAL	Date Analyzed	Analyst Seq. #
DISSOLVED OXYGEN		· ·	MCL	E360.1	Units: mg	
Oxygen.Dissolved	10	2		1	10/31/00 17:45 (	C_V 45770
NITROGEN, KJELDAHL, TOTAL	••		MCL	E351.3	Units: mg	/L
Nitrogen,Kjeldahl,Total	ND	0.3			11/06/00 11:30	JS 46508

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

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- Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreputable due to Dilution Mi - Matrix Interference



#### Winmar Consulting Services

Certificate of Analysis Number:

#### 00100892

Report To: Fax To:	Winmar Consulting James Wiseman 5700 NW Central Dri Suite 150 Houston TX 77092- ph: (713) 895-8240 Winmar Consulting So James Wiseman	ivə fax: (713) 89		Project Name: Site: Site Address: PO Number: State: State: State Cert. No.: Date Reported:	HI 135/136 #WM0D70 HI 135/136 #WM0D70 11/9/00		
CII	ent Sample ID	Lab Sample ID	Matrix	Date Collected	Date Received	COC ID	HOLD
2822-1		00100892-01	Water		10/30/00 12:04:00 PM		•
2822-1		00100892-01	Water	:	10/30/00 12:04:00 PM	087140	.
2822-2		00100892-02	Water		10/30/00 12:04:00 PM		
822-2		00100892-02	Water	1	10/30/00 12:04:00 PM	087140	li
822-3		00100892-03	Water	[	10/30/00 12:04:00 PM		
2822-3		00100892-03	Water		10/30/00 12:04:00 PM	087140	
<u>e22-4</u>		00100892-04	Water	:	10/30/00 12:04:00 PM		
!-4		00100892-04	Water		10/30/00 12:04:00 PM	087140	
co22-5		00100892-05	Water		10/30/00 12:04:00 PM		
2822-5		00100892-05	Water		10/30/00 12:04:00 PM	087140	
822-6		00100892-06	Water		10/30/00 12:04:00 PM	į	
2822-6		00100892-06	Water		10/30/00 12:04:00 PM	087140	
2822-7		00100892-07	Water		10/30/00 12:04:00 PM		
2822-7		00100892-07	Water		10/30/00 12:04:00 PM	087140	[ ]
822-8		00100892-08	Water		10/30/00 12:04:00 PM		
2822-8		00100892-08	Water	•	10/30/00 12:04:00 PM	087140	

Neschich,Paul Senior Project Manager 11/9/00

Date

Joe! Grice Laboratory Director

Ted Yen Quality Assurance Officer



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77064 (713) 660-0901

Client Sample ID 2822-1		Colle	cted:		SPL Sample ID:	00100892-01
		Site:	HI	135/136 #WM0070		
Analyses/Method	Result	Rep.Limit	<u></u> , <u>_</u>	Dil. Factor QUAL	Date Analyzed An	alyst Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg/L	
Oxygen, Dissolved	7	2		1	10/31/00 17:45 C_V	457695
NITROGEN, KJELDAHL. TOTAL			MCL	E351.3	Units: mg/L	
Nitrogen,Kjeidahl,Total	1.6	0.3		1	11/06/00 11:30 JS	465077

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Mothod Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference



HOUSTON LABORATORY 8880 INTERCHANGE ORIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2822-2		Colle	cted:		SPL Sample ID:	0010	0892-02
		Site:	HI	35/136 #WM0070			
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg/i	L	
Oxygen,Dissolved	5.5	2		1	10/31/00 17:45 C	_v	457698
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg/	L	
Nitrogen, Kjeldahl, Total	0.85	0.3		1	11/06/00 11:30 J	IS	465078

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the according Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surroyate Recovery Unreportable due to Dilution MI - Matrix Interference

11/9/00 4:53:10 PM



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 {713) 660-0901

Client Sample ID 2822-3		Colle	SPL Sample ID:	0010	0892-03		
		Site:	HI 1	135/136 #WM0070			
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg/	/L	
Oxygen, Dissolved	4.9	2		1	10/31/00 17:45 C	:_v	457700
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg/	۲L	
Nitrogen.Kjeldahl, Total	1.1	0.3		1	11/06/00 11:30	JS	465079

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference



HOUSTON LABORATORY 8450 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

ليستوار وسفاد وترأوا التسارات والدامدين الجام

Client Sample ID 2822-4		Collected:			SPL Sample ID:	00100892-04
		Site:	HI	135/136 #WM0070	)	
Analyses/Nethod	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed A	nalyst Seq.#
DISSOLVED OXYGEN		·	MCL	E360.1	Units: mg/L	
Oxygen Dissolved	5.1	2		1	10/31/00 17:45 C_V	√ 457 <b>7</b> 0
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg/L	
Nitrogen, Kjeldahl, Total	ND	0.3		1	11/06/00 11:30 JS	46508

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 560-0901

Client Sample ID 2822-5		Colle	cted:	SPL Sample ID: 0	0100892-05	
		Site:	HI	35/136 #WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Anal	yst Seq.#
DISSOLVED OXYGEN	• •		MCL	<b>E360</b> .1	Units: mg/L	
Oxygen.Dissolved	9.4	2		1	10/31/00 17:45 C_V	457702
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg/L	
Nitrogen, Kjeldahl, Total	1.1	0.3		1	11/06/00 11:30 JS	465082

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2822-6		Colle		SPL Sample ID	: 00100	892-06	
		Site:	HI	135/136 #WM0070	)		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN			MCL	E360.1	Units: mg	/∟	
Oxygen, Dissolved	10	2		1	10/31/00 17:45 0	C_V	457703
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg	/L	
Nitrogen, Kjeldahl, Total	ND	0.3		1	11/06/00 11:30	JS	465083

Qualifiers:

ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/9/00 4:53:11 PM



HOUSTON LABORATORY 8580 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2822-7		Colle	cted:	SPL Sample ID:	00100	0892-07	
		Site:	HI 1	135/136 #WM0070	)		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
DISSOLVED OXYGEN	·		MCL	E360.1	Units: mg/	L	
Oxygen, Dissolved	10	2		1	10/31/00 17:45 C	_V	<b>45</b> 7704
NITROGEN, KJELDAHL, TOTAL			MCL	E351.3	Units: mg/	L	
Nitrogen, Kjeldahl, Total	2.8	0.3		1	11/06/00 11:30 J	IŞ	465084

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable OC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 Mi - Matrix Interference



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0907

Client Sample ID 2822-8		Colle	cted:		SPL Sample ID: 0	0100892-08
		Site:	HI	135/136 #WM0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Anal	yst Seq. #
DISSOLVED OXYGEN	••	•	MCL	<b>E360</b> , 1	Units: mg/L	
Oxygen Dissolved	10	2		1	10/31/00 17:45 C_V	457705
NITROGEN, KJELDAHL, TOTAL	,		MCL	E351.3	Units: mg/L	
Nitrogen, Kjeldahl, Total	ND	0.3	•	1	11/06/00 11:30 JS	465085

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Molhod Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 P - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

11/9/00 4:53:12 PM



# 7. Results and Observations – 2823

## a. Sample Locations Observations

Time constraints interfered with this test, and only one water sample was obtained. The sample was of standing water in this pipeline segment. The pipeline was flushed and filled with seawater in 1994. The testing information is still included, as it is of some value, however it is not as complete as for the other segments tested.

A five foot (5') sample of the pipeline was removed, which included the tubeturn to pipeline weld.

#### b. Gas Composition Observations

This line was completely filled with inhibited seawater, therefore no gas was present.

#### c. Flushwater Composition Observations

For the water that stood in the pipeline for 6 years, the mineral pattern relative to NSW is summarized below:

Alkalinity (bicarb) – Higher Barium – Equal Calcium – Lower Iron – Higher Magnesium – Lower Potassium – Similar/Lower Chloride – Lower Sulfate – Higher

As with pipeline 2822, iron concentrations in 2823 were very high. This sample from the standing water yielded an iron concentration of 117 PPM, or 34,000 times NSW levels.

#### d. Oil and Grease Observations

Due to time constraints, no oil and grease samples were taken for this line.

#### e. Pipe Cutout Observations



This pipeline sample was in better condition than 2822. It showed light surface corrosion only, and no deep pitting at the weld. This sample did not have the thick coating of debris on the interior as seen in sample 2822.



Photo #20 - Sample Photo - Segment 2823



#### SHELL OFFSHORE INC. HI-135-2

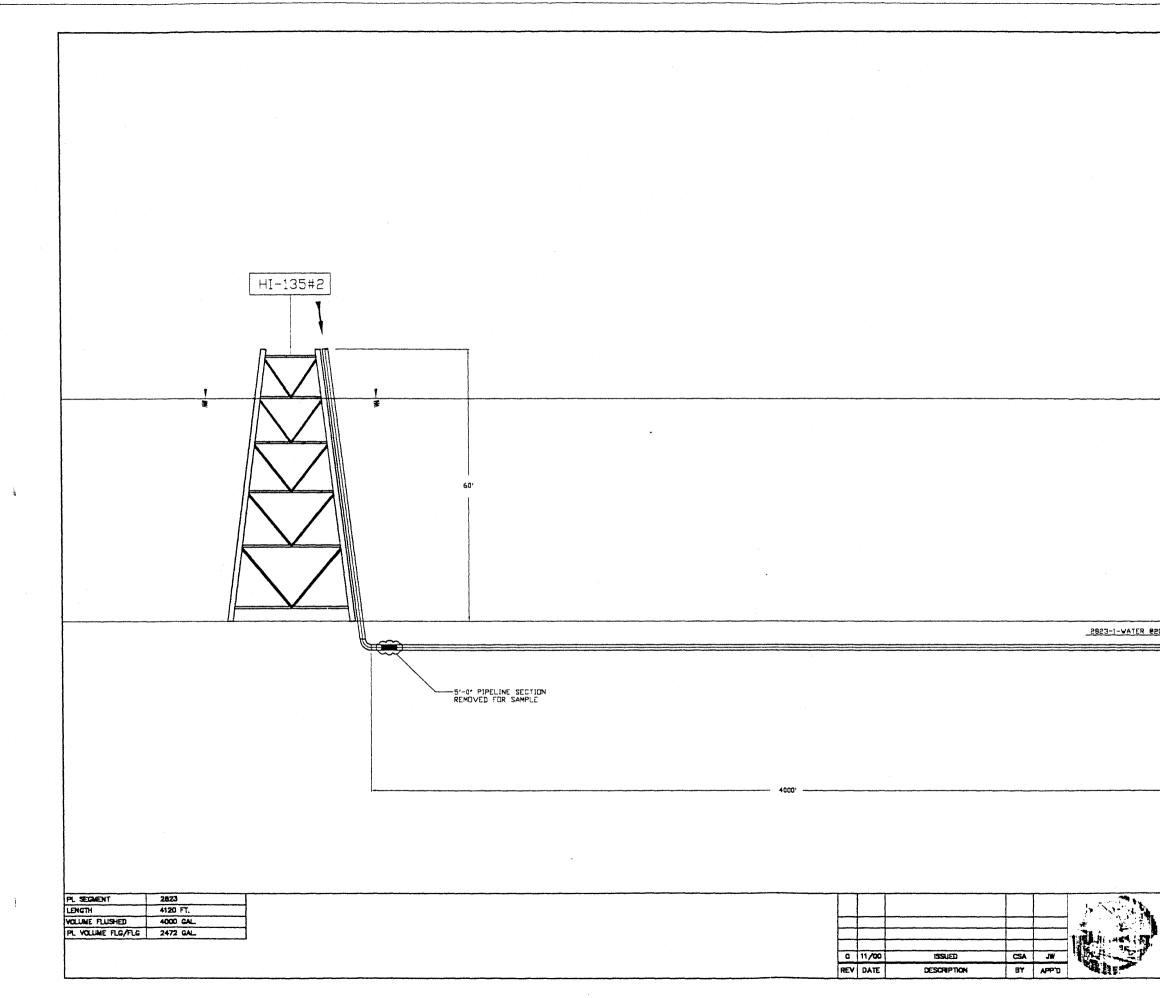
			Р	LATFOR	M				
MMS Gene	eral	0	DS Genera	al	MMS Lo	cation	MMS Fa	cility	
Water	50 feet	Function	WP		Lease	741 feet	Helideck	Yes	
Major	No	Piles	NA		Complex	10015	Quarters	None	
Decks	1	Slots	3		Longitude	-94.112	Generator	No	
Slots	3	ODS ID	739		Latitude	29.260	Cranes	NA	
Wells	2	Previous 1	NA		Х	3,558,137'	Gas	Yes	
Flare	No	Previous 2	NA		Y	551,391'	Oil	Yes	
Installed	01 1964	Previous 3	NA		To Shore	25 miles	Comp	No	
Revised	05 1998	Previous 4	NA		N-S feet	S 4911'	8 hour	No	
Removed	NA	Notes	NA		E-W feet	W 3141'	24 hour	No	
			PIF	PELINES	MMS				
Segment	2821		2822		2823				
Origin	HI-135-2		HI-135-2		HI-135-2		#N/A		
Terminus	HI-136-A		HI-136-A		HI-136-A		#N/A		
O.D.	3"		4"		4"		#N/A		
Length	4,000'		4,000'		4,000'		#N/A		
Product	BLKG		BLKG		BLKG		#N/A		
Status	ACT		OUT		PABN		#N/A		
Installed	NA		NA		NA		#N/A		
Abandon	NA		NA		NA		#N/A		
Revised	Aug-94		Oct-94		Aug-94		#N/A		
Operator	SHELL O	FFSHORE INC	SHELL C	FFSHORE	SHELL OFF	SHORE INC.	#N/A		
			V	VELLS N	IMS				
API Well ID	Well	Spud	Revised	Status	MD	Bot Lease	Sur Long	Sur Lat	
427080004600	1	03 1964	05 1964	COM	10,777'	741	-94.119	29.259	
427080004700	2	10 1964	05 1986	ST	8,678'	741	-94.112	29.260	
427080004701	2	05 1986	07 1986	COM	9,450'	741	-94.115	29.260	
427080004800	3	11 1964	08 1985	PA	9,663'	741	-94.104	29.265	
427080004900	4	11 1964	06 1990	PA	11,483'	741	-94.116	29.269	
427080005000	5	04 1965	08 1985 PA		9,006'	741	-94.089	29.262	
427080006100	6	06 1965	06 1965	PA	9,085'	741	-94.089	29.271	
427080007300	8	07 1967	08 1967	ST	9,452'	741	-94.103	29.260	
427080007301	9	08 1967	08 1967	COM	9,020'	741	-94.106	29.260	
427084023300	11	04 1986	01 1996	ST	10,824'	741	-94.075	29.262	
427084023301	11	01 1996	04 1996	COM	11,033'	741	-94.075	29.262	
427084023400	9	08 1986	09 1986	TA	10,230'	742	-94.132	29.259	



#### PIPELINE FLUSHING AND SAMPLING RECORD

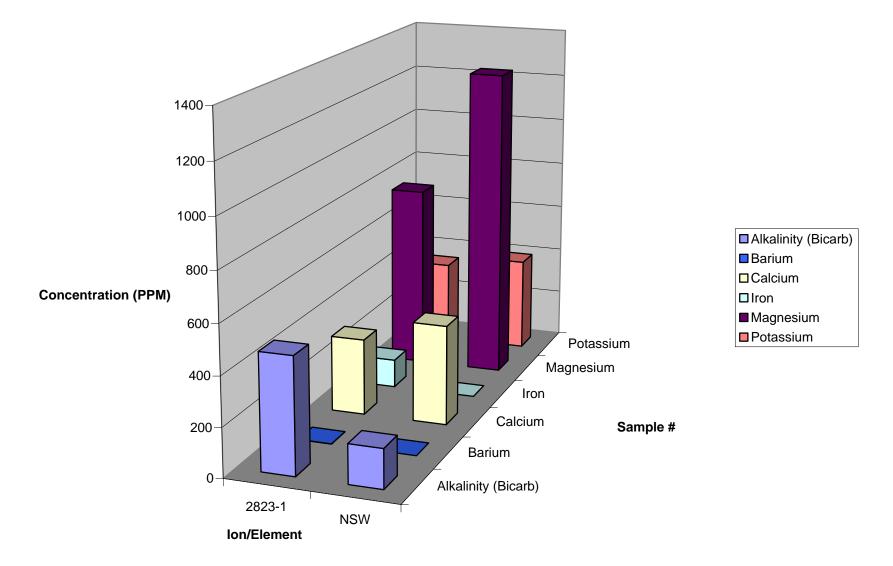
I. Pipeline Information	0000
MMS Segment No.	2823
Date:	10/27/2000
Pipeline Origination	
Area	High Island
Block	135
Platform	#2
Lease	OCS-G-0741
Pipeline Destination	
Area	High Island
Block	136
Platform	A
Lease	OCS-G-0742
Pipeline Size (in)	4
Pipelines Length (ft)	4,000
Pipeline Volume (bbls)	62
U. Ehrshimmlafarren itar	
II. Flushing Information	
Flushing Information	4000-
Volume Flushed	4000g
Flow Rate (GPM)	100
Pigged Used	No
Type of Pig	No
Size of Pig	No
Clean Returns	Yes
Inhibitor	
Chemical Inhibitor Used	
Type of Chemical	
Quantity of Chemical	
Origination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Destination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Comments:	
Company Representative	
Signature	

III. Sampling Data - Tracking	g Information			
Sample Location Platform: Pipeline Sampling Site: Flushing Start Time:	HI 136A End of Hose a 18:00	at Tank		
Gas Samples Vacuum Tubes	Sample ID N/A	Sample Date	Vol. Flushed (g)	H2S (PPM)
Plastic Bags	N/A			
Water Samples Mineral Pattern Analysis	Sample ID	Sample Date	Vol. Flushed (g)	Notes
Oil and Grease Analysis	2823-1	10/27/2000	2,50	
Comments: Company Representative			constraints and harsh pool sample taken as	
Signature				



HI	136A	
	PROJECT# WM007 PIPELINE SAMPLE LOCAT HIGH ISLAND 135#2 / HIGH IS CSA DATE 11/00 WM	TONS





# Figure 20 - Chloride and Sulphate - 2823

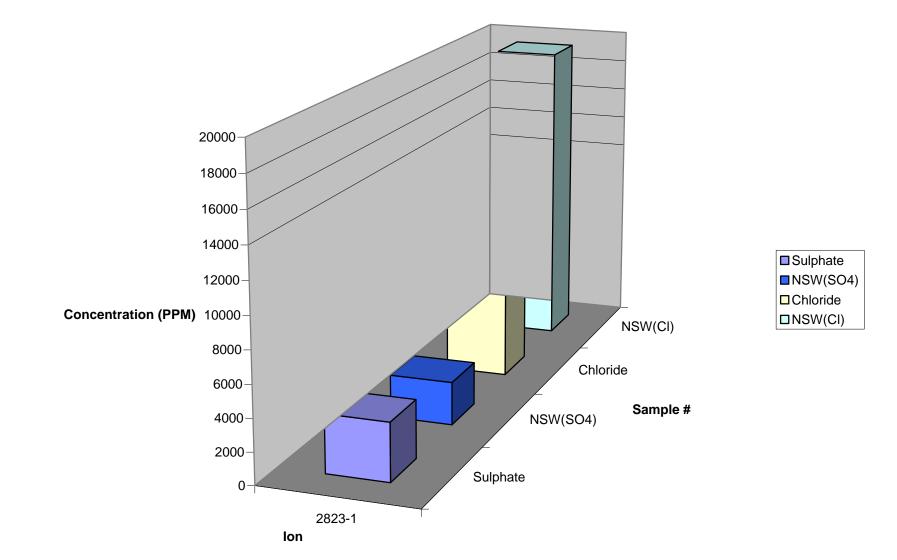
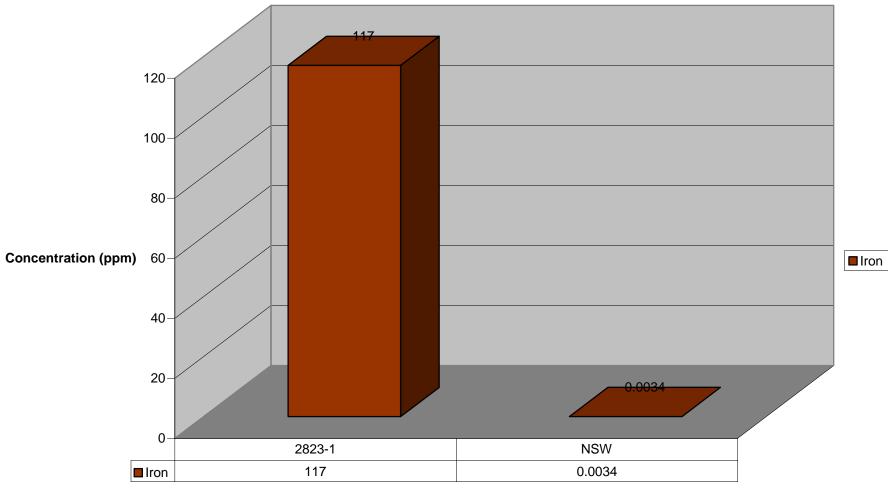


Figure 21 - Iron Concentration - 2823



Sample #



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 282	Colle	acted:		SPL Sample ID:	00100896-01	
		Site:	HI 1	135/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Ana	lyst Seq. #
ALKALINITY, BICARB Alkalinity, Bicarbonate	ONATE 473	2	MCL	<b>M2320 B</b> 1	Units: mg/L 11/01/00 14:00 SN	461517
ALKALINITY, CARBO Alkalinity, Carbonate	NATE 295	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460244
CHLORIDE, TOTAL Chioride	15300	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468311
METALS BY METHOD	6010B, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	1.14	0.005		1	11/10/00 21:04 E_B	471274
Calcium	315	D.1		1	11/10/00 21:04 E_B	471274
Iron	117	Û.Û2		1	11/10/00 21:04 E_B	471274
Magnesium	763	0.2		2	11/13/00 15:42 JN	471762
Potassium	343	2		1	11/13/00 15:38 JM	471761
Run iD/Seq #: TJ/	_001110C-471274					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA	_001113B-471761					
Prep Method	Prep Date	Preo Initials				
SW3010A	11/01/2000 8:30	MR				
	_001113B-471762					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
PH	9,1	0.10	MCL	<b>E150</b> .1	Units: pH Unit 11/01/00 16:00 EC	<b>ts</b> 458734
RESISTANCE @ 25 C Resistance	0.0093	0.00100	MCL	<b>120.1</b>	Units: Mohms 11/03/00 9:15 C_V	461397
SPECIFIC GRAVITY Specific Gravity	1.02	0	MCL	ASTM D-1429 1	Units: Specifi 11/06/00 11:00 C_V	c Gravity@ 462228
SULFATE, TOTAL Sulfate	3620	500	MCL	<b>E375.4</b> 500	Units: mg/L 11/01/00 10:00 SN	458823
TOTAL DISSOLVED S Total Dissolved Solids,	and the second sec	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471966
TOTAL SODIUM, CAL Total Sodium, Calculate		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471983

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/14/00 10:58:55 AM

Filterable)



. ....

HOUSTON LABORATORY B880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2823-1		Collected:			SPL Sample ID: 0010		0896-01
•		Site:	HI 135/136				
Analyses/Method	Result	Rep.Llmit	Dil. Factor	QUAL	Date Analyzed	Analyst	Seq. #
TOTAL SUSPENDED SOLIDS Suspended Solids (Residue, Non-	1020		MCL E	160.2	Units: mg		461985

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/14/00 10:58:59 AM



# 8. Results and Observations – 2824

### a. Sample/Locations Observations

Gas samples were taken at the top of the riser before the blind flange was removed. Samples were taken when the odor of natural gas was present. All bolts and flange seals were intact before testing and did not indicate any leakage. H2S length of stain tests were performed at this. This gas line was one of the longer ones tested, allowing for more samples of both gas and water.

The water samples taken at the top of riser bleed valve seemed uniform, and representative of the flush fluid stream.

A five foot (5') sample of the pipeline was removed, which included the tubeturn to pipeline weld.

### b. Gas Composition Observations

The results of the gas analysis are plotted and summarized in the results section. Atmospheric air composition is also plotted for reference/comparison purposes.

The three gas samples were high in methane, and low in atmospheric components, such as nitrogen and oxygen. This indicates that the line was probably not opened in the past, and contaminated with atmospheric air. This is useful to compare to other gas lines tested, where samples were a mixture of methane and atmospheric air. This gas did not contain any H2S and negligible amounts of CO2.

#### c. Flushwater Composition Observations

The flushwater composition for segment 2824 is plotted in the results section. Natural Seawater composition is also plotted for comparison purposes. The ions/elements plotted are: Alkalinity (CO3), Barium, Calcium, Iron, Magnesium, and Potassium. Because of their high values (in PPM), Chlorides and Sulfates are plotted on a separate chart.

For the flushwater, the mineral pattern relative to NSW is summarized below:

Alkalinity (bicarb) – Higher for first sample, then Equal Barium – Higher for first sample, then Equal Calcium – Higher for first sample, then Equal Iron – Higher Magnesium – Lower for first sample, then Equal



Potassium – Lower for first sample, then Equal Chloride – Higher for first sample, then Equal Sulfate – Lower for first sample, then Equal

Again, the iron content is plotted as a separate graph in order to focus on these values. The first sample taken at the very front of the flushwater "slug" has an extremely high iron concentration of 302 ppm (ppm also equals milligrams/liter). The concentration is over 88,000 times greater than NSW. Observations from the field could explain this very high concentration. In anticipation of the incoming fluid, the sampling valve was left open so that the very first fluid out of the pipeline was taken as the first water sample. This slug picked up quite a bit of debris, and was very high in condensate, as is evidenced in the photographs. This debris included metal particles which were picked up from the pipe wall. The sampling procedure "dissolved" these metal particles and recorded them as a concentration value. The following four samples were lower in concentration, but still much higher than NSW values.



Photo #21

The ions/elements to focus on from this analysis are those found in steel corrosion products: FeO2, FeS. The samples showed higher than NSW concentrations of iron, but Sulfate was at NSW levels.

The first sample (2824-1) is vastly different than all the other water samples taken. Concentrations of elements/ions were either much higher or much lower than NSW concentrations.

## d. Oil and Grease Observations

Oil and grease was non-detectible in the final samples taken. The detection limit is 2.5 PPM. As noted above, the very first sample was high in hydrocarbons because it contained a good deal of the condensate that was present in the line. The graph shows



a very rapid drop in oil and grease concentration, with the non-detectable limit appearing to be reached at 1.25x flush volume.

### e. Pipe Cutout Observations

A five foot section of pipe was retrieved. This section was taken near the base of the platform, and included pipe on both sides of the riser/tubeturn weld.



Photo #22 - Sample Photo - 2824

This pipe shows evidence of standing water/fluid present at the 5-7 o'clock position. There is some metal loss corrosion in this region, as indicated in the sample photographs. These patches were small, and no deeper than 0.1t however.





Photos #23 and #24 - Sample Photos - 2824



Photo #24 - The tubeturn/pipeline weld appeared to be in good condition.



# SHELL OFFSHORE INC. HI-135-5

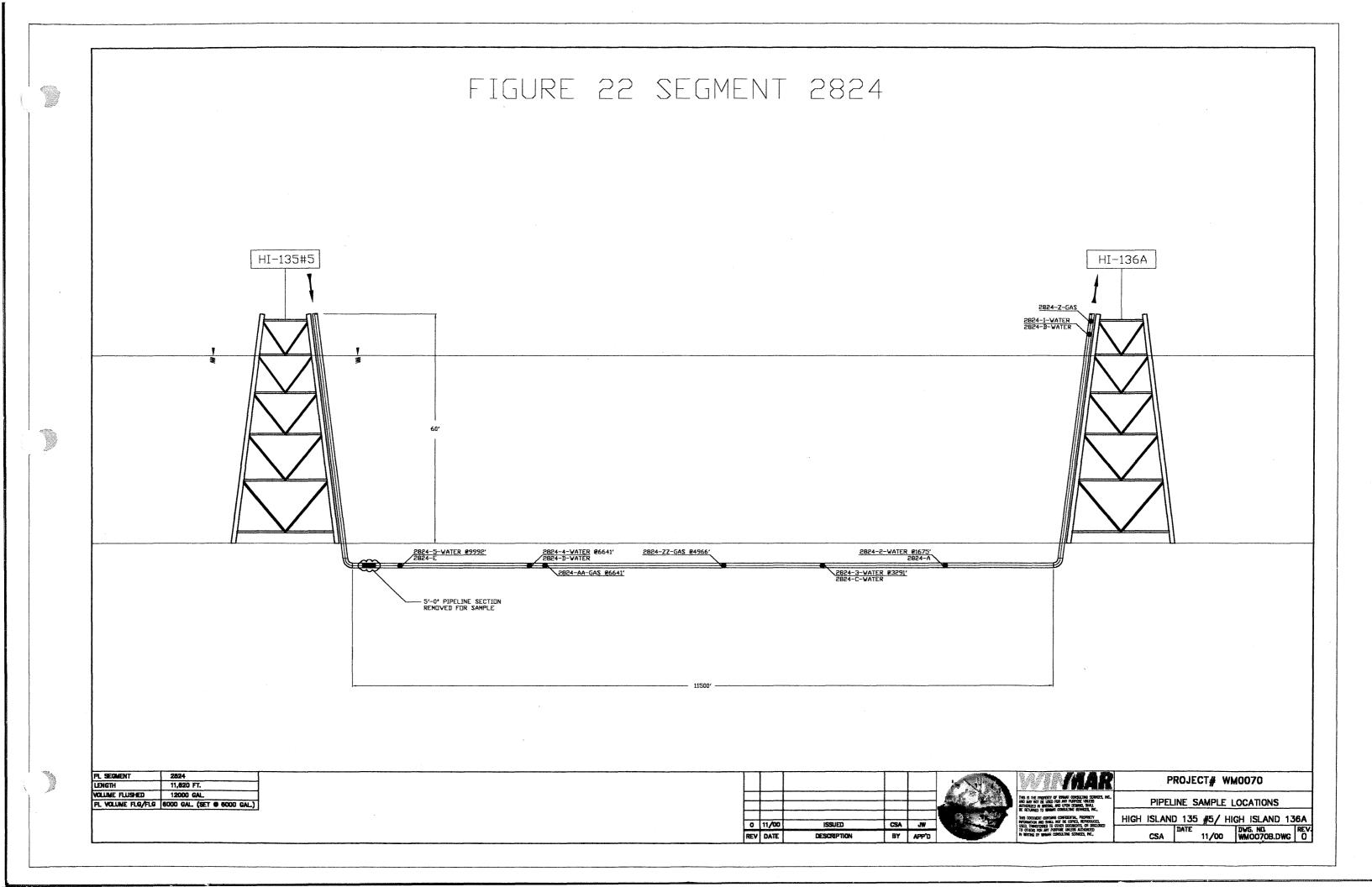
			Р	LATFOF	RM			
MMS General ODS Ge		DS Gener	al	MMS Lo	MMS Location		MMS Facility	
Water	50 feet	Function	WP		Lease	741 feet	Helideck	Yes
Major	No	Piles	NA		Complex	10014	Quarters	None
Decks	1	Slots	3		Longitude	-94.090	Generator	No
Slots	3	ODS ID	742		Latitude	29.262	Cranes	NA
Wells	2	Previous 1	NA		X	3,564,986'	Gas	Yes
Flare	No	Previous 2	NA		Y	552,334'	Oil	No
Installed	01 1965	Previous 3	NA		To Shore	25 miles	Comp	No
Revised	12 1998	Previous 4	NA		N-S feet	S 5854'	8 hour	No
Removed	NA	Notes	NA		E-W feet	E 5850'	24 hour	No
			PIF	PELINES	6 MMS			
Segment	2824							
Origin	HI-135-5							
Terminus	HI-136-A							
O.D.	4"							
Length	11,500'							
Product	BLKG							
Status	ACT							
Installed	NA							
Abandon	NA							
Revised	Aug-94							
Operator	SHELL O	FFSHORE INC	•					
			۷	VELLS N	MMS			
API Well ID	Well	Spud	Revised	Status	MD	Bot Lease	Sur Long	Sur Lat
427080004600	1	03 1964	05 1964	COM	10,777'	741	-94.119	29.259
427080004700	2	10 1964	05 1986	ST	8,678'	741	-94.112	29.260
427080004701	2	05 1986	07 1986	COM	9,450'	741	-94.115	29.260
427080004800	3	11 1964	08 1985	PA	9,663'	741	-94.104	29.265
427080004900	4	11 1964	06 1990	PA	11,483'	741	-94.116	29.269
427080005000	5	04 1965	08 1985	PA	9,006'	741	-94.089	29.262
427080006100	6	06 1965	06 1965	PA	9,085'	741	-94.089	29.271
427080007300	8	07 1967	08 1967	ST	9,452'	741	-94.103	29.260
427080007301	9	08 1967	08 1967	COM	9,020'	741	-94.106	29.260
427084023300	11	04 1986	01 1996	ST	10,824'	741	-94.075	29.262
	11	01 1996	04 1996	COM	11,033'	741	-94.075	29.262
427084023301					1			

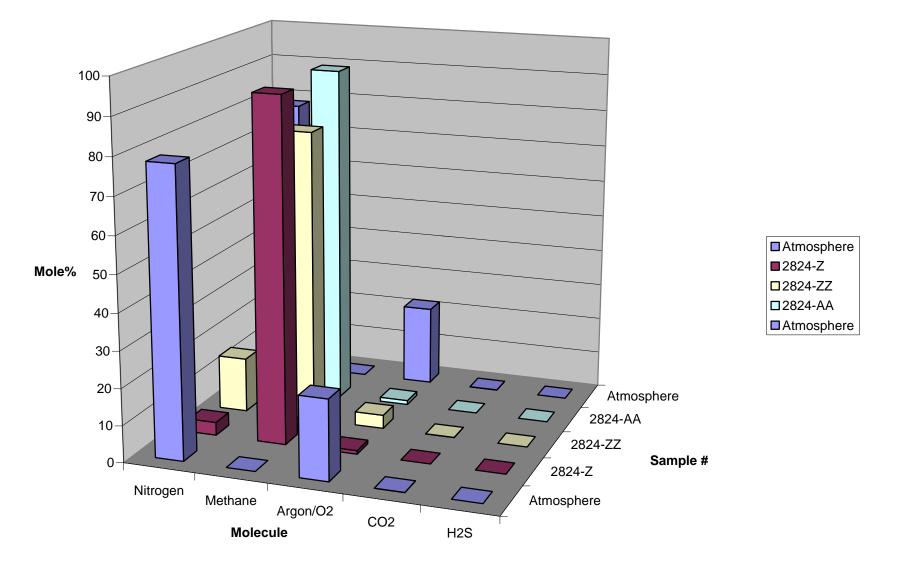


#### PIPELINE FLUSHING AND SAMPLING RECORD

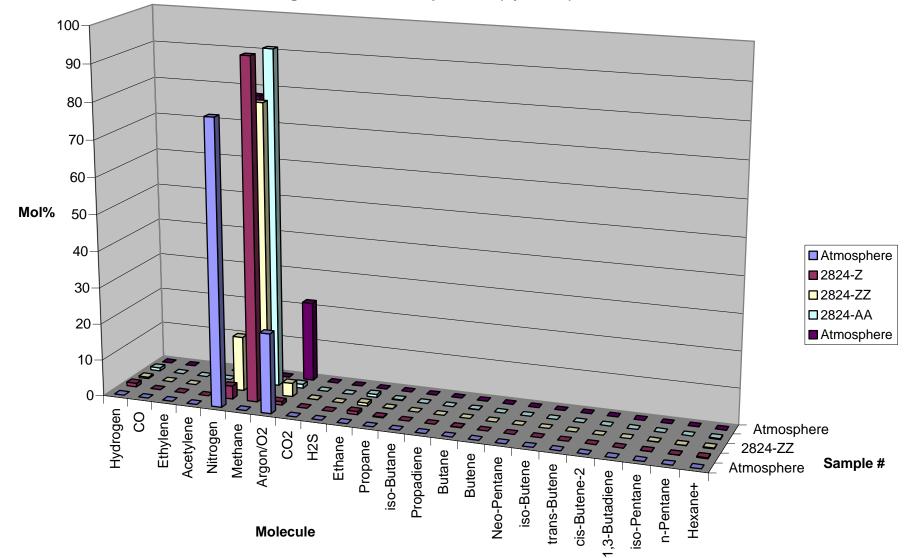
I. Pipeline Information	
MMS Segment No.	2824
Date:	10/26/2000
Pipeline Origination	
Area	High Island
Block	135
Platform	#5
Lease	#5 OCS-G-0742
Pipeline Destination	003-0-0742
Årea	High Island
Block	136
Platform	A
Lease	OCS-G-0742
Pipeline Size (in)	4
Pipelines Length (ft)	11500
Pipeline Volume (bbls)	179
II. Flushing Information	
Flushing Information	
Volume Flushed	12,000g
Flow Rate (GPM)	100
Pigged Used	No
Type of Pig	No
Size of Pig	
Clean Returns	ves
Inhibitor	
Chemical Inhibitor Used	
Type of Chemical	
Quantity of Chemical	
Origination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Destination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Comments:	
Company Representative	
Signature	

III. Sampling Data - Tracking	g Information			
Sample Location Platform: Pipeline Sampling Site: Flushing Start Time:	Hi 136A	tion w/ Sampling Spo	ool	
Gas Samples	Sample ID	Sample Date	Vol. Flushed (g)	H2S (PPM)
Vacuum Tubes			(g)	
	None			
Plastic Bags	2824-Z	10/2/ /2000		
	2824-Z	10/26/2000		
	2824-22 2824-AA	10/26/2000	4,000	
	2024-AA	10/28/2000	4,000	
Water Samples	Sample ID	Sample Date	Vol. Flushed (g)	Notes
Mineral Pattern Analysis				
	2824-1	10/26/2000	6,000	
	2824-2	10/26/2000		
	2824-3	10/26/2000		
	2824-4	10/26/2000		
Oil and Grease Analysis	2824-5	10/26/2000		
	2824-B	10/26/2000	6,000	
	2824-A	10/26/2000		
	2824-C	10/26/2000		
	2824-D	10/26/2000		
	2824-E	10/26/2000	12,000	
Comments:				
Company Representative	James Wisen	nan		
Signature				





# Figure 23 - Gas Composition (Major Constituents) - 2824



### Figure 24 - Gas Composition (by Mol%) - 2824

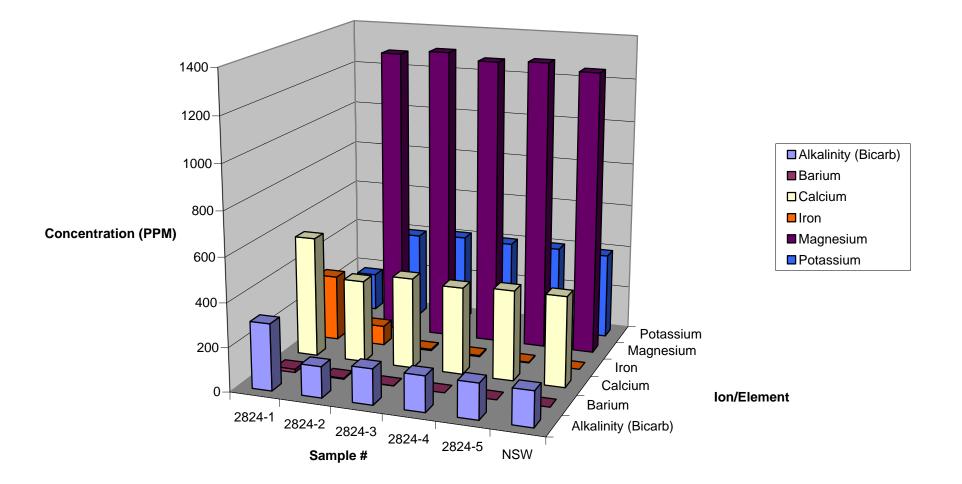


Figure 26 - Chloride and Sulfate - 2824

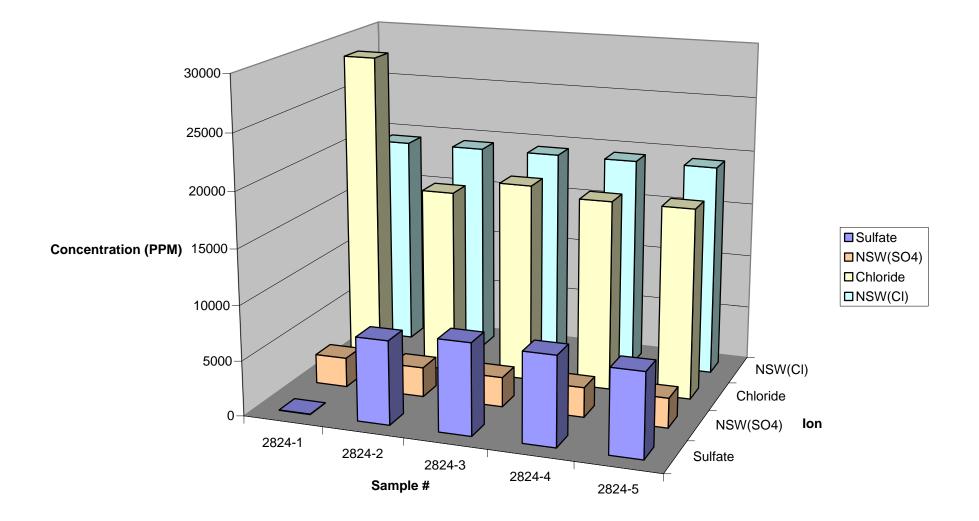
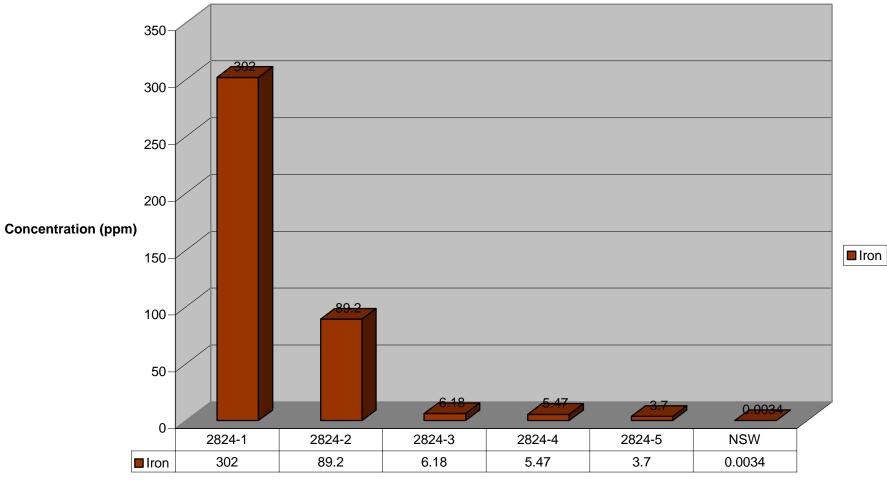
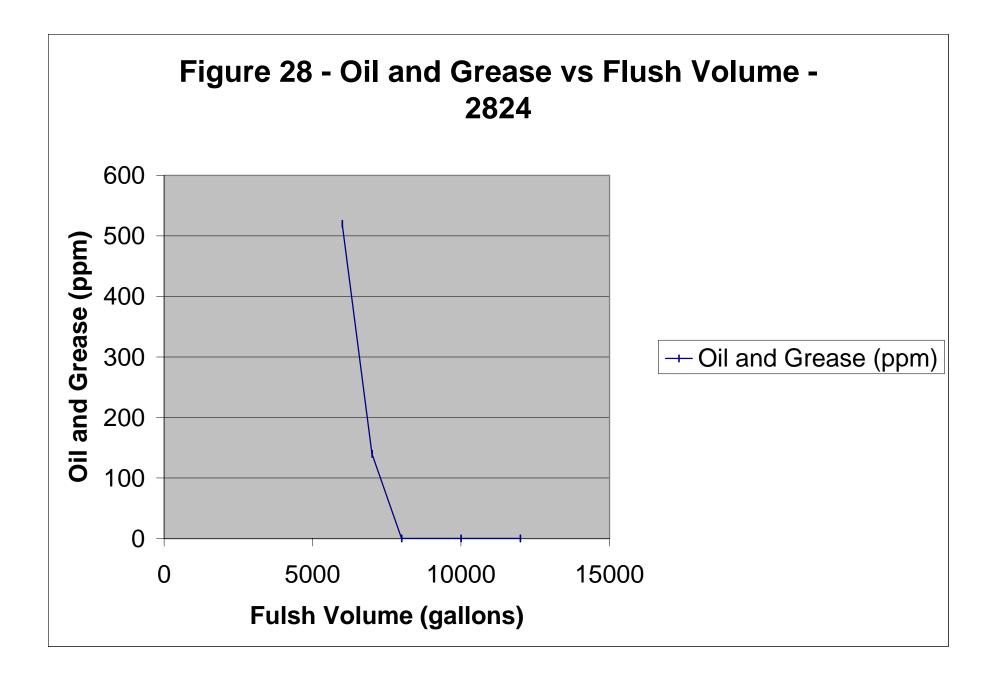
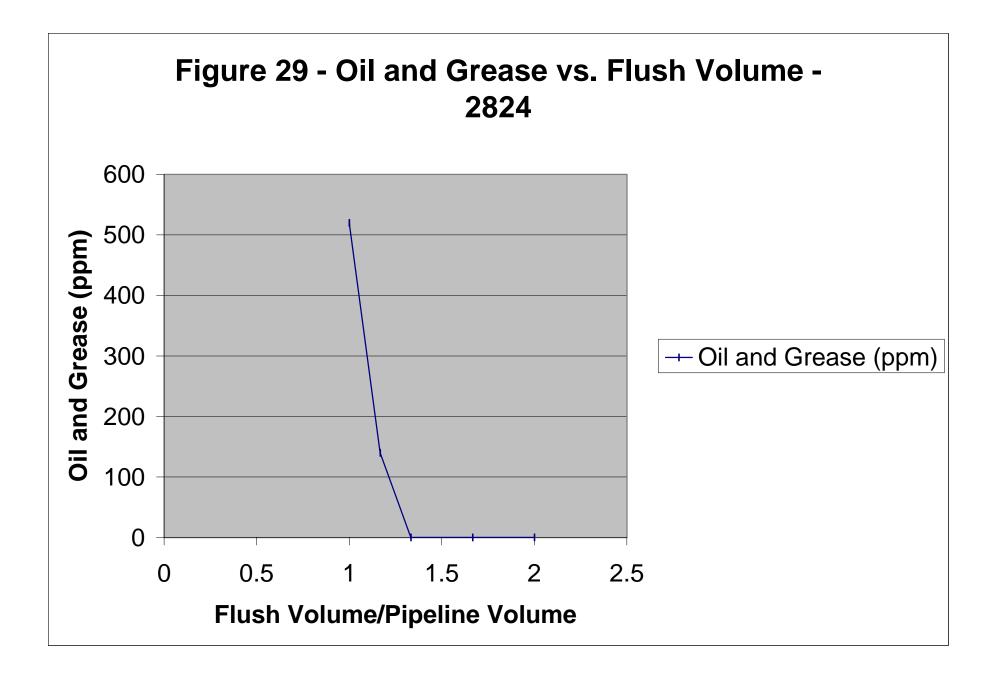


Figure 27 - Iron Concentration



Sample #





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HOUSTON LABORATORY 8680 INTERCHANGE DRIVE HOUSTCIN, TEXAS 77054 PHONE (713) 660-0901

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PAGE 1 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-003A

Sample ID.: 2824-AA : 10/26/00 WM0070

For : Winmar Attn: James Wiseman : 5700 NW Central Dr. Suite 150 : Houston, Texas 77092

TCD Analysis:

~

COMPONENTS	SAMPLE MOL %
	***********
Hydrogen	0.817
Carbon Dioxide	0.000
Ethylene	0.000
Ethane	0.824
Acetylene	0.000
Argon/Oxygen	1.108
Nitrogen	4.276
Methane	92.472
Carbon Monoxide	0.000

UnNormalized, Mol% : 96.842

Specific Gravity : 0.5879 (Air = 1.000 @ 60F)

(MIL = 1.000 @ 00.	- /	Net	Gross
BTU / ft3	:	872.8 Dry	969.0 Dry
( @ 14.65 & 60F)		857.6 Wet	952.2 Wet

SOUTHERN PETROLEUM LABORATORIES, INC.

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Fred C. DeAngelo

11-30-00:14:18 :SPL HOUSTON



HOUSTON LABORATORY 8680 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

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PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-003A

Sample ID.: 2824-AA : 10/26/00 WM0070

For : Winmar

Attn: James Wiseman

- : 5700 NW Central Dr. Suite 150
  - : Houston, Texas 77092

FID Analysis:

COMPONENTS \_\_\_\_\_ Hexanes Plus Methane Ethane/Ethylene Propane Propylene iso-Butane Propadiene n-Butane Butene-1 Neo-Pentane iso-Butene trans-Butene-2 cis-Butene-2 1,3-Butadiene iso-Pentane n-Pentane

SAMPLE MOL % \_\_\_\_\_\_ 0.264 92.472 0.824 / 0.000 0.133 0.000 0.031 0.000 0.032 0.000 None Detected 0.000 0.000 0.000 0.000 0.022

0.023

SOUTHERN PETROLEUM LABORATORIES, INC.

Fruch We Un Fred C. DeAngelo

IN-SO-COINSIGE (SPL HOUSTON)



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

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PAGE 3 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-003A

Sample ID.: 2824-AA : 10/26/00 WM0070

For : Winmar

- Attn: James Wiseman
  - : 5700 NW Central Dr. Suite 150
    - : Houston, Texas 77092

Completed Analysis:

Component	MOL %			WT%		
Hydrogen	0.817			0.097		
Carbon Dioxide	0.000			0.000		
Carbon Monoxide	0.000			0.000		
Ethylene	0.000			0.000		
Acetylene/Propylene	0.000	/	0.000	0.000	1	0.000
Argon/Oxygen	1.108			2.082		
Nitrogen	4.276			7.038		
Methane	92.472			7.116		
Ethane	0.824			1.455	•	
Propane	0.133			0.344		
iso-Butane	0.031			0.105		
Propadiene	0.000			0.000		
n-Butane	0.032			0.109		
Butene-1	0.000			0.000		
Neo-Pentane	0.000			0.000		
iso-Butene	0.000			0.000		
trans-Butene-2	0.000			0.000		
cis-Butene-2	0.000			0.000		
1,3-Butadiene	0.000			0.000		
iso-Pentane	0.022			0.092		
n-Pentane	0.023			0.097		
Hexane Plus	0.264			1.466		
	100.000		10	0.000		

SOUTHERN PETROLEUM LABORATORIES, INC.

Freed She Angelo

Fred C. DeAngelo

11-30-00:14:18 (SPL HOUSTON



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

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PAGE 1 OF 3

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CERTIFICATE OF ANALYSIS NUMBER : 110003-004A

Sample ID.: 2824-ZZ : 10/26/00 WM0070

For : Winmar Attn: James Wiseman : 5700 NW Central Dr. Suite 150 : Houston, Texas 77092

TCD Analysis:

COMPONENTS	SAMPLE MOL %
Indrogon	0.414
Hydrogen Carbon Dioxide	0.020
Ethylene	0.020
Ethane	0.761
Acetylene	0.000
Argon/Oxygen	3.709
Nitrogen	15.000
Methane	79.591
Carbon Monoxide	0.000

UnNormalized, Mol% : 95.310 Specific Gravity : 0.6486 (Air = 1.000 @ 60F)Net Gross \_\_\_\_ \_ \_ \_ \_ \_ BTU / ft3 754.2 Dry : (@ 14.65 & 60F) 741.1 Wet

837.2 Dry 822.6 Wet

SOUTHERN PETROLEUM LABORATORIES, INC.

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11-30-00:14:18 :SPL HOUSTON

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-004A

Sample ID.: 2824-ZZ : 10/26/00 WM0070

For : Winmar

Attn: James Wiseman

- : 5700 NW Central Dr. Suite 150
  - : Houston, Texas 77092

FID Analysis:

COMPONENTS -----Hexanes Plus Methane Ethane/Ethylene Propane Propylene iso-Butane Propadiene n-Butane Butene-1 Neo-Pentane iso-Butene trans-Butene-2 cis-Butene-2 1,3-Butadiene iso-Pentane n-Pentane

SAMPLE MOL % -----0.274 79.591 0.761 0.000 0.127 0.000 0.029 0.000 0.031 0.000 None Detected 0.000 0.000 0.000 0.000 0.021

0.023

SOUTHERN PETROLEUM LABORATORIES, INC.

Fred C. DeAngelo

11-50-00:14:18 :SPL HOUSTON

:7138608025

HOUSTON LABORATORY B680 INTERCHANGE DRIVE HOUSTCIN, TEXAS 77054 PHONE (713) 660-0901

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PAGE 3 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 110003-004A

Sample ID.: 2824-ZZ : 10/26/00 WM0070

For : Winmar

~`}

Attn: James Wiseman

- : 5700 NW Central Dr. Suite 150
  - : Houston, Texas 77092

Completed Analysis:

Component	MOL %			WT%	
Hydrogen	0.414			0.044	
Carbon Dioxide	0.020			0.048	
Carbon Monoxide	0.000			0.000	
Ethylene	0.000			0.000	
Acetylene/Propylene		/	0.000	0.000 /	0.000
Argon/Oxygen	3.709			6.318	
Nitrogen	15.000			22.379	
Methane	79.591			67.965	
Ethane	0.761			1.218	
Propane	0.127			0.298	
iso-Butane	0.029			0.090	
Propadiene	0.000			0.000	
n-Butane	0.031			0.095	
Butene-1	0.000			0.000	
Neo-Pentane	0.000			0.000	
iso-Butene	0.000			0.000	
trans-Butene-2	0.000			0.000	
cis-Butene-2	0.000			0.000	
1,3-Butadiene	0.000			0.000	
iso-Pentane	0.021			0.080	
n-Pentane	0.023			0.087	
Hexane Plus	0.274			1.377	
	100.000			100.000	

SOUTHERN PETROLEUM LABORATORIES, INC.

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054

(713) 660-0901

Client Sample ID 2824-B		Collected:			SPL Sample ID: 001	0898-02
-		Site: WM0070				
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analys	t Seq.#
OIL & GREASE, TOTAL RECOVI			MCL	E413.1	Units: mg/L	101005
Oil & Grease, Total Recoverable	520	2.0		1 E	11/06/00 9:00	461295

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogato Recovery Unreportable due to Dilution MI - Matrix Interference



HOUSTON LABORATORY 8980 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 650-0901

Client Sample ID 2824-A		Co	lected:		SPL Sample ID:	00100898-01
		Site	e; WM	0070		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed An	alyst Seq. #
OIL & GREASE, TOTAL RECOVI	ERABLE		MCL	E413.1	Units: mg/L	
Oil & Grease, Total Recoverable	140	2.0		1 E	11/06/00 9:00	461292

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:56:27 AM



HOUSTON LABORA.TORY 5380 INTERCMANGE DRIVE HOUSYON, TEXAS 77054 (713) 650-0901

Client Sample ID 2824-C		Collected:			SPL Sample ID: 001	0898-03
		Site	: WMO	)7 <b>0</b>		
Analyses/Method	Result	Rep.Limit	C	il Factor QUAL	Date Analyzed Analyst	Søq.#
OIL & GREASE, TOTAL RECOVE	RABLE		MCL	E413.1	Units: mg/L	
Oil & Grease, Total Recoverable	ND	2.0		1	11/06/00 9:00	461297

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MC\_) D. Surrogate Recovery Unreportable due to Dilution MI - Matrix interference

11,7/00 11:56:27 AM



HOUSTON LABORATORY 8860 INTERCHANGE ERIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824-D		Collected:				SPL Sample ID: 0010		0898-04
		Site: WM0070						
Analyses/Method	Result	Rep.Limit		Dil. Factor	QUAL	Date Analyzed	Analyst	Seq.#
OIL & GREASE, TOTAL RECOVERABLE			MCL	E	413.1	Units: mg/	Ľ	
Oil & Grease, Total Recoverable	ND	2.0		1		11/06/00 9:00		461299

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

11/7/00 11:58:27 AM



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, YEXAS 77054 (713) 660-0901

Client Sample ID 2824-E		Collected:			SPL Sample ID: 0010		0898-05
		Site	e: WMO				
Analyses/Method	Result	Rep.Limit	1	Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
OIL & GREASE, TOTAL RECOVE	RABLE		MCL	E413.1	Units: mg/	L.	
Oil & Grease, Total Recoverable	ND	2.0		1	11/06/00 9:00		461301

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

11/7/00 11:55:28 AM



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 282	4-1	Colle	ected:		SPL Sample ID: 0010	0896-02
		Site:	HI 1	135/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARE Alkalinity, Bicarbonate	BONATE 303	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	461519
ALKALINITY, CARBO Alkalinity, Carbonate	NATE ND	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460245
CHLORIDE, TOTAL Chloride	28800	500	MCL	<b>E325.3</b> 500	Units: mg/L 11/08/00 11:00 CV	468312
METALS BY METHOD	0 6010B, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	15.7	0,005		1	11/10/00 21:29 E_B	471277
Calcium	550	1		10	11/10/C0 21:35 E_B	471278
Iron	302	0.2		10	11/10/00 21:35 E_B	471278
Magnesium	128	0.1		1	11/13/C0 15:47 JM	471753
Potassium	176	2		1	11/13/00 15:47 JM	471763
Run ID/Seq #: TJ.	A_001110C-471277					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seg #: TJ	A_001110C-471278					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJ	A_001113B-471763					
Prep Method	Prep Date	Prep Initials				
SW3010A	11/01/2000 8:30	MR			· · · · · · · · · · · ·	
PH			MCL	E150.1	Units: pH Units	· · · · · · · · · · · · · · · · · · ·
рH	5.5	0.10		1	11/01/00 16:00 EC	458736
	•	•	MOL	400.4	t state and a state of the stat	
RESISTANCE @ 25 C	0.054	0.00100	MCL	120.1	Units: Mohms/cm 11/03/00 9:15 C V	461398
Resistance	U.U.U.	0.00100		<b>1</b>	1103/00 3.10 0_0	401000
SPECIFIC GRAVITY	· · · · · · · · · · · · · · · · · · ·	••••••••••••••••	MCL	ASTM D-1429	Units: Specific G	avity @
Specific Gravity	1.045	0		1	11/06/00 11:00 C_V	462230
CHARATE TOTAL			MOL	5. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	Limitary man /	
SULFATE, TOTAL	35	5	MCL	<b>E375.4</b>	Units: mg/L 11/01/00 10:00 SN	458826
Sulfate	30			<b>.</b>		-10020
TOTAL DISSOLVED S Total Dissolved Solids,		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471967
TOTAL SODIUM, CAL Total Sodium, Calculate		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471984

----Qualifiers:

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ND/U - Not Detected at the Reporting Limit

6 - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

. ..

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogale Recovery Unreportable due to Dilution MI - Matrix Interference

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HOUSTON LABORATORY 8850 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824-1		Collected:			SPL Sample ID:	00100	0896-02
		Site	: HI1	35/136			
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed	Analyst	Seq. #
TOTAL SUSPENDED SOLIDS			MCL	E160.2	Units: mg/	Ĺ	
Suspended Solids (Residue,Non- Filterable)	9380	20		5	11/02/00 15:00 E	EC	461987

Qualifiers:

ND/U - Not Detected at the Reporting Limit

6 - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

11/14/00 10:58:00 AM



HOUSTON LABORATORY BBBJ INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824-	2	Coli	ected:	SPL Sample ID: 00100896-0		
		Site	: HI1	35/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARBO Alkalinity, Bicarbonale	<b>NATE</b> 141	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	461521
ALKALINITY, CARBON Alkalinity, Carbonate	ND	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	460245
CHLORIDE, TOTAL Chioride	16900	500	MCL	<b>E325.3</b> 500	Units: mg/L 11/08/00 11:00 CV	468313
METALS BY METHOD 6	010B, TOTAL		MCL	SW6010B	Units: mg/L	
Barium	4.59	0.005		1	11/10/00 21:44 E_B	471279
Caicium	375	0.1		1	11/10/00 21:44 E_B	471279
Iron	69.2	0.02		1	11/10/00 21:44 E_B	471279
Magnesium	1320	1		10	11/13/00 16:04 JM	471767
Potassium	397	20		10	11/13/00 16:04 JM	471767
Run ID/Seq #: TJA_	001110C-471279		• • • •			•••
·• ·· =	Prep Date	Prep Initials				
SW3010A	1/01/2000 8:30	MR				
Run ID/Seq #: TJA_	001113B-471767					
Prep Method	Prep Date	Prep initials				
SW301DA	11/01/2000 8:30	MR			· · · · · ·	
РН			MCL	E150.1	Units: pH Units	
рн	7.8	D.10		1	11/01/00 16:00 EC	458737
	•	•		120.4	Linier Blainn (am	•••
RESISTANCE @ 25 C Resistance	0.0042	0.00100	MCL	120.1	Units: Mohms/cm 11/03/00 9:15 C V	461399
Resistance	0.0042	0.00100				401050
SPECIFIC GRAVITY			MCL	ASTM D-1429	Units: Specific Gr	avity @
Specific Gravity	1.029	0		1	11/06/00 11:00 C_V	462231
SULFATE, TOTAL			MCL	E375.4	Units: mg/L	
Sulfate	7600	1000		1000	11/01/00 10:00 SN	458827
		1			1. 1	: : ·
TOTAL DISSOLVED SO Total Dissolved Solids, Ca	· ·	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471968
a and a second		-	. <u>.</u>			
TOTAL SODIUM, CALC Total Sodium, Calculated	ULATED 11400	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471985
TOTAL SUSPENDED So Suspended Solids (Reside Filterable)		16	MCL	E160.2 4	Units: mg/L 11/02/00 15:00 EC	461988

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank.

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogale Recovery Unreportable due to Dilution MI - Matrix Interference Page 1



HOUSTON LABORATORY 8860 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824-3			Collected:			SPL Sample ID: 00	100896-04
			Site	: ні 1	35/136		
Analyses/Method	R	esuit	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analys	•
ALKALINITY, BICARE Alkalinity, Bicarbonate	BONATE	162	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	461524
ALKALINITY, CARBO Alkalinity, Carbonate	NATE	ND	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460247
CHLORIDE, TOTAL Chlaride	1	8200	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468316
METALS BY METHOD Barium	ware set of the set	).487	0.005	MCL	SW6010B	Units: mg/L 11/10/00 21:53 E_B	471280
Calcium Iron		413 6.18	0.1 0.02		1 1	11/10/C0 21:53 E_B 11/10/C0 21:53 E B	471280 471280
Magnesium Potassium		1340 410	1 20		10 10	11/13/C0 16:09 JM 11/13/C0 16:09 JM	471768 471768
Prep Method	A_001110C-471280		Prep Initials			· · · ·	
	11/01/2000 8:30 A_001113B-471768	ł	MR				
Prep Method SW3010A	Prep Date 11/01/2000 8:30	· · · ·	Prep Initials				
<b>ԲH</b> թH		8.1	0.10	MCL	E150.1	Units: pH Units 11/01/00 16:00 EC	458738
RESISTANCE @ 25 C Resistance		ND	0.00100	MCL	<b>120.1</b>	Units: Mohms/ci 11/03/00 9:15 C_V	<u>m</u> 461400
SPECIFIC GRAVITY Specific Gravity	- 1	1.019	0	MCL	ASTM D-1429 1	Units: Specific C 11/06/00 11:00 C_V	Gravity @ 452232
SULFATE, TOTAL Sulfate	· · · · ·	8300	1000	MCL	E375.4 1000	Units: mg/L 11/01/00 10:00 SN	458828
TOTAL DISSOLVED S Total Dissolved Solids.		1400	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471969
TOTAL SODIUM, CAL Total Sodium, Calculate		2600	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471986
TOTAL SUSPENDED Suspended Solids (Res Filterable)		41	4	MCL	E160.2	Units: mg/L 11/03/00 15:45 EC	-462116

------Qualifiers:

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ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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11/14/00 10:59:02 AM

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824	-4	Co	llected:		SPL Sample ID: 0	0100896-05
		Sit	e: HI1	35/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analy	yst Seq.#
ALKALINITY, BICARBO Alkalinity, Bicarbonate		2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:C0 SN	461526
ALKALINITY, CARBON Alkalinity, Carbonate	ND	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	460248
CHLORIDE, TOTAL Chloride	17400	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468317
METALS BY METHOD	6010B, TOTAL	· ·	MCL	SW6010B	Units: mg/L	
Barium	0.184	0.005		1	11/10/00 21:58 E_B	471281
Cacium	397	0.1		1	11/10/00 21:58 E_B	47128
Iron	5.47	0.02		1	11/10/00 21:58 E_B	471281
Magnesium	1310	1		10	11/13/00 16:13 JM	471769
Potassium	402	20		10	11/13/00 16:13 JM	471768
Run ID/Seg #: TJA	_001110C-471281					•
Prep Method	Prep Date	Prep Initials	2			
5W3010A	11/01/2000 8:30	MR				
Run ID/Seq #: TJA	_001113B-471769					
Prep Method	Prep Date	Prep Initials	i.			
SW3010A	11/01/2000 8:30	MR				
РН			MCL	E150.1	Units: pH Units	5
рн	8.2	0.10		1	11/01/00 16:00 EC	458739
• • ·	· · · ·			· · · · · · · · · · · · · · · · · · ·	· · · · · · · · · · · · · · · · · · ·	
RESISTANCE @ 25 C Resistance	ND	D.00100	MCL	120.1	Units: Mohms/ 11/03/00 9:15 C_V	<u>cm</u> 46140
SPECIFIC GRAVITY			MCL	ASTM D-1429	Units: Specific	Gravity @
Specific Gravity	1.013	0	122. 2.44	1	11/06/00 11:00 C_V	46223
SULFATE, TOTAL Sulfate	8100	1000	MCL	<b>E375.4</b> 1000	Units: mg/L 11/01/00 10:00 SN	45882
TOTAL DISSOLVED S Total Dissolved Solids, (		10	MCL	TDS-MINERAL 1	Units: mg/L 11/13/00 18:00 ES	47197
TOTAL SODIUM, CAL Total Sodium, Calculate		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	47198
TOTAL SUSPENDED S Suspended Solids (Resi Filterable)	n et de la companya d	4	MCL	E160.2	Units: mg/L 11/03/00 15:45 EC	46211

Qualifiers:

ND/U - Not Detected at the Reporting Limit

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B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

Client Sample ID 2824	4-5	Coll	ected:		SPL Sample ID: 00100	896-06
		Site	: HI1	35/136		
Analyses/Method	Result	Rep.Limit		Dil. Factor QUAL	Date Analyzed Analyst	Seq. #
ALKALINITY, BICARB Alkalinity, Bicarbonate	<b><u>ONATE</u></b> 162	2	MCL	M2320 B	Units: mg/L 11/01/00 14:00 SN	461527
ALKALINITY, CARBO Alkalinity, Carbonate	ND ND	2	MCL	M2320 B 1	Units: mg/L 11/01/00 14:00 SN	460249
CHLORIDE, TOTAL Chioride	17400	250	MCL	<b>E325.3</b> 250	Units: mg/L 11/08/00 11:00 CV	468318
METALS BY METHOD	6010B, TOTAL 0.0416	0.005	MCL	SW6010B	Units: mg/L 11/10/00 22:02 E_B	471282
Barium Calcium	410	0.1	M IA	1	11/10/00 22:02 E_B	471282
iron Magnesium	3 7 1320	0.02		1 10	11/10/C0 22:02 E_B 11/13/C0 16:17 JM	471282
Polassium Run ID/Seq #: TJA Prep Method	401 _001110C-471282 Prep Date	20 Prep Initials		10	11/13/C0 16:17 JM	471770
SW3010A Run ID/Seq #: TJA	11/01/2000 8:30	MR.				
Prep Method SW3010A	Prep Date 11/01/2000 8:30	Prep Inilials				
PH pH	8.2	0.10	MCL	<b>E150.1</b>	Units: pH Units 11/01/00 16:09 EC	458740
RESISTANCE @ 25 C Resistance	ND	0.00100	MCL	1 <b>20.1</b>	Units: Mohms/cm 11/03/00 9:15 C V	.∷ 461402
SPECIFIC GRAVITY Specific Gravity	1.016	0	MCL	ASTM D-1429	Units: Specific Grav 11/06/00 11:00 C_V	
SULFATE, TOTAL Sulfate	7600	1000	MCL	<b>E375.4</b> 1000	Units: mg/L 11/01/00 10:00 SN	458830
TOTAL DISSOLVED S Total Dissolved Solids, (		10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471971
TOTAL SODIUM, CAL	CULATED	10	MCL	TDS-MINERAL	Units: mg/L 11/13/00 18:00 ES	471988
TOTAL SUSPENDED S Suspended Solids (Resi Filterable)	SOLIDS	4	MCL	E160.2 1	<u>Units: mg/L</u> 11/03/00 15:45 EC	462119

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference



# 9. Results and Observations – 2826

## a. Sample/Locations Observations

Gas samples were taken at the top of the riser before the blind flange was removed and when the odor of natural gas was present. All bolts and flange seals were intact before testing and did not indicate any leakage. H2S length of stain tests were performed at this time. The first gas samples were taken using both vacuum tubes and Tedlar bags. The remaining samples were taken using Tedlar bags. The riser appears to have been disconnected and blind flanged at some point in the past.

The water samples taken at the top of riser bleed valve seemed uniform, and representative of the flush fluid stream.

A five foot (5') sample of the pipeline was removed, which included the tubeturn to pipeline weld.

## b. Gas Composition Observations

The results of the gas analysis are plotted and summarized in the results section. Atmospheric air composition is also plotted for reference/comparison purposes.

Four gas samples were taken. Only the last sample (just before the slug of water arrived) was high in methane. The first sample was also about 30% methane. The gas samples from the center of the pipeline were mainly composed of atmospheric air. One explanation for this would be for the line to have been bled down and opened for some time-period. Since air is heavier than methane, the air would have "sunk" to the bottom of the pipeline, leaving the gas at the tops of the risers. The data confirms this, due to the fact that the sample furthest from the production platform (where the pipeline was opened) was almost completely methane.

The gas samples did not contain any H2S or CO2.

## c. Flushwater Composition Observations

The flushwater composition for segment 2826 is plotted in the results section. Natural Seawater composition is also plotted for comparison purposes. The ions/elements plotted are: Alkalinity (CO3), Barium, Calcium, Iron, Magnesium, and Potassium. Because of their high values (in PPM), Chlorides and Sulfates are plotted on a separate chart.

For the flushwater, the mineral pattern relative to NSW is summarized below:



Alkalinity (bicarb) – Higher Barium – Higher for first two samples Calcium – Higher for first two samples Iron – Higher Magnesium – Lower for first two samples, then Equal Potassium – Lower/Equal Chloride – Higher for first two samples, then Equal Sulfate – Lower for first two samples

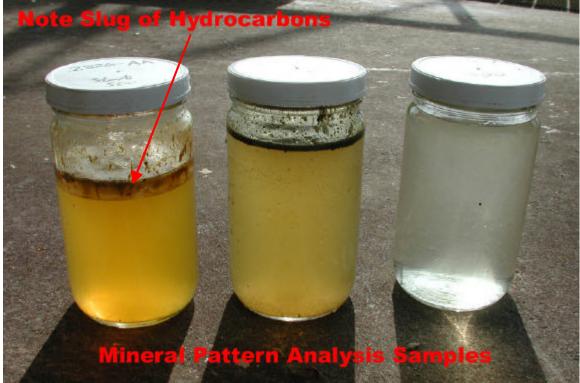


Photo #25

Again, the iron content is plotted as a separate graph in order to focus on these values. The first sample taken at the very front of the flushwater "slug" has an extremely high iron concentration of 117 ppm (ppm also equals milligrams/liter). The concentration is over 34,000 times greater than NSW. As with segment 2824, the first sample was taken right at the very front of the flushwater "slug." In fact, in anticipation of the incoming fluid, the sampling valve was left open, so that the very first fluid out of the pipeline was taken as the first water sample. The slug picked up quite a bit of debris, and was very high in condensate, as is evidenced in the photographs. This debris included metal particles which were picked up from the pipe wall. The sampling procedure "dissolved" these metal particles and recorded them as a concentration value. The following four samples were lower in concentration, but still much higher than NSW values.



The ions/elements to focus on from this analysis are those found in steel corrosion products: FeO2, FeS. The samples showed higher than NSW concentrations of iron, but Sulfate was below NSW levels for the first two samples, and above NSW levels for the last sample.

## d. Oil and Grease Observations

At completion of flushing, oil and grease was non-detectible in the samples taken. The detection limit is 2.5 PPM. As noted above, the very first sample was high in hydrocarbons because it contained a good deal of the condensate that was present in the line. The photos show this condensate as a frothy brown/orange mixture on top of the water sample. The graph shows a very rapid drop in oil and grease concentration, with the non-detectable limit appearing to be reached at 1.75x flush volume.

## e. Pipe Cutout Observations

A five foot section of pipe was retrieved. This section was taken from the base of the platform, and included pipe on both sides of the weld connecting the pipeline to the riser/tubeturn. The sample included the pipe/tubeturn weld.



Photo #26 Sample Photo – 2826



This pipe sample appeared to be in very good condition. Light surface rust was present, but no metal loss patches were evident in the section retrieved.

The tubeturn/pipeline weld appeared to be in good condition as well.



Photo #27 - Sample Photo – 2826 Pipeline to Tubeturn Weld



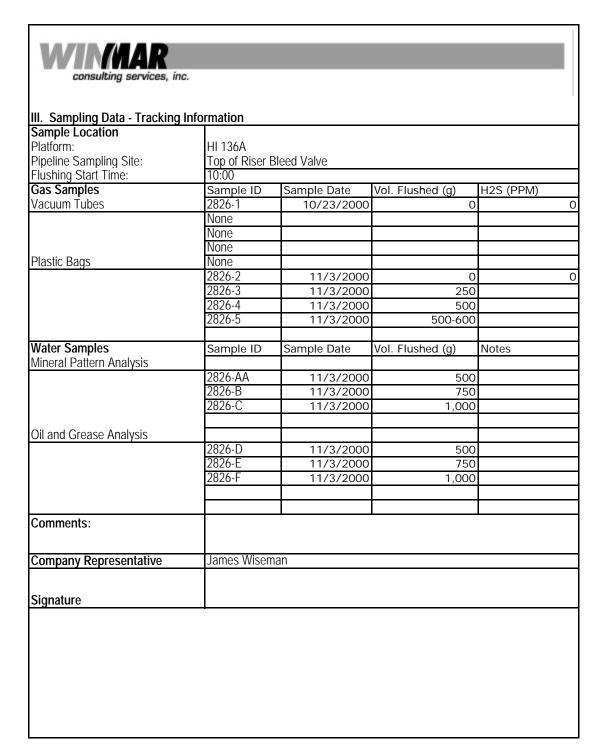
## SHELL OFFSHORE INC. HI-136-2

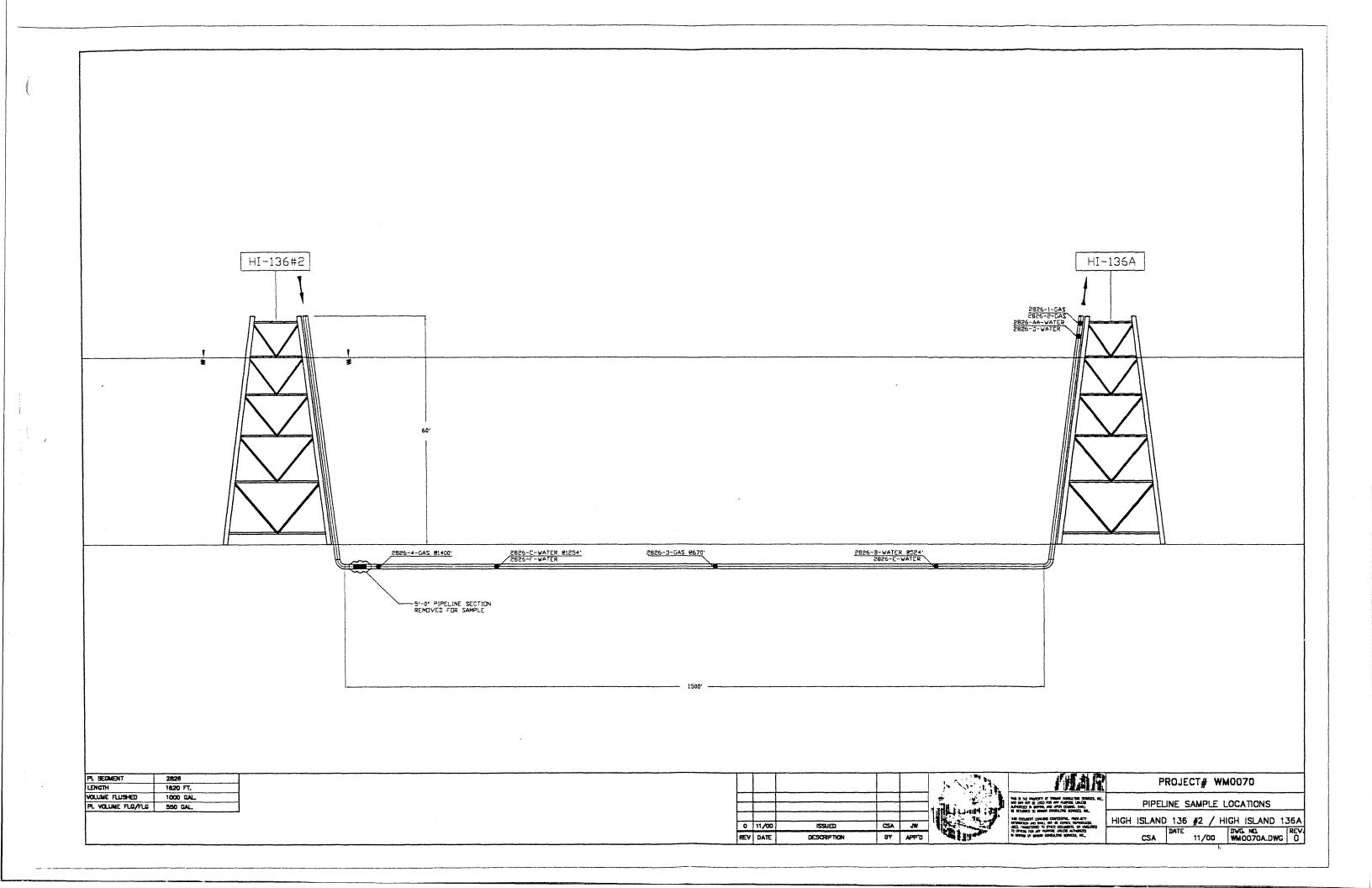
			Р	LATFOF	RM			
MMS Gene	S General ODS General MMS Location MMS Facility				cility			
Water	50 feet	Function	WP		Lease	742 feet	Helideck	Yes
Major	No	Piles	NA		Complex	10027	Quarters	None
Decks	1	Slots	3		Longitude	-94.126	Generator	No
Slots	3	ODS ID	743		Latitude	29.261	Cranes	NA
Wells	2	Previous 1	NA		X	3,553,610'	Gas	Yes
Flare	No	Previous 2	NA		Y	551,545'	Oil	No
Installed	01 1965	Previous 3	NA		To Shore	25 miles	Comp	No
Revised	12 1998	Previous 4	NA		N-S feet	S 5065'	8 hour	No
Removed	NA	Notes	NA		E-W feet	E 1386'	24 hour	No
		I	PIF	PELINES	S MMS			
Segment	2826							
Origin	HI-136-2							
Terminus	HI-136-A							
O.D.	3"							
Length	1,500'							
Product	BLKG							
Status	ACT							
Installed	NA							
Abandon	NA							
Revised	Oct-98							
Operator	SHELL O	FFSHORE INC	•					
-			V		MMS		I	
API Well ID	Well	Spud	Revised	Status	MD	Bot Lease	Sur Long	Sur Lat
427080004300	1	05 1964	06 1964	PA	12,516'	742	-94.126	29.255
427080005100	2	09 1964	10 1985	ST	9,341'	742	-94.126	29.261
427080005101	2	10 1985	12 1989	COM	9,500'	742	-94.126	29.261
427080005200	3	12 1964	11 1995	ST	10,490'	742	-94.125	29.263
427080005201	3	11 1995	02 1996	COM	9,853'	742	-94.134	29.263
427080005300	4	01 1965	05 1975	PA	10,604'	742	-94.138	29.257
427080005400	5	02 1965	02 1965	COM	9,192'	742	-94.135	29.249
427080005500	6	05 1965	06 1965	PA	9,134'	742	-94.134	29.269
427080006300	7	07 1965	08 1965	PA	9,690'	742	-94.149	29.254
427084000100	8	06 1971	11 1985	ST	9,130'	742	-94.131	29.261
427084000101	8	11 1985	01 1986	COM	11,829'	742	-94.146	29.261
427084022000	10	12 1985	12 1995	ST	10,500'	741	-94.111	29.275
427084022001	10	12 1995	12 1995	COM	10,883'	741	-94.116	29.275
427084042200	10	01 1996	02 1996	СОМ	10,800'	742	-94.135	29.275

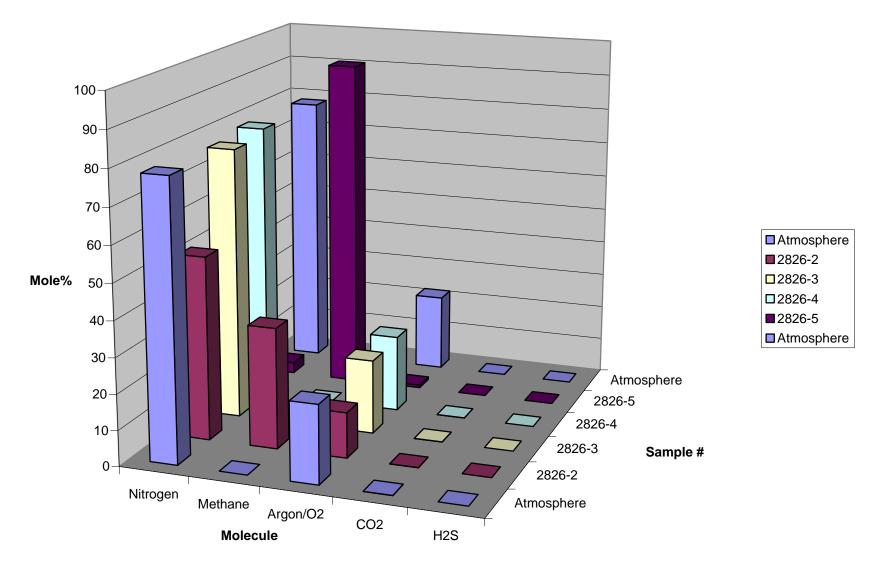


### PIPELINE FLUSHING AND SAMPLING RECORD

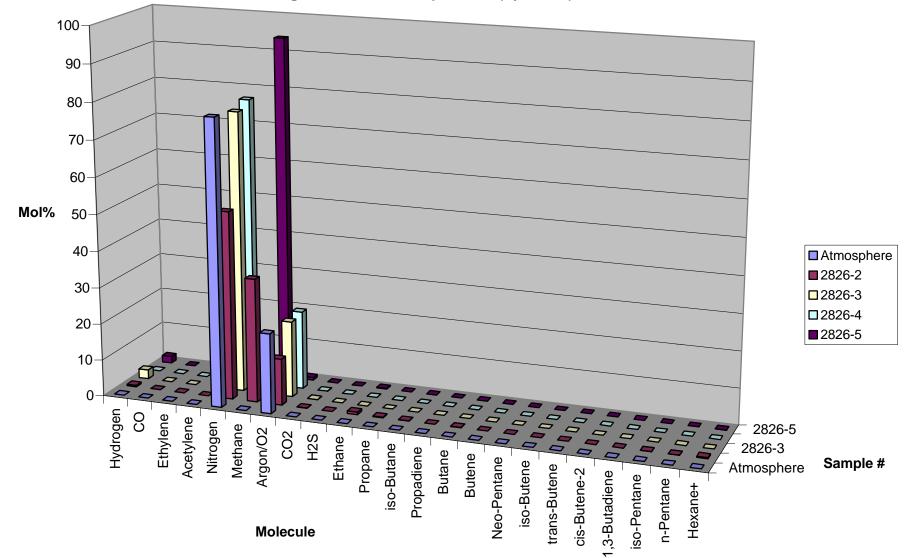
Dinalina Information	
I. Pipeline Information MMS Segment No.	2826
Date:	11/3/2000
Pipeline Origination	
Area	High Island
Block	136
Platform	#2
Lease	#Z OCS-G-0742
Pipeline Destination	003-0-0742
Area	High Island
Block	136
Platform	A
Lease	OCS-G-0742
Pipeline Size (in)	3
Pipelines Length (ft)	1,500
Pipeline Volume (bbls)	13
	1
II. Flushing Information	
Flushing Information	
Volume Flushed	1000 Gallons
Flow Rate (GPM)	~ 100
Pigged Used	No
Type of Pig	
Size of Pig	
Clean Returns	Yes
Inhibitor	
Chemical Inhibitor Used	
Type of Chemical	
Quantity of Chemical	
Origination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Destination Riser	
Riser blind flanged w/ vent valve	Yes
Pipeline Tagged	Yes
Comments:	
Company Representative	
Signature	
อเมาสนาย	







## Figure 31 - Gas Composition (Major Constituents) - 2826



## Figure 32 - Gas Composition (by Mol%) - 2826

Figure 33 - Flushwater Composition - 2826

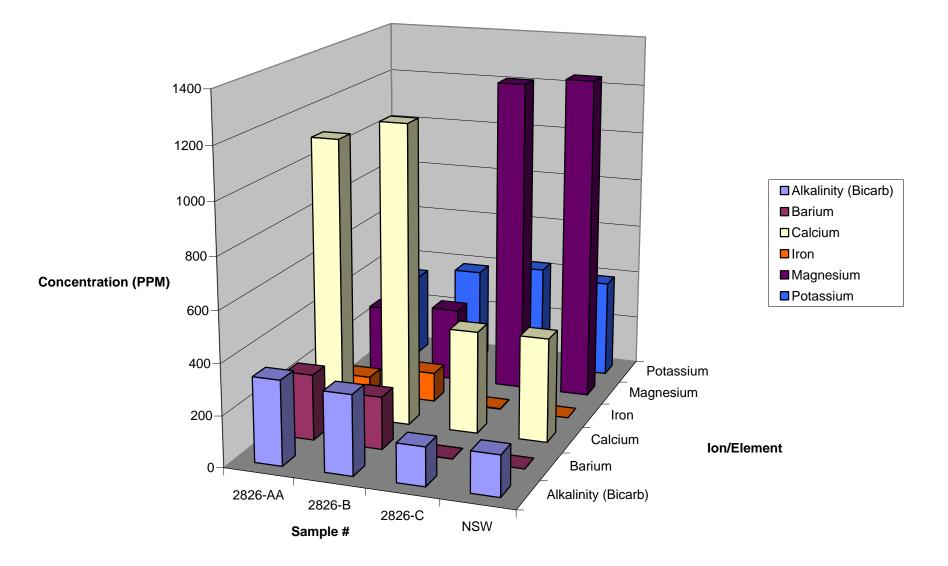


Figure 34 - Chloride and Sulfate - 2826

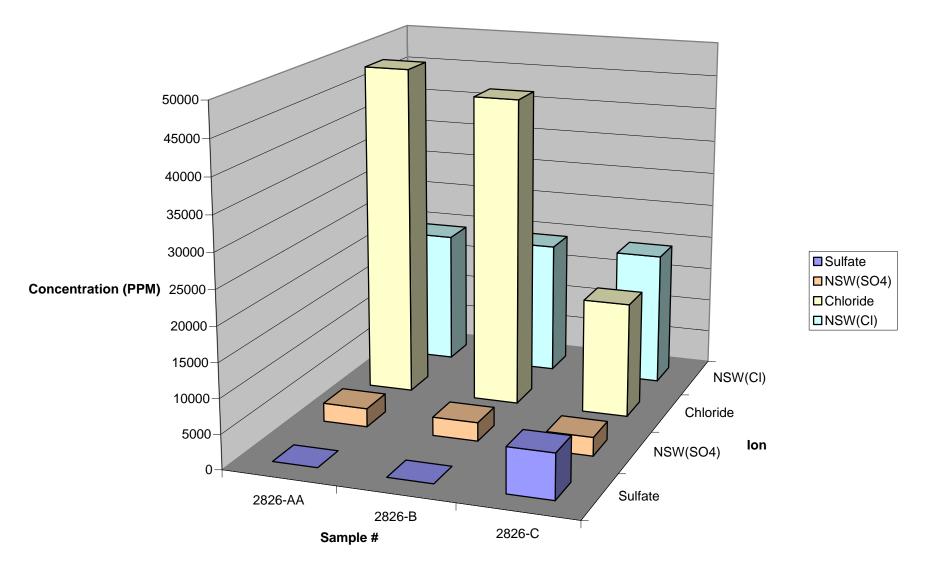
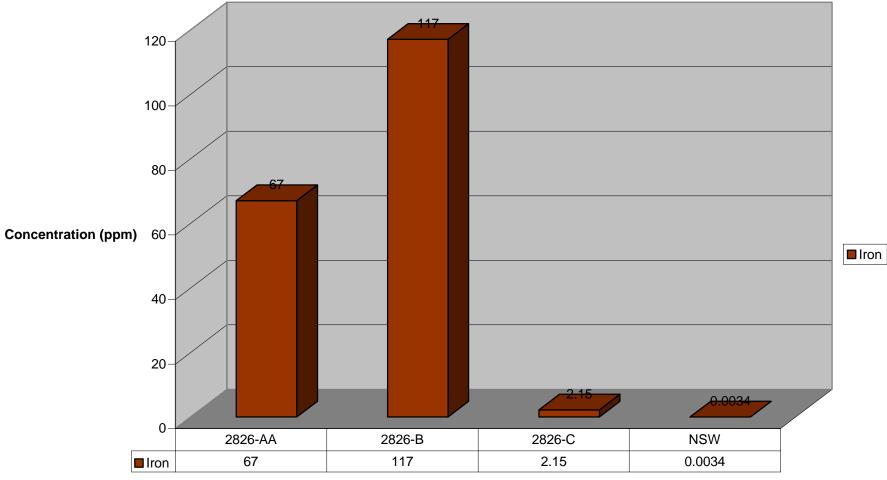
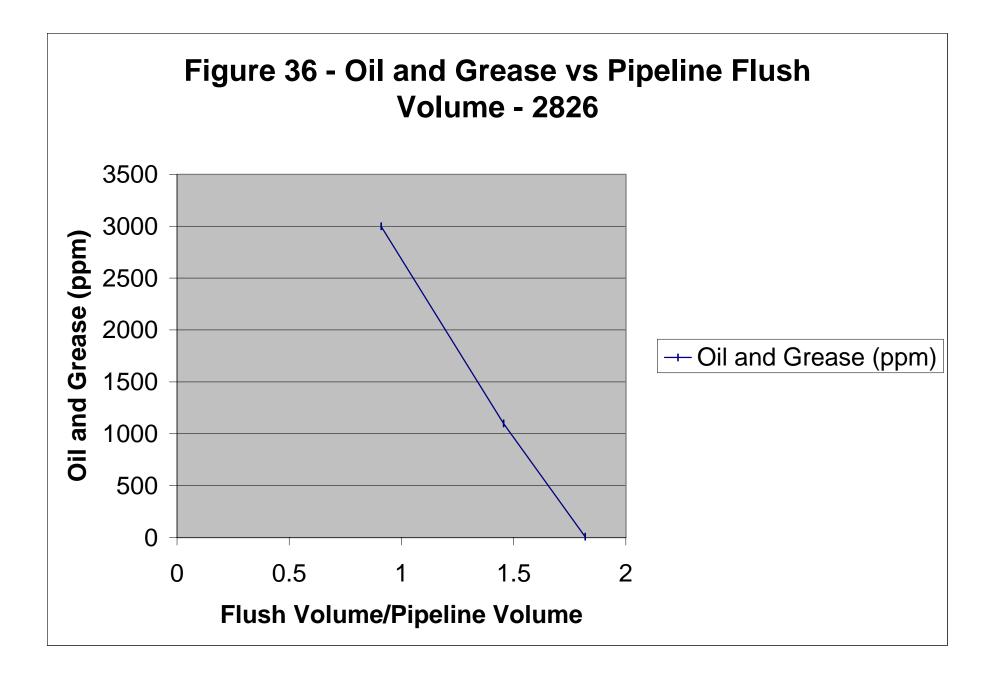
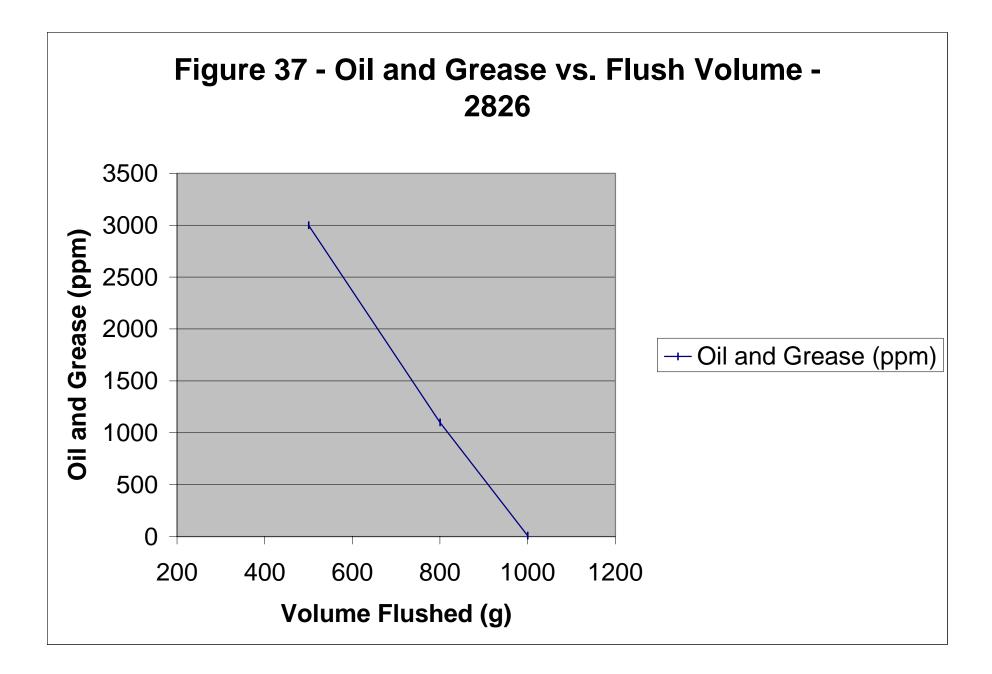


Figure 35 - Iron Concentration - 2826



Sample #





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PAGE 1 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 2000110048-001A

Sample ID.: 2826-2 : 11/03/00 :

For : WINMAR Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

TCD Analysis:

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COMPONENTS	SAMPLE MOL %
Hydrogen Carbon Dioxide Ethylene Ethane Acetylene Argon/Oxygen Nitrogen Methane Carbon Monoxide	$\begin{array}{c} 0.409\\ 0.049\\ 0.000\\ 0.702\\ 0.000\\ 12.764\\ 51.534\\ 33.872\\ 0.000\end{array}$

UnNormalized, Mol% :	99.566	
Specific Gravity : (Air = 1.000 @ 60F)	0.8526	
	Net	Gross
BTU / ft3 : (@14.65 & 60 F)	344.9 Dry 338.9 Wet	382.3 Dry 375.7 Wet

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PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 2000110048-001A

Sample ID.: 2826-2 : 11/03/00

For : WINMAR

Attn: JAMES WISEMAN

: 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

FID Analysis:

COMPONENTS	SAMPLE MOL %
Hexanes Plus	0.324
Methane	33.872
Ethane/Ethylene	0.702 / 0.000
Propane	0.144
Propylene	0.000
iso-Butane	0.067
Propadiene	0.000
n-Butane	0.055
Butene-1	0.000
Neo-Pentane	None Detected
iso-Butene	0.000
trans-Butene-2	0.000
cis-Butene-2	0.000
1,3-Butadiene	0.000
iso-Pentane	0.044
n-Pentane	0.036

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PAGE 3 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-001A

#### Sample ID.: 2826-2 : 11/03/00

#### For : WINMAR

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Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

Completed Analysis:

| Component           | MOL %   |   |       | WI%     |   |       |
|---------------------|---------|---|-------|---------|---|-------|
| Hydrogen            | 0.409   |   |       | 0.033   |   |       |
| Carbon Dioxide      | 0.049   |   |       | 0.088   |   |       |
| Carbon Monoxide     | 0.000   |   |       | 0.000   |   |       |
| Ethylene            | 0.000   |   |       | 0.000   |   |       |
| Acetylene/Propylene | 0.000   | 1 | 0.000 | 0.000   | 1 | 0.000 |
| Argon/Oxygen        | 12.764  |   |       | 16.534  |   |       |
| Nitrogen            | 51.534  |   |       | 58.474  |   |       |
| Methane             | 33.872  |   |       | 21.998  |   |       |
| Ethane              | 0.702   |   |       | 0.855   |   |       |
| Propane             | 0.144   |   |       | 0.258   |   |       |
| iso-Butane          | 0.067   |   |       | 0.158   |   |       |
| Propadiene          | 0.000   |   |       | 0.000   |   |       |
| n-Butane            | 0.055   |   |       | 0.130   |   |       |
| Butene-1            | 0.000   |   |       | 0.000   |   |       |
| Neo-Pentane         | 0.000   |   |       | 0.000   |   |       |
| iso-Butene          | 0.000   |   |       | 0.000   |   |       |
| trans-Butene-2      | 0.000   |   |       | 0.000   |   |       |
| cis-Butene-2        | 0.000   |   |       | 0.000   |   |       |
| 1,3-Butadiene       | 0.000   |   |       | 0.000   |   |       |
| iso-Pentane         | 0.044   |   |       | 0.128   |   |       |
| n-Pentane           | 0.036   |   |       | 0.104   |   |       |
| Hexane Plus         | 0.324   |   |       | 1.241   |   |       |
|                     | 100.000 |   |       | 100.000 |   |       |

#### SOUTHERN PETROLEUM LABORATORIES, INC.

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PAGE 1 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 2000110048-002A

Sample ID.: 2826-3 : 11/03/00 :

For : WINMAR Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

TCD Analysis:

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|               | COMPONENTS  | SAMPLE<br>MOL %  |
|---------------|---|--|
|               |   |  |
| , gesitterer, | Hydrogen<br>Carbon Dioxide<br>Ethylene<br>Ethane<br>Acetylene<br>Argon/Oxygen<br>Nitrogen<br>Methane<br>Carbon Monoxide | 2.506<br>0.038<br>0.000<br>0.000<br>20.839<br>76.616<br>0.000<br>0.000 |
|               |   |  |

UnNormalized, Mol%: 79.960 Specific Gravity: 0.9736 (Air = 1.000 @ 60F) Net Gross BTU / ft3 : 6.8 Dry 8.1 Dry (@ 14.65 & 60 F) 6.7 Wet 8.0 Wet

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HOUSTON LABORATORY 8680 INTERCHANGE DRIVE HOUSTCN, TEXAS 77054 PHONE (713) 660-0901

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PAGE 2 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 2000110048-002A

Sample ID.: 2826-3 : 11/03/00

For : WINMAR

- Attn: JAMES WISEMAN
  - : 5700 NW CENTRAL DRIVE
    - : HOUSTON TX. 77092

FID Analysis:

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| COMPONENTS      | SAMPLE<br>MOL % |       |
|-----------------|-----------------|-------|
| Hexanes Plus    | 0.000           |       |
| Methane         | 0.000           |       |
| Ethane/Ethylene |                 | 0.000 |
| Propane         | 0.000           |       |
| Propylene       | 0.000           |       |
| iso-Butane      | 0.000           |       |
| Propadiene      | 0.000           |       |
| n-Butane        | 0.000           |       |
| Butene-1        | 0.000           |       |
| Neo-Pentane     | None Detected   |       |
| iso-Butene      | 0.000           |       |
| trans-Butene-2  | 0.000           |       |
| cis-Butene-2    | 0.000           |       |
| 1,3-Butadiene   | 0.000           |       |
| iso-Pentane     | 0.000           |       |
| n-Pentane       | 0.000           |       |

#### SOUTHERN PETROLEUM LABORATORIES, INC.



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 PHONE (713) 660-0901

PAGE 3 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-002A

Sample ID.: 2826-3 : 11/03/00

#### For : WINMAR

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/m )

Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

Completed Analysis:

| Component                                       | MOL %                              |   |       | WT%                                |   |       |
|---|------------------------------------|---|-------|------------------------------------|---|-------|
| Hydrogen<br>Carbon Dioxide<br>Carbon Monoxide   | 2.506<br>0.038<br>0.000            |   |       | 0.179<br>0.059<br>0.000            |   |       |
| Ethylene<br>Acetylene/Propylene<br>Argon/Oxygen | 0.000<br>0.000<br>20.839<br>76.616 | / | 0.000 | 0.000<br>0.000<br>23.637<br>76.124 | 1 | 0.000 |
| Nitrogen<br>Methane<br>Ethane                   | 0.000                              |   |       | 0.000                              |   |       |
| Propane<br>iso-Butane<br>Propadiene             | 0.000<br>0.000<br>0.000            |   | •     | 0.000<br>0.000<br>0.000            |   |       |
| n-Butane<br>Butene-1                            | 0.000<br>0.000                     |   |       | 0.000                              |   |       |
| Neo-Pentane<br>iso-Butene<br>trans-Butene-2     | 0.000<br>0.000<br>0.000            |   |       | 0.000<br>0.000<br>0.000            |   |       |
| cis-Butene-2<br>1,3-Butadiene<br>iso-Pentane    | 0.000<br>0.000<br>0.000            |   |       | 0.000<br>0.000<br>0.000            |   |       |
| n-Pentane<br>Hexane Plus                        | 0.000<br>0.000                     |   |       | 0.000                              |   |       |
|   | 100.000                            |   |       | 100.000                            |   |       |

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PAGE 1 OF 3

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CERTIFICATE OF ANALYSIS NUMBER : 2000110048-003A

Sample ID.: 2826-4 : 11/03/00 :

For : WINMAR Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

TCD Analysis:

SAMPLE MOL % COMPONENTS \_\_\_\_\_ \_\_\_\_\_ Hydrogen 0.000 Carbon Dioxide 0.048 Ethylene 0.000 0.000 Ethane Acetylene 0.000 Argon/Oxygen 21.505 78.447 Nitrogen 0.000 Methane Carbon Monoxide 0.000

UnNormalized, Mol% : 84.391

Specific Gravity : 0.9971

| (Air = 1.000 @ | 60F) |                    |                    |
|----------------|------|--------------------|--------------------|
|                |      | Net                | Gross              |
|                |      |                    |                    |
| BTU / ft3      | :    | 0.0 Dry<br>0.0 Wet | 0.0 Dry<br>0.0 Wet |
| (@14.65 & 60   | F )  | 0.0 Wel            | U.U Met            |

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PAGE 2 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-003A

Sample ID.: 2826-4 : 11/03/00

For : WINMAR

Attn: JAMES WISEMAN

: 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

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FID Analysis:

| COMPONENTS   | SAMPLE<br>MOL %  |       |
|--|--|-------|
| Hexanes Plus<br>Methane<br>Ethane/Ethylene<br>Propane<br>Propylene<br>iso-Butane<br>Propadiene<br>n-Butane<br>Butene-1<br>Neo-Pentane<br>iso-Butene<br>trans-Butene-2<br>cis-Butene-2<br>1,3-Butadiene | 0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>None Detected<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000<br>0.000 | 0.000 |
| iso-Pentane<br>n-Pentane   | 0.000<br>0.000   |       |

#### SOUTHERN PETROLEUM LABORATORIES, INC.

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PAGE 3 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-003A

Sample ID.: 2826-4 : 11/03/00

#### For : WINMAR

/m )

Attn: JAMES WISEMAN

- : 5700 NW CENTRAL DRIVE
  - : HOUSTON TX. 77092

Completed Analysis:

| Component           | MOL %   |   | WT%                 |
|---------------------|---------|---|---------------------|
| Hydrogen            | 0.000   |   | 0.000               |
| Carbon Dioxide      | 0.048   |   | 0.073               |
| Carbon Monoxide     | 0.000   |   | 0.000               |
| Ethylene            | 0.000   |   | 0.000               |
| Acetylene/Propylene | 0.000   | / | 0.000 0.000 / 0.000 |
| Argon/Oxygen        | 21.505  |   | 23.818              |
| Nitrogen            | 78.447  |   | 76.109              |
| Methane             | 0.000   |   | 0.000               |
| Ethane              | 0.000   |   | 0.000               |
| Propane             | 0.000   |   | 0.000               |
| iso-Butane          | 0.000   |   | 0.000               |
| Propadiene          | 0.000   |   | 0.000               |
| n-Butane            | 0.000   |   | 0.000               |
| Butene-1            | 0.000   |   | 0.000               |
| Neo-Pentane         | 0.000   |   | 0.000               |
| iso-Butene          | 0.000   |   | 0.000               |
| trans-Butene-2      | 0.000   |   | 0.000               |
| cis-Butene-2        | 0.000   |   | 0.000               |
| 1,3-Butadiene       | 0.000   |   | 0.000               |
| iso-Pentane         | 0.000   |   | 0.000               |
| n-Pentane           | 0.000   |   | 0.000               |
| Hexane Plus         | 0.000   |   | 0.000               |
|                     | 100.000 |   | 100.000             |

### SOUTHERN PETROLEUM LABORATORIES, INC.

11-18-00:19:23 (SPL HOUSTON)



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PAGE 1 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-004A

Sample ID.: 2826-5 : 11/03/00 :

For : WINMAR Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

TCD Analysis:

| COMPONENTS   | SAMPLE<br>MOL %  |
|--|--|
| Hydrogen<br>Carbon Dioxide<br>Ethylene<br>Ethane<br>Acetylene<br>Argon/Oxygen<br>Nitrogen<br>Methane | 1.909<br>0.000<br>0.170<br>0.000<br>0.774<br>3.117<br>93.955 |
| Carbon Monoxide  | 0.000  |

UnNormalized, Mol% : 95.319

| Specific Gravity :<br>(Air = 1.000 @ 60F) | 0.5639                 |                        |
|---|------------------------|------------------------|
| (,  | Net                    | Gross                  |
|   |                        |                        |
| BTU / ft3 :<br>(@14.65 & 60 F)            | 862.0 Dry<br>847.0 Wet | 957.9 Dry<br>941.2 Wet |

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PAGE 2 OF 3

#### CERTIFICATE OF ANALYSIS NUMBER : 2000110048-004A

Sample ID.: 2826-5 : 11/03/00

For : WINMAR

Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

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FID Analysis:

SAMPLE COMPONENTS MOL 8 \_\_\_\_\_ \_\_\_\_\_ Hexanes Plus 0.022 Methane 93.955 Ethane/Ethylene 0.170 / 0.000 Propane 0.026 Propylene 0.000 iso-Butane 0.010 Propadiene 0.000 n-Butane 0.007 0.000 Butene-1 Neo-Pentane None Detected iso-Butene 0.000 trans-Butene-2 0.000 0.000 cis-Butene-2 1,3-Butadiene 0.000 iso-Pentane 0.005 n-Pentane 0.005

#### SOUTHERN PETROLEUM LABORATORIES, INC.

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PAGE 3 OF 3

CERTIFICATE OF ANALYSIS NUMBER : 2000110048-004A

Sample ID.: 2826-5 : 11/03/00

For : WINMAR

Attn: JAMES WISEMAN : 5700 NW CENTRAL DRIVE : HOUSTON TX. 77092

Completed Analysis:

| <br> | <br> | ~ | <br> | - | - | <br> | · | - | - | <br> | - | - | <br> | • - | <br> | - | - | <br> | · - | <br> | <br> | <br>- | <br> | <br>- | - | <br> | - | • | <br> | - | - • | <br> | - | <br> | - | <br>- |
|------|------|---|------|---|---|------|---|---|---|------|---|---|------|-----|------|---|---|------|-----|------|------|-------|------|-------|---|------|---|---|------|---|-----|------|---|------|---|-------|
|      |      |   |      |   |   |      |   |   |   |      |   |   |      |     |      |   |   |      |     |      |      |       |      |       |   |      |   |   |      |   |     |      |   |      |   |       |

| Component                  | MOL %          |   |       | WT%     |   |       |
|----------------------------|----------------|---|-------|---------|---|-------|
| Eydrogen<br>Carbon Dioxide | 1.909<br>0.000 |   |       | 0.236   |   |       |
| Carbon Monoxide            | 0.000          |   |       | 0.000   |   |       |
| Ethylene                   | 0.000          |   |       | 0.000   |   |       |
| Acetylene/Propylene        | 0.000          | / | 0.000 | 0.000   | / | 0.000 |
| Argon/Oxygen               | 0.774          |   |       | 1.517   |   |       |
| Nitrogen                   | 3.117          |   |       | 5.349   |   |       |
| Methane                    | 93.955         |   |       | 92.284  |   |       |
| Ethane                     | 0.170          |   |       | 0.313   |   |       |
| Propane                    | 0.026          |   |       | 0.071   |   |       |
| iso-Butane                 | 0.010          |   |       | 0.035   |   |       |
| Propadiene                 | 0.000          |   |       | 0.000   |   |       |
| n-Butane                   | 0.007          |   |       | 0.024   |   |       |
| Butene-1                   | 0.000          |   |       | 0.000   |   |       |
| Neo-Pentane                | 0.000          |   |       | 0.000   |   |       |
| iso-Butene                 | 0.000          |   |       | 0.000   |   |       |
| trans-Butene-2             | 0.000          |   |       | 0.000   |   |       |
| cis-Butene-2               | 0.000          |   |       | 0.000   |   |       |
| 1,3-Butadiene              | 0.000          |   |       | 0.000   |   |       |
| iso-Pentane                | 0.005          |   |       | 0.024   |   |       |
| n-Pentane                  | 0.005          |   |       | 0.021   |   |       |
| Hexane Plus                | 0.022          |   |       | 0.126   |   |       |
|                            | 100.000        |   |       | 100.000 |   |       |

SOUTHERN PETROLEUM LABORATORIES, INC.



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 560-0901

## Winmar Consulting Services

Certificate of Analysis Number: 00110147

| Report To:                 | Winmar Consulting<br>James Wiseman<br>5700 NW Central D<br>Suite 150                  |  |                         |     | Project Nar<br>Site<br>Site Addres              |             | WM0070<br>Hi 136   |                      |                              |      |
|----------------------------|---|--|-------------------------|-----|---|-------------|--|----------------------|------------------------------|------|
| Fax To:                    | Houston<br>TX<br>77092-<br>ph: (713) 895-8240<br>Winmar Consulting S<br>James Wiseman | fax: (713) 89<br>Services<br>fax: (713) 89 |                         |     | PO Numbe<br>State:<br>State Cert.<br>Date Repor | <u>No.:</u> | Texas<br>11/9/00   |                      | arrayonaan termin ing tan At |      |
| CII                        | ient Sample ID  | Lab Sample ID                              | Matrix                  | Dat | e Collected                                     |             | Date Received  | coc                  | ID                           | HOLD |
| 2826-D<br>2826-E<br>2826-F |   | 00110147-01<br>00110147-02<br>00110147-03  | Water<br>Water<br>Water |     | 11/3/00<br>11/3/00<br>11/3/00                   | •           | 11/6/00 1:15:00 PM<br>11/6/00 1:15:00 PM<br>11/6/00 1:15:00 PM | 0865<br>0865<br>0865 | 80                           |      |

Neschich, Paul Senior Project Manager

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11/10/00

Date

Joel Grice Laboratory Director

Ted Yen Quality Assurance Officer

11/10/00 12:45.03 PM

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 669-0901

| Client Sample ID 2826-D |           | Col       | lected: 11 | /3/00          | SPL Sample IC | ): 0011 | 0147-01 |
|-------------------------|-----------|-----------|------------|----------------|---------------|---------|---------|
|                         |           | Site      | e: HI 136  |                |               |         | _       |
| Analyses/Method         | Result    | Rep.Limit | Dil        | Factor QUAL    | Date Analyzed | Analyst | Seq. #  |
| OIL & GREASE, TOTAL RE  | COVERABLE |           | MCL        | <b>E</b> 413.1 | Units: mg     | J/L     |         |

3000 2 11/07/00 13:00 G\_T **Qil & Grease**, Total Recoverable 1 464849 Run ID/Seq #: EX\_001107F-464849

Prep Initials Prep Method Prep Date 11/07/2000 13:00

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D. Surrogate Rocovery Unreportable due to Dilution MI - Matrix Interference

11/10/00 12:45:12 PM

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

| Client Sample ID 2826-E        | Colle  | ected:        | 11/3/00 | SPL Sample ID:   | 00110147-02        |              |
|--------------------------------|--------|---------------|---------|------------------|--------------------|--------------|
|                                |        | Site:         | HI1     | 136              |                    |              |
| Analyses/Method                | Result | Rep.Limit     |         | Dil. Factor QUAL | Date Analyzed Ar   | nalyst Seq.# |
| OIL & GREASE, TOTAL RECOVE     | ERABLE |               | MCL     | E413.1           | Units: mg/L        |              |
| Oil & Grease Total Recoverable | 1100   | 2             |         | 1 .              | 11/07/00 13:00 G_1 | г 464850     |
| Run ID/Seq #: EX_001107F-46    | 54850  |               |         |                  |                    |              |
| Prep Method Prep Date          |        | Prep Initials |         |                  |                    |              |

Qualifiers:

ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surregule Recovery Unreportable due to Dilution MI - Matrix Interference

11/10/00 12:49:12 PM

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 560-0901

| Client Sample ID 2826-F         |               | Colle         | cted: | 11/3/00          | SPL Sample ID:     | 00110147-03  |
|---------------------------------|---------------|---------------|-------|------------------|--------------------|--------------|
|                                 |               | Site:         | HI 1  | 36               |                    |              |
| Analyses/Method                 | Result        | Rep.Limit     |       | Dil. Factor QUAL | Date Analyzed Ana  | alyst Seq. # |
| OIL & GREASE, TOTAL RECOV       | ERABLE        |               | MCL   | E413.1           | Units: mg/L        |              |
| Oil & Grease, Total Recoverable | 7.8           | 2             |       | 1                | 11/07/00 13:00 G_T | 464851       |
| Run ID/Seq #: EX_001107F-4      | 64 <b>851</b> |               |       |                  |                    |              |
| Prep Method Prep Date           |               | Prep Initials |       |                  |                    |              |

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Blank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D Surrogale Receivery Unreportable due to Dilution MI - Matrix Interference

11(10/00 12:45:12 PM



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

## Winmar Consulting Services

#### Certificate of Analysis Number:

00110146

| Report To: | Winmar Consulting<br>James Wiseman    | Services                             |         | Project Name   | -                  |        |      |
|------------|---------------------------------------|--------------------------------------|---------|----------------|--------------------|--------|------|
|            | 5700 NW Central Di                    | rive                                 |         | <u>Site:</u>   | HI 136             |        | •    |
|            | Suite 150                             |                                      |         | Site Address:  | _                  |        | •    |
|            | Houston                               |                                      |         | 55 N           |                    |        |      |
|            | TX                                    |                                      |         | PO Number:     |                    |        |      |
|            | 77092-                                | ( <b>1</b> 7 ( <b>1</b> ) - <b>1</b> |         | State:         | Texas              |        |      |
|            | ph: (713) 895-8240                    | fax: (713) 89                        | 95-8270 | State Cert. No | <b>b.</b> :        |        | 1    |
| Fax To:    | Winmar Consulting \$<br>James Wiseman | Services<br>fax: (713) 89            | 95-8270 | Date Reported  |                    |        |      |
| Cli        | ient Sample ID                        | Lab Sampie ID                        | Matrix  | Date Collected | Date Received      | COC ID | HOLD |
| 2826-AA    |                                       | 00110146-01                          | Water   | 11/3/00        | 11/6/00 1:15:00 PM | 086579 |      |
| 2826-B     |                                       | 00110146-02                          | Water   | 11/3/00        | 11/6/00 1:15:00 PM | 086579 |      |
| 2826-C     |                                       | 00110146-03                          | Water   | 11/3/00        | 11/6/00 1:15:00 PM | 086579 |      |
|            |                                       |                                      |         |                |                    |        |      |

Neschich Paul Senior Project Manager

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11/20/00

Date

Joel Grice Laboratory Director

Ted Yen Quality Assurance Officer

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HOUSTON LABORATORY S8E0 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

| ALKALINITY, CARBONATE<br>Alkalinity. Carbonate         ND         2         MCL         M2320 B         Units: mg/L           Alkalinity. Carbonate         ND         2         1         11/06/00 15:35 SN         465520           CHLORIDE, TOTAL<br>Charide         MCL         E325.3         Units: mg/L           Charide         46300         1000         1000         11/16/00 10:45 CV         469447           METALS BY METHOD 6010B, TOTAL<br>Banum         MCL         SW6010B         Units: mg/L         476886           Calcium         1110         2         20         11/16/00 19:45 E, B         476886           Iron         67         0.02         1         11/17/00 18:06 E, B         476886           Megnesium         275         0.1         1         11/16/00 19:24 E, B         476886           Prep Method         Prep Date         Prep Initials         SW0010A         11/17/7200 19:20         476886           Sw0010A         11/07/2000 13:00         R, T         R, T         ND         1         11/15/00 19:24 E, B         476886           Prep Method         Prep Date         Prep Initials         SW0010A         11/07/2000 13:20         R, T           Run ID/Seg #: TJA_001116D-4768893         Prep Initials         SW  | Client Sample ID 282   | 6-AA                                  | Coll   | ected: | 11/3/00          | SPL Sample ID: 0011  | 0146-01 |
|---|--|---------------------------------------|--|--------|------------------|--|---------|
| ALKALINITY, BICARBONATE<br>Alkalinity, Bicarbonatz         333         2         MCL         M2320 B         Units: mg/L           Alkalinity, Carbonatz         333         2         1         11/06/00 15:35 SN         465493           ALKALINITY, CARBONATE<br>Alkalinity, Carbonatz         ND         2         1         11/06/00 15:35 SN         465493           ALKALINITY, CARBONATE<br>Alkalinity, Carbonatz         ND         2         1         11/06/00 15:35 SN         465493           ALKALINITY, CARBONATE<br>Alkalinity, Carbonatz         ND         2         1         11/06/00 15:35 SN         465493           CHLORDE, TOTAL<br>Chloride         ND         2         MCL         E325.3         Units: mg/L           Chloride         48300         1000         11/10/00 10:45 CV         469447           METALS BY METHOD 6010B, TOTAL<br>Barum         MCL         SW6010B         Units: mg/L           METALS BY METHOD 6010B, TOTAL<br>Calcium         MCL         SW6010B         Units: mg/L           Magnesium         275         0.1         1         11/16/00 18:05 E, B         476885           Prep Method         Prep Date         Prep Initials         Vision 11:00 71600 18:05 R, TA         66850           SW3010A         11/07/2000 13:00 R, T         R, T         Ru  |  |                                       | Site   | : HI 1 | 36               |  |         |
| Alkalinity. Bicarbonate         333         2         1         11/06/00 15:35         5N         46549:           ALKALINITY, CARBONATE         ND         2         1         11/06/00 15:35         N         46549:           Alkalinity. Carbonate         ND         2         1         11/06/00 15:35         N         46549:           CHLORIDE, TOTAL         MCL         E325.3         Units: mg/L         11/06/00 15:35         N         46549:           Chloride         48300         1000         1000         1000         11/10/00 10:45         CV         45947:           METALS BY METHOD 6010B, TOTAL         MCL         SW6010B         Units: mg/L         56         476883           Barium         260         0.25         5C         11/1/10/00 19:46         E_5         476883           Galcium         1110         2         20         11/17/10.19:46         E_5         476883           Magnessium         275         0.1         1         11/16/00 19:24         E_8         476683           Prep Method         Prep Date         Prep Initials         Prep Initials         Prep Initials         Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00 </th <th>Analyses/Method</th> <th>Result</th> <th>Rep.Limit</th> <th></th> <th>Dil. Factor QUAL</th> <th>Date Analyzed Analyst</th> <th>Seq. #</th> | Analyses/Method  | Result                                | Rep.Limit  |        | Dil. Factor QUAL | Date Analyzed Analyst  | Seq. #  |
| Alkalinity_Carbonate         ND         2         1         11/06/00 15:35         SN         465520           CHLORIDE, TOTAL<br>Charide         MCL         E325.3         Units: mg/L           Charide         46300         1000         1000         11/10/00 10:45         CV         469443           METALS BY METHOD 6010B, TOTAL<br>Banum         MCL         SW6010B         Units: mg/L         465520           Banum         260         0.25         50         11/16/00 19:36         E_B         476886           Calcium         1110         2         20         11/16/00 19:24         E_B         476886           Iron         67         0.02         1         11/16/00 19:24         E_B         476886           Presp.Method         Presp.Date         Presp.Initials         N338         2         1         11/16/00 19:24         E_B         476886           Presp.Method         Presp.Date         Presp.Initials         N338         2         1         11/16/00 19:24         E_B         476886           Presp.Method         Presp.Date         Presp.Initials         N356 qt.         14.30         11/16/16/16         Units: pt Units           SW3010A         11/107/2000 13:00         R_T         Run  | · · · · ·  |                                       | 2  | MCL    |                  |  | 465495  |
| Chloride         48300         1000         11/10/00         10.45         CV         469447           METALS BY METHOD 6010B, TOTAL<br>Barium         MCL         SW6010B         Units: mg/L           Barium         260         0.25         50         11/16/00         19:45         E_B         476885           Calcium         1110         2         20         11/16/00         19:36         E_B         476885           Calcium         275         0.1         1         11/16/00         19:24         E_B         476885           Potassium         275         0.1         1         11/16/00         19:24         E_B         476885           Run ID/Seq #: TJA_001116D-476885         Prep Initials         SW3010A         I1/107/2000         13:00         R_T           Run ID/Seq #: TJA_001116D-476885         Prep Initials         SW3010A         I1/107/2000         R_T           Run ID/Seq #: TJA_001116D-476885         Prep Initials         SW3010A         I1/107/2000         R_T           Run ID/Seq #: TJA_001116D-4768893         Prep Initials         SW3010A         I1/107/2000         R_T           pH         6.6         0.00100         1         11/06/00         10.000         E_E   | •  |                                       | 2  | MCL    | M2320 B          | · • • •  | 465520  |
| Barium         260         0.25         50         11/16/00 19:46         E_B         476885           Calcium         1110         2         20         11/16/00 19:30         E_B         476885           Iron         57         0.02         1         11/17/00 18:06         E_B         476885           Potassium         275         0.1         1         11/17/00 19:24         E_B         476885           Prep Method         Prep Date         Prep Initials         50/3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials         SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-476893         Prep Initials         SW3010A         11/17/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials         SW3010A         11/17/2000 12:45         R_T           PH         6.6         0.10         1         11/06/00 17:00         EC         485805           SW3010A         11/17/472000 12:45         R_T         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00         EC         485805           SW3010A         11/17/7200  |  | 48300                                 | 1000   | MCL    |                  | ×  | 469447  |
| Calcium         1110         2         20         11/16/00         19:30         E,B         476886           Iron         67         0.02         1         11/17/00         18:06         E_B         476896           Magnesium         275         0.1         1         11/17/00         18:06         E_B         476896           Potassium         338         2         1         11/16/00         19:24         E_B         476886           Run ID/Seq #: TJA_001116D-476865         Prep Date         Prep Initials         476886         476886           Prep Method         Prep Date         Prep Initials         476886         476886           SW3010A         11/07/2000         13:00         R_T         Run ID/Seq #: TJA_001117A-478593         47692           Prep Method         Prep Date         Prep Initials         SW3010A         11/17/2000         12:45         R_T           PH         6.6         0.10  | METALS BY METHOD   | 6010B, TOTAL                          | •  | MCL    | SW6010B          | Units: mg/L  |         |
| Iron       67       0.02       1       11/17/00 18:06 E_B       476502         Magnesium       275       0.1       1       11/16/00 19:24 E_B       476802         Potassium       338       2       1       11/16/00 19:24 E_B       476802         Potassium       338       2       1       11/16/00 19:24 E_B       476802         Prep. Method       Prep. Date       Prep. Initials       5       476802         SW3010A       11/07/2000 13:00       R_T       R       7         Run ID/Seq #: TJA_001116D-476886       Prep. Initials       5       8       7         SW3010A       11/07/2000 13:00       R_T       R       7         Run ID/Seq #: TJA_001116D-476886       Prep. Initials       5       8       7         SW3010A       11/07/2000 13:00       R_T       R       7         Run ID/Seq #: TJA_001116D-476888       Prep. Initials       5       5         SW3010A       11/07/2000 12:45       R_T       7         PH       6.6       0.10       1       11/06/00 17:00 EC       468450         SPECIFIC GRAVITY       Specific Gravity       1.045       0       1       11/09/00 18:20 JS       468450         SUI  | Barium   | 260                                   | 0.25   |        | 50               | 11/16/00 19:46 E_B   | 476889  |
| Magnesium         275         0.1         1         1/16/00 19:24         E_B         476885           Potassium         338         2         1         11/16/00 19:24         E_B         476885           Prep Method         Prep Date         Prep Initials         1         11/16/00 19:24         E_B         476885           Prep Method         Prep Date         Prep Initials         1         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials         SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476889         Prep Initials         SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials         SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1         11/06/00 17:00         EC         4685805           RESISTANCE @ 25 C         MCL         120.1         Units: PH Units         468450           SPECIFIC GRAVITY         Specific Gravity         1.045         0         1         11/09/00 18:20         JS         468450           SULFATE, TOTAL         ND         1         1         11/06/00 16:00         EC         478055  | Calcium  | 111D                                  | 2  |        | 20               | 11/16/00 19:30 E_B   | 476886  |
| Potassium         338         2         1         11/16/00 19:24 E_B         476885           Prep. Method         Prep. Date         Prep. Initials           476885           SW3010A         11/07/2000 13:00         R_T               Run ID/Seq #: TJA_001116D-476886         Prep. Method         Prep. Date         Prep. Initials              SW3010A         11/07/2000 13:00         R_T   <  | Iron   | 67                                    | 0.02   |        | 1                | 11/17/00 18:06 E_B   | 478593  |
| Run ID/Seq #: TJA_001116D-476885           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476889         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1           pH         6.6         0.10         1         11/06/00 17:00 EC         468450           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           Specific Gravity         1.045         0         1         11/09/00 18:20 JS         468450           SULFATE, TOTAL         MCL         ASTM D-1429         Units: Specific Gravity @         11/14/00 16:00 EC         478053           Sulfate         ND   | Magnesium  | 275                                   | 0.1  |        | 1                | 11/16/00 19:24 E_B   | 476885  |
| Run ID/Seq #: TJA_001116D-476885           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476888         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476889         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-476593         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1           pH         6.6         0.10         1           Resistance         6         0.00100         1         11/09/00 18:20 JS           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         1         11/06/00 16:00 SN         463672           Sulfate  | Potassium  | 338                                   | 2  |        | 1                | 11/16/00 19:24 E B   | 476885  |
| Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476888         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476888         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-476893         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-476893         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1           pH         6.6         0.10         1           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20         JS           Specific Gravity         1.045         0         1         11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         1         11/06/00 16:00 SN         463672  |  | A_001116D-476885                      |  |        |                  | -  | •       |
| SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476886         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-476889         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1           pH         6.6         0.10         1           resistance         6         0.00100         1         11/09/00 18:20         JS           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/05/00 16:00         EC         478053           SUEFATE, TOTAL         MCL         E375.4         Units: mg/L         463672           Sulfate         ND         1         1         11/06/00 16:00         SN         463672   |  |                                       | Prep Initials  |        |                  |  |         |
| Run ID/Seg #: TJA_001116D-476886           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seg #: TJA_001116D-476869         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seg #: TJA_0011176-478693         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         6.6         0.10         1           pH         6.6         0.10         1           Resistance         6         0.00100         1         11/09/00 18:20         JS           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/09/00 16:00         EC         478053           SULFATE, TOTAL         MCL         E375.4         Units: mg/L         1         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L         1   | and the second s | 11/07/2000 13:00                      |  |        |                  |  |         |
| Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001116D-4763893         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Date         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         Prep Date         Prep Initials           SW3010A         11/17/2000 12:45         R_T           PH         6.6         0.10           pH         6.6         0.10           Resistance         6         0.00100           SPECIFIC GRAVITY         MCL         ASTM D-1429           Specific Gravity         1.045         0           SULFATE, TOTAL         MCL         E375.4         Units: mg/L           Sulfate         ND         1         11/05/00 16:00         SN           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | Run ID/Seg #: TJ   | A_001116D-476886                      |  |        |                  |  |         |
| Run ID/Seq #: TJA_001116D-476859           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00         EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm         Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @         30000         1         11/09/00 18:20         JS         468450           SULFATE, TOTAL         MCL         ASTM D-1429         Units: Specific Gravity @         30000         1         11/09/00 16:00         EC         478053           SULFATE, TOTAL         ND         1         11/06/00 16:00         SN         463672           Sulfate         ND         1         1         11/06/00 16:00         SN         463672   | Prep Method  | Prep Date                             | Prep Initials  |        |                  |  |         |
| Run ID/Seq #: TJA_001116D-476859           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Initials           Prep Method         Prep Date         Prep Initials           SW3010A         11/07/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00         EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm         Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @         30000         1         11/09/00 18:20         JS         468450           SULFATE, TOTAL         MCL         ASTM D-1429         Units: Specific Gravity @         30000         1         11/09/00 16:00         EC         478053           SULFATE, TOTAL         ND         1         11/06/00 16:00         SN         463672           Sulfate         ND         1         1         11/06/00 16:00         SN         463672   | SW3010A  | 11/07/2000 13:00                      | R_T  |        |                  |  |         |
| SW3010A         11/07/2000 13:00         R_T           Run ID/Seq #: TJA_001117A-478593         Prep Date         Prep Initials           SW3010A         11/17/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00 EC         485805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @         10/11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         11/06/00 16:00 SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | Run ID/Seq #: TJ/  | A_001116D-476889                      | • -  |        |                  |  |         |
| Run ID/Seq #: TJA_001117A-478593           Prep Method         Prep Date         Prep Initials           SW3010A         11/17/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00 EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @         11/11/4/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         1         11/06/00 16:00 SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | Prep Method  | Prep Date                             | Prep Initials  |        |                  |  |         |
| Prep Method         Prep Date         Prep Initials           SW3010A         11/17/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00 EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm         668450           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @         11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         11/06/00 16:00 SN         463672           Sulfate         ND         1         11/06/00 16:00 SN         463672           TOTAL DISSOLVED SOLIDS         MCL TDS-MINERAL         Units: mg/L  | SW3010A  | 11/07/2000 13:00                      | R_T  |        |                  |  |         |
| SW3010A         11/17/2000 12:45         R_T           PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00 EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         1         11/06/00 16:00 SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | Run ID/Seq #: TJ/  | A_001117A-478593                      |  |        |                  |  |         |
| PH         MCL         E150.1         Units: pH Units           pH         6.6         0.10         1         11/06/00 17:00 EC         465805           RESISTANCE @ 25 C         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20 JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/14/00 16:00 EC         478053           SULFATE, TOTAL         ND         1         1         11/05/00 16:00 SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L   | Prep Method  | Prep Date                             | Prep Initials  |        |                  |  |         |
| pH         6.6         0.10         1         11/06/00 17:00         EC         465805           RESISTANCE @ 25 C<br>Resistance         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY<br>Specific Gravity         MCL         ASTM D-1429         Units: Specific Gravity @<br>11/14/00 16:00         EC         478053           SULFATE, TOTAL<br>Sulfate         ND         1         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L         Units: mg/L  | 5W3010A  | 11/17/2000 12:45                      | R_T  |        |                  |  |         |
| pH         6.6         0.10         1         11/06/00 17:00         EC         465805           RESISTANCE @ 25 C<br>Resistance         MCL         120.1         Units: Mohms/cm           Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY<br>Specific Gravity         MCL         ASTM D-1429         Units: Specific Gravity @<br>11/14/00 16:00         EC         478053           SULFATE, TOTAL<br>Sulfate         ND         1         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L         Units: mg/L  |  |                                       | · · · · · ·  | MOL    | E460.4           | linita, ali Unita  |         |
| Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/14/00 16:00         EC         478053           SULFATE, TOTAL         MCL         E375.4         Units: mg/L           Sulfate         ND         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L   |  | 6.6                                   | 0.10   | WOL    |                  | The second s | 465805  |
| Resistance         6         0.00100         1         11/09/00 18:20         JS         468450           SPECIFIC GRAVITY         MCL         ASTM D-1429         Units: Specific Gravity @           Specific Gravity         1.045         0         1         11/14/00 16:00         EC         478053           SULFATE, TOTAL         MCL         E375.4         Units: mg/L           Sulfate         ND         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L   | RESISTANCE @ 25 C  | • • • • • • • • • • • • • •           | · · ·  | MCL    | 120.1            | Units: Mohms/cm  | ••      |
| Specific Gravity         1.045         0         1         11/14/00 16:00         EC         478053           SULFATE, TOTAL         MCL         E375.4         Units: mg/L           Sulfate         ND         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | <b>7</b>   |                                       | 0.00100  |        |                  |  | 468450  |
| Specific Gravity         1.045         0         1         11/14/00 16:00         EC         478053           SULFATE, TOTAL         MCL         E375.4         Units: mg/L           Sulfate         ND         1         11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | SPECIFIC GRAVITY   |                                       | na ku<br>Na ku na br>Na ku na ku | MCL    | ASTM D-1429      | Units: Specific Gra  | vity @  |
| Sulfate         ND         1         1 11/06/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  |  | 1.045                                 | 0  |        |                  |  | 478053  |
| Sulfate         ND         1         1         11/05/00 16:00         SN         463672           TOTAL DISSOLVED SOLIDS         MCL         TDS-MINERAL         Units: mg/L  | SULFATE, TOTAL   | • • • • • • • • • • • • • • • • • • • |  | MCL    | E375.4           | Units: mg/L  |         |
|   |  | ND                                    | 1  |        | 1                | -  | 463672  |
| Total Dissolved Solids, Calculated 80000 10 1 11/20/00 15:30 ES 47957 1   | TOTAL DISSOLVED S  | OLIDS                                 |  | MCL    | TDS-MINERAL      | Units: mg/L  | *       |
|   | Total Dissolved Solids,  | Calculated 80000                      | 10   |        | 1                | 11/20/00 15:30 ES  | 479571  |

Qualifiers:

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ND/U - No! Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D • Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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COMPRESS OF



HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

| Client Sample ID 2826-AA                      |        | Col       | ected: | 11/3/00          | SPL Sample ID: 00110  | 0146-01 |
|---|--------|-----------|--------|------------------|-----------------------|---------|
|   |        | Site      | : HI1  | 136              |                       |         |
| Analyses/Method                               | Result | Rep.Limit |        | Dil. Factor QUAL | Date Analyzed Analyst | Seq. #  |
| TOTAL SODIUM, CALCULATED                      |        |           | MCL    | TDS-MINERAL      | Units: mg/L           | "       |
| Total Sodium, Calculated                      | 29300  | 10        |        | 1                | 11/20/00 15:30 ES     | 479548  |
| TOTAL SUSPENDED SOLIDS                        |        | •• •••    | MCL    | E160.2           | Units: mg/L           |         |
| Suspended Solids (Residue.Non-<br>Filterable) | 340    | 8         |        | 2                | 11/09/00 12:30 EC     | 468153  |

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

- B Analyte detected in the associated Method Hlank
- \* Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Suriogate Recovery Unreportable due to Dilution MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054

(713) 660-0901

| Client Sample ID 282                          | 6-B                                      | Coll           | ected:       | 11/3/00             | SPL Sample ID: 0                 | 0110146-02     |
|---|--|----------------|--------------|---------------------|----------------------------------|----------------|
|   |  | Site           | HI1          | 136                 |                                  |                |
| Analyses/Method                               | Result                                   | Rep.Limit      |              | Dil. Factor QUAL    | Date Analyzed Analy              | /st Seq. #     |
| ALKALINITY, BICARE<br>Alkalinity, Bicarbonate | BONATE<br>313                            | 2              | MCL          | <b>M2320 B</b><br>1 | Units: mg/L<br>11/06/00 15:35 SN | 465496         |
| ALKALINITY, CARBO<br>Alkalinity, Carbonate    | NATE                                     | 2              | MCL          | M2320 B<br>1        | Units: mg/L<br>11/06/00 15:35 SN | 465521         |
| CHLORIDE, TOTAL<br>Chloride                   | 45000                                    | 1000           | MCL          | <b>E325.3</b> 1000  | Units: mg/L<br>11/10/00 10:45 CV | 469448         |
| METALS BY METHOD                              |  |                | MCL          | SW6010B             | Units: mg/L                      |                |
| Barium  | 207                                      | 0.025          |              | 5                   | 11/16/00 19:57 E_B               | 476891         |
| Calcium                                       | 1190                                     | 0.5            |              | 5                   | 11/16/00 19:57 E_B               | 476891         |
| Iron  | 117                                      | 0.02           |              | 1                   | 11/17/00 18:10 E_B               | 478594         |
| Magnesium                                     | 298                                      | 0.1            |              | 1                   | 11/16/00 19:50 E_B               | 476890         |
| Potassium                                     | 386                                      | 2              |              | 1                   | 11/16/00 19:50 E_B               | 476890         |
| Run ID/Seq #: TJ/                             | A_001116D-476890                         |                |              |                     |                                  |                |
| Prep Method                                   | Prep Date                                | Prep Initials  |              |                     |                                  |                |
| SW3010A                                       | 11/07/2000 13:00                         | R_T            |              |                     |                                  |                |
| Run ID/Seq #: TJ/                             | 4_001116D-476891                         |                |              |                     |                                  |                |
| Prep Method                                   | Prep Date                                | Prep Initials  |              |                     |                                  |                |
| SW3010A                                       | 11/07/2000 13:00                         | R_T            |              |                     |                                  |                |
|   | A_001117A-478594                         | ····           |              |                     |                                  |                |
| Prep Method                                   | Prep Date                                | Prep Initials  |              |                     |                                  |                |
| SW3010A                                       | 11/17/2000 12:45                         | IR_T           |              |                     |                                  |                |
| PH  | · · · · · · · · · · · · · · · · · · ·    |                | MCL          | E150.1              | Units: pH Units                  | 1              |
| pH  | 7.8                                      | 0.10           |              | 1                   | 11/06/00 17:00 EC                | 435809         |
|   | n an |                | · <u>·</u> . |                     | •                                | •••            |
| RESISTANCE @ 25 C                             | <b>C</b> 4                               | 0.00400        | MCL          | 120.1               | Units: Mohms/                    |                |
| Resistance                                    | 6.4                                      | 0.00100        |              | 1                   | 11/09/00 18:20 JS                | 468452         |
| SPECIFIC GRAVITY                              | • • • • •                                | • •·           | MCL          | ASTM D-1429         | Units: Specific                  | Gravity @      |
| Specific Gravity                              | 1.054                                    | 0              |              | 1                   | 11/14/00 16:00 EC                | 478055         |
| SULFATE, TOTAL<br>Sulfate                     | ND                                       | 1<br>1         | MCL          | E375.4              | Units: mg/L<br>13/06/00 16:00 SN | 463673         |
| TOTAL DISSOLVED S                             |  | 10 <u>- 10</u> | MCL          | TDS-MINERAL         | Units: mg/L<br>11/20/00 15:30 ES | <b>47</b> 9572 |
| i ulai pissoiveu ouilus,                      |  | IU             |              | ۱                   | 11/20/00 10:30 ES                | 4/30/2         |
| TOTAL SODIUM, CAL<br>Total Sodium, Calculate  |  | - 10           | MCL          | TDS-MINERAL         | Units: mg/L<br>11/20/00 15:30 ES | 479549         |

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 660-0901

| Client Sample ID 2826-B                       |        | Col       | lected: | 11/3/00          | SPL Sample ID:    | 00110146-02             |
|---|--------|-----------|---------|------------------|-------------------|-------------------------|
|   |        | Site      | e: HI1  | 136              |                   |                         |
| Analyses/Method                               | Result | Rep.Limit |         | Dil. Factor QUAL | Date Analyzed A   | nalyst Seq. #           |
| TOTAL SUSPENDED SOLIDS                        |        |           | MCL     | E160.2           | Units: mg/L       | ана <b>Т</b> арана<br>1 |
| Suspended Solids (Residue,Non-<br>Filterable) | 728    | 8         |         | 2                | 11/09/00 12:30 EC | 468154                  |

Qualifiers:

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ND/U - Not Detected at the Reporting Limit

5 - Analyte detected in the associated Method Blank

\*- Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 560-0801

| Client Sample ID 2826                           | -C                            | Colle                | acted: | 11/3/00                  | SPL Sample ID: 00                        | 110146-03                      |
|---|-------------------------------|----------------------|--------|--------------------------|--|--------------------------------|
|   |                               | Site                 | HL 1   | 136                      |  |                                |
| Analyses/Method                                 | Result                        | Rep.Limit            |        | DII. Factor QUAL         | Date Analyzed Analys                     | st Seq. #                      |
| ALKALINITY, BICARBO<br>Alkalinity, Bicaroonate  | <b>DNATE</b><br>152           | 2                    | MCL    | <b>M2320 B</b><br>1      | Units: mg/L<br>11/06/00 15:35 SN         | 465498                         |
| ALKALINITY, CARBON<br>Alkalinity, Carbonate     | ND                            | 2                    | MCL    | M2320 B                  | Units: mg/L<br>11/06/00 15:35 SN         | 46552:                         |
| CHLORIDE, TOTAL<br>Chloride                     | 16700                         | 250                  | MCL    | <b>E325.3</b><br>250     | Units: mg/L<br>11/10/00 10:45 CV         | 469450                         |
| METALS BY METHOD                                | 6010B, TOTAL<br>0.181         | 0.005                | MCL    | SW6010B                  | Units: mg/L<br>11/16/00 20:19 E B        | 476892                         |
| Calcium   | 406<br>2.15                   | 0.1<br>0.02          |        | •                        | 11/16/00 20:19 E_B<br>11/17/00 17:29 E B | 476892                         |
| Magnesium<br>Potassium                          | 1260<br>4 <b>2</b> 5          | 0.5<br>10            |        | 5                        | 11/16/00 20:23 E_B<br>11/16/00 20:23 E B | 476893<br>476893               |
| Run ID/Seq #: TJA<br>Prep Method                | _001116D-476892<br>Prep Date  | Prep Initials        |        |                          |  |                                |
| Run ID/Seq #: TJA                               |                               | R_T                  |        |                          |  |                                |
| Prep Method<br>SW3010A                          | Prep Date<br>11/07/2000 13:00 | Prep Initials        |        |                          |  |                                |
| Run ID/Seq #: TJA<br>Prep Method<br>SW3010A     | Prep Date<br>11/17/2000 12:45 | Prep Initials<br>R_T |        |                          |  |                                |
| PH<br>pH  | 8.2                           | 0.10                 | MCL    | E150.1                   | Units: pH Units<br>11/06/00 17:00 EC     | 45581                          |
| RESISTANCE @ 25 C<br>Resistance                 | 18.3                          | 0.00100              | MCL    | 120.1                    | Units: Mohms/c<br>11/09/00 18:20 JS      | <b>m</b><br>4 <del>5</del> 845 |
| SPECIFIC GRAVITY<br>Specific Gravity            | 1.018                         | · <u>innen</u>       | MCL    | ASTM D-1429<br>1         | Units: Specific (<br>11/14/00 16:00 EC   | Gravity @<br>478056            |
| SULFATE, TOTAL<br>Sulfate                       | 6500                          | 500                  | MCL    | <b>E375.4</b><br>500     | Units: mg/L<br>11/06/00 16:00 SN         | 463674                         |
| TOTAL DISSOLVED SO<br>Total Dissolved Solids, C |                               | 10                   | MCL    | TD <b>S-MINERAL</b><br>1 | Units: mg/L<br>11/20/00 15:30 ES         | 47957                          |
| TOTAL SODIUM, CALC<br>Total Sodium, Calculated  | · · ·                         | 10                   | MCL    | TDS-MINERAL              | Units: mg/L<br>11/20/00 15:30 ES         | 47955(                         |

Qualifiers;

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MC\_) D - Surrogate Recovery Unreportable due to Dilution MI - Matrix Interference

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HOUSTON LABORATORY 8880 INTERCHANGE DRIVE HOUSTON, TEXAS 77054 (713) 550-0901

| Client Sample ID 2826-C                                  |        | Coll      | ected: | 11/3/00          | SPL Sample ID:  | 00110     | 146-03 |
|--|--------|-----------|--------|------------------|-----------------|-----------|--------|
|  |        | Site      | : HI1  | 36               |                 |           |        |
| Analyses/Method  | Result | Rep.Limlt |        | Dil. Factor QUAL | Date Analyzed A | naiyst    | Seq. # |
| TOTAL SUSPENDED SOLIDS<br>Suspended Solids (Residue Non- | 32     | 4         | MCL    | E160.2           | Units: mg/L     | • • • • • | 468155 |
| Filterable)  |        |           |        |                  |                 | -         |        |

Qualifiers:

ND/U - Not Detected at the Reporting Limit

B - Analyte detected in the associated Method Blank

\* - Surrogate Recovery Outside Advisable QC Limits

J - Estimated Value between MDL and PQL

>MCL - Result Over Maximum Contamination Limit(MCL)
 D - Surrogate Recovery Unreportable due to Dilution
 MI - Matrix Interference

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## 10. Results and Observations 11513

Gas and water samples were taken from this line. However, these samples were not tested because the line contained approximately 100 barrels of what appeared to be drill mud. Because it was impossible to get uncontaminated samples from throughout the line, it was decided to halt sampling and discard this data.



## 11. Recommendation/Conclusions

All of the pipelines tested for this project were installed in 1964. Based on the Results and Observations for the four pipeline segments where complete data was obtained -WINMAR was able to qualitatively rank the pipeline conditions. This is shown in Table 1 below. The pipelines were ranked according to the criteria listed in the table header: presence of pits, metal loss, pooled water, flushwater iron concentration, and weld erosion.

| Segment | Rank | Pits    | Metal Loss | Pooled Water | Highest Iron  | Weld    |
|---------|------|---------|------------|--------------|---------------|---------|
|         |      | Present |            |              | Concentration | Erosion |
|         |      |         |            |              |               |         |
| 2826    | #1   | No      | No         | No           | 117 PPM       | No      |
| 2824    | #2   | No      | Yes        | Yes          | 302 PPM       | Yes     |
| 2820    | #3   | Yes     | No         | No           | 70 PPM        | Yes     |
| 2822    | #4   | Yes     | Yes        | Yes          | 76 PPM        | No      |

Table 1 – Pipeline Ranking

The conclusions and recommendations in this section are based on the Results and Observations from sections 6.0 through 10.0. Each conclusion will be presented, then followed by the relevant photos or results for that conclusion.

**Conclusion 1)** Having the unused pipeline open to air versus sealed doesn't seem to have an impact on the line condition.

This is based on analysis of segments 2824 and 2826. **Figure 23** shows that segment 2824 was sealed and remained filled predominantly with methane. **Figure 31** shows that segment 2826 was predominantly filled with atmospheric air (containing oxygen, whereas the methane filled line did not contain much oxygen). Since the air is heavier than methane, it was present in the bottom of the pipeline. These two gas pipelines were in very similar condition however, despite being filled with different fluids. Segment 2824 showed metal loss corrosion but this was in the bottom of the pipeline - which contained standing water. The "dry" portions of the lines were in very much the same condition, despite the presence of oxygen in the air-filled line.

**Conclusion 2)** Standing water from wellstream production pools in the pipeline and causes metal loss corrosion. The standing water also provides a medium for the growth of sulfide reducing bacteria.



This standing water seemed to be the primary cause of loss of integrity for the pipelines tested. Segment 2824 showed evidence of standing water (as seen in the following photos). Since this line had never been pigged (non-pigabble line) the water present in the pipeline must have come from wellstream production. The sample taken near the well-protector platform showed evidence of being filled approximately 15% with water. Depending on the pipeline elevation (high and low spots) along its length, it could have contained either more, or less water. This is demonstrated in the following pictures (from segment 2824):



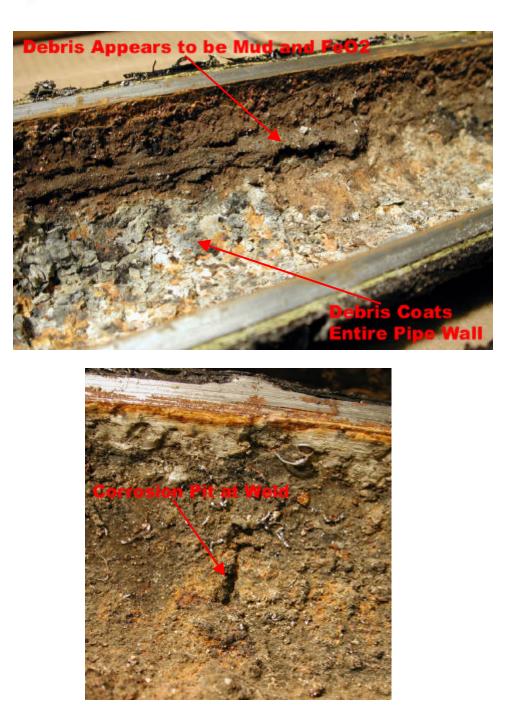




Conclusion 3) Composition of pipeline flush and fill water is important.

If possible, the operator should endeavor to NOT suck up any mud or particulate matter from the ocean, when filling a pipeline with seawater. This mud/sludge contains a "soup" of bacteria in much higher concentrations than found in surface seawater. These bacteria can and will contribute to Microbial Induced Corrosion (MIC). Segment 2822 shows evidence of being filled with seawater with a very high amount of suspended solids. It also shows signs of pitting corrosion at the pipeline/tubeturn weld. This is shown in the following photos from the segment 2822 pipeline sample:





**Conclusion 4)** Pitting corrosion is highly variable and unpredictable. Pits are present in some of the pipelines tested and not in others - despite the exact same production. Pits were present in the pipeline filled with water (2822), and also in one pipeline that was not (segment 2820). Based on this information, and this small sample size, it is necessary to conclude that we cannot correlate pitting corrosion to pipeline conditions



for these tests. It may be present to correlate the two after future flushing tests yield a larger sample size.

**Recommendation)** Because pooled water in out of service gas pipelines seems to be the primary factor in loss of integrity, any measure that can remove this water from the line should improve its condition. If a well or wells are "playing out" and the wells will be taken offline in the near future, WINMAR recommends that the operator examine the watercut of the gas. If the production is low volume (equaling a low fluid velocity in the pipeline) and shows a high water content, then pooled water may be present in the pipeline. One way to remove this water would be to temporarily close the well, bleed down the pipeline, disconnect the pipeline. The well can then be brought back online in order to run the pig, and then shut-in again when the operator wishes to temporarily abandon the well and flowline. This dewatering method would be the least expensive and most effective way to protect a non-piggable line that an operator wishes to take out of service, but not fill with inhibited seawater.



## 12. Appendix - Corrosion Inhibitor Information

Many different kinds of inhibitors are available, each serving its own different function. The three most common are:

- Oxygen Scavenger (Uses sulfite to bind oxygen SO2 → SO4)
- Corrosion Inhibitor (Amine coating "seals" internal pipe wall)
- Biocide (Kills bacteria that cause corrosion (Sulfide Reducing Bacteria (SRB's))

According to the vendors and contractors polled, Oxygen scavenger is not always necessary. For closed lines, oxygen will be depleted quickly, and once it is all used, that type of corrosion ceases. Large new lines can be designed for this very small amount of corrosion.

Biocide is the most important inhibitor for out of service lines because SRB's can sit in an out of service line and cause pits. The SRB's use the sulfate in seawater as a respiration source, making sulfuric acid, which causes pitting. In an out of service line, these bacteria have a perfect environment (Moist/Wet, oxygen poor, abundant sulfate source, etc.)

Information sheets were gathered from Champion Technologies and Baker Petrolite. These are included in this Appendix as reference material.

Baker Petrolite's Oxygen Depletion graph/information differs from ours. They show oxygen depletion versus time for a pipeline that is filled with uninhibited seawater and closed. This is interesting information that will be relevant for the Flushing Phase B project.



# Protection of Pipelines During Hydrostatic Testing





#### **INTRODUCTION**

Before a new or rehabilitated pipeline is placed into service, it must be tested for integrity at a pressure above its designed working pressure. This is usually done with water, which may remain in the system for an extended period of time.

Water used in hydrostatic testing usually comes from one of several sources: aquifers, rivers, ponds, seas, etc. The use of water from any of these sources can cause corrosion and introduce bacteria into the pipeline. The severity of the problem is dependent upon the type and quality of water used, the length of time the water remains in the line, and the ambient temperature.

While the line is filled with water it is subjected to three types of corrosion:

- 1. Direct reaction of dissolved oxygen with the steel pipe to form ferric oxide/hydroxide. Pitting may be initiated. This mechanism is not generally serious because the concentration of dissolved oxygen in the water is rapidly depleted due to the reaction with the pipe wall. Our tests indicate that corrosion due to oxygen content, even with air-saturated waters, is usually minimal in a closed steel pipeline and problems rarely result from this mechanism.
- 2. Localized pitting and corrosion resulting from the growth of sulfate reducing bacteria (SRB) and acid producing bacteria (APB).
- 3. Attack by hydrogen sulfide produced as a result of SRB growth.

Mechanisms 2 and 3, which involve bacterial growth, are the most serious concerns in hydrotest waters. Sea water and high TDS brines have a greater potential for corrosion than fresh water due to their higher conductivity and sulfate levels.

Conventional wisdom has it that to protect against corrosion during hydrostatic testing, you must add three types of chemicals to the water: an oxygen scavenger, a biocide and a corrosion inhibitor. For large or long pipelines, this can be exceedingly expensive.

Baker Petrolite Corporation research data indicates that much of this expense is not necessary. In a closed system, oxygen is exhausted long before pitting due to oxygen becomes a problem. Controlling bacterial growth is generally sufficient to protect a pipeline from hydrotest

damage. This can be accomplished by adding a biocide to the water prior to introducing it into the pipeline.

#### CHEMICAL TREATMENT RECOMMENDATIONS FOR HYDROSTATIC TEST WATERS

The following recommendations apply to both fresh water and sea water.

**Biocide:** X-CIDE<sup>®</sup> 102 is recommended for hydrostatic test waters. It should be used at a concentration of 250 ppm to 1000 ppm based on the total volume of water. Biocides are always recommended for hydrostatic test water unless chlorinated water (from a city water supply) is used.

**Corrosion Inhibitor:** If a corrosion inhibitor is desired, CRW 201 is recommended at a concentration of 100 ppm to 200 ppm based on the total volume of water.





**Oxygen Scavenger:** Although Baker Petrolite laboratory experiments and field experience indicate that corrosion problems due to the oxygen content of hydrostatic test waters rarely occur, an oxygen scavenger is sometimes requested as additional protection. In such a case, Baker Petrolite OSW 490C oxygen scavenger is recommended for removal of dissolved oxygen. Recommended dosage is 11 ppm OSW 490C oxygen scavenger for each ppm oxygen in the hydrotest water. Fresh water at 68 degrees F may contain up to 9 ppm dissolved oxygen.

#### PRODUCT APPLICATION

Before pumping the hydrostatic test water into the pipeline, a specific treatment regime should be followed to avoid interactions between products. The oxygen scavenger will deactivate the biocide, so they should not be mixed. The following is a recommended procedure for treating and mixing the water.

- 1. If oxygen removal is desired, measure the amount of dissolved oxygen in the water to be treated. Determine the amount of oxygen scavenger needed (11ppm OSW 490C oxygen scavenger per ppm oxygen in the water).
- 2. To remove oxygen, add the required amount of OSW 490C oxygen scavenger to the water tank. Mix gently; do not overmix; avoid introducing extra air into the tank. Allow approximately 15 minutes for complete scavenging. Since X-CIDE 102 biocide will interact with the oxygen scavenger, it is important to allow the recommended scavenging time to avoid biocide deactivation.
- 3. Add the required amount of X-CIDE 102 biocide to the tank and mix gently.
- 4. A corrosion inhibitor, CRW 201 can then be added to the hydrotest water if desired.

Steps 3 and 4 may be interchanged as the X-CIDE 102 biocide and CRW 201 are fully compatible.

#### FLUID DISPOSAL

HYDROSTATIC TEST FLUIDS CONTAINING RESIDUAL LEVELS OF BIOCIDE, AND/OR CORROSION INHIBITOR, SHOULD BE DISPOSED OF IN ACCORDANCE WITH PERTINENT STATE AND FEDERAL REGULATIONS.

The two most commonly practiced methods of disposal for hydrostatic test waters are direct discharge to receiving waters or discharge to a wastewater treatment plant. When test fluids are discharged directly to a receiving water, caution should be exercised to ensure that the level of residual biocide is below the threshold level which is capable of producing toxic effects in aquatic organisms. The hydrostatic test water can also be disposed in any salt water disposal well which is classified to handle oilfield waste.

Hydrostatic test fluids containing X-CIDE 102 biocide may be detoxified prior to their release to surface waters. Based on the residual level of biocide, a 1:1 ratio of OSW 490C oxygen scavenger should be used. An in-line mixer or surge tank should be used to promote mixing of the detoxifying agent with the hydrostatic test water. A 30-second contact time is sufficient for detoxification to take place. If the discharge from a hydrostatic test displaces a substantial percentage of the receiving water (e.g., a stream or a small bay), then the discharge should be re-aerated to avoid a fish kill due to the lack of oxygen.





Hydrostatic test fluids containing X-CIDE 102 biocide may also be discharged to a wastewater treatment plant. Aerobic bacteria are capable of utilizing X-CIDE 102 biocide as a nutrient source at concentrations of 25 ppm or less. Studies have shown that X-CIDE 102 biocide has an affinity for any type of proteinaceous material and will bind to it irreversibly. Bound X-CIDE 102 biocide is also readily biodegraded.

The results of aquatic toxicity tests carried out with X-CIDE 102 biocide and CRW 201 corrosion inhibitor are in the attached EcoTox<sup>™</sup> reports.

Standard BOD/COD tests were performed with each product. Results of the studies indicate that both the biocide and corrosion inhibitor are readily biodegraded.

The octanol/water partition coefficient for X-CIDE 102 biocide indicates that this chemical has little propensity to bioconcentrate in the environment.

#### PRODUCT EVALUATION

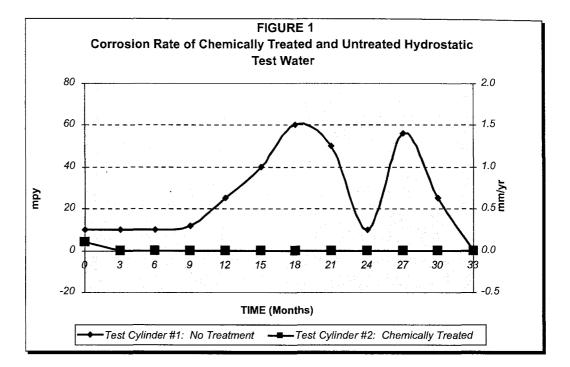
A series of tests were conducted to (a) assess the need for chemical inhibition of hydrostatic test waters and (b) identify and evaluate the most effective program having the widest applicability.

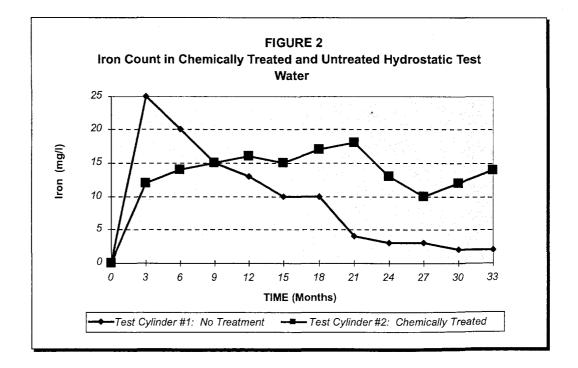
- 1. Long term field evaluations were carried out in conjunction with a major oil company. These tests consisted of periodic monitoring of test cylinders made from sealed pipeline lengths. One test cylinder contained uninhibited sea water, while the sea water in the other cylinder was treated with an oxygen scavenger, a biocide, and a corrosion inhibitor. Monitoring consisted of LPR readings (instantaneous corrosion rate), soluble iron concentration measurements, and SRB enumeration (by the API RP 38 method), each taken periodically over 33 months.
- 2. The rate of oxygen depletion in air saturated waters was measured in test cylinders made from sealed pipeline sections of various diameters containing fresh water and sea water. In addition, the effect of the reaction of oxygen with the pipe wall was assessed over a period of time.
- 3. The effectiveness of selected chemicals was assessed in laboratory studies over an extended period of time in both fresh water and sea water.

The results of monitoring corrosion rate, iron concentration, and number of SRBs in the pipeline test cylinders are shown in Figures 1, 2, and 3. These outdoor tests were carried out in a temperate climate where the cylinders were subjected to ambient temperature fluctuations. Test cylinder 1 contained untreated sea water, whereas the sea water in cylinder 2 was treated with an oxygen scavenger, biocide, and corrosion inhibitor. Figure 1 shows a plot of instantaneous corrosion rate against time. The corrosion rate in cylinder 1 fluctuates widely between 8 mpy (0.2 mm/yr) and 59 mpy (1.5 mm/yr), the peaks occurring in the summer months when the ambient temperatures are the highest. The treated cylinder showed very low corrosion rates in the 0.04 to 0.08 mpy range (0.001-0.002 mm/yr). In Figure 3, the SRB levels rise to a constant 1-9 colonies/mL in the untreated fluid, whereas they remain zero in the treated cylinder. In quiescent conditions such as these, SRB colonies will attach to the pipe wall rather than float freely in the water, so low populations in the test cylinder water would be expected. A measure of bacterial activity can be gained from Figure 2 which shows a plot of soluble iron concentration in the water versus time. In the treated cylinder, the iron level remains relatively constant at 10-20 ppm; however, in the untreated cylinder, the soluble iron concentration rises initially up to 25 ppm and then falls to below 2 ppm. This is caused by the precipitation of insoluble iron sulfide, which is a result of dissolved iron reacting with hydrogen sulfide produced by SRB growth.



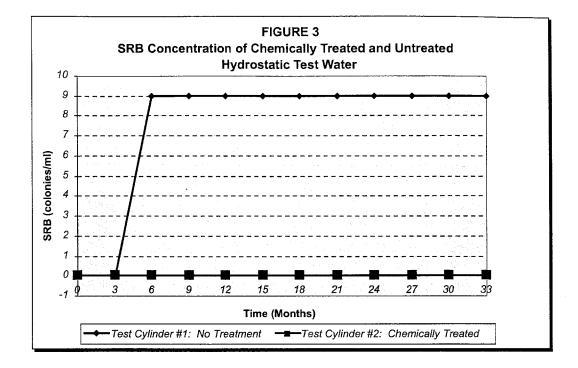












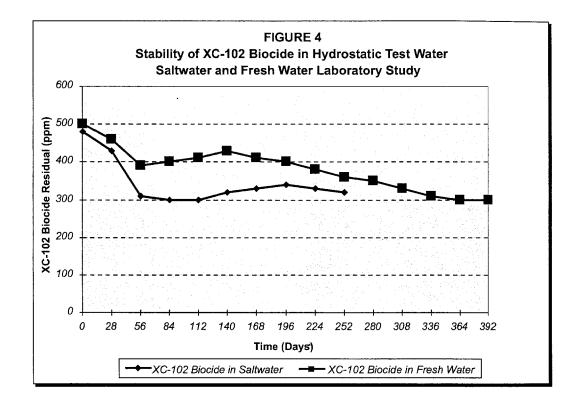
The time required for oxygen depletion to approximately 100 ppb in both fresh and sea water in a range of pipe diameters is shown in Table 1. The oxygen in the largest diameter (10" or 250 mm) cylinder was depleted in 48 hours. Metal coupons suspended in the water in the cylinders were examined after 4 months for signs of oxygen attack. No evidence of pitting was observed.

| TABLE 1<br>Oxygen Depletion in Water-Filled Pipelines |        |       |                |          |  |
|---|--------|-------|----------------|----------|--|
| Line  | Size   | Water | Initial        | Hours to |  |
| MM.   | Inches | Туре  | 0 <sub>2</sub> | 100 ppb  |  |
|   |        |       | (ppm)          |          |  |
| 250   | 10     | fresh | 7.2            | 48       |  |
| 100   | 4      | fresh | 6.8            | 26       |  |
| 50  | 2      | fresh | 7.0            | 20       |  |
| 250   | 10     | sea   | 4.5            | 48       |  |
| 100   | 4      | sea   | 4.0            | 30       |  |
| 50  | 2      | sea   | 4.5            | 18       |  |
| Temperature 20 - 22°C                                 |        |       |                |          |  |





The results of biocide stability tests for X-CIDE 102 biocide are shown in Figure 4. The concentration of X-CIDE 102 biocide falls from 500 ppm and stabilizes at about 300 ppm in both fresh water and sea water.



6





# CRW132 Corrosion Inhibitor

#### DESCRIPTION:

**Baker Petrolite** 

영상 이 가슴 옷을 다.

CRW132 is a water-soluble blend of filming amines, surfactant, and oxygen scavenger. It is an excellent packer fluid inhibitor as well as a hydrostatic test and general waterflood inhibitor.

#### APPLICATION:

Applications vary with specific system conditions. Contact your local Baker Petrolite products representative for advice on your system.

Your Baker Petrolite representative can evaluate your system's performance, specify the appropriate treatment and equipment, and design a comprehensive application program.

## TYPICAL PROPERTIES:

Specific Gravity, 77°F(25°C) Specific Weight, 77°F(25°C) Flash Point, PMCC Pour Point Solubility (brine) Solubility (water) 0.97 8.07 lbs/US gal 88°F(31°C) -40°F(-40°C) Soluble Soluble

#### FEATURES AND BENEFITS:

#### Feature:

• Combination formula

#### Benefit:

Minimizes product inventory

#### Feature:

• Residuals easily monitored

#### Benefit:

Treatment cost minimized

#### Feature:

• Excellent cold weather handling properties **Benefit:** 

· Minimal storage and pumping requirements

### MATERIAL COMPATIBILITY:

#### Suitable:

| Metals:      | admiralty brass, copper, 304 stainless |
|--------------|--|
|              | steel, 316 stainless steel             |
| Plastics:    | PLEXIGLAS, HD polyethylene,            |
|              | HD polypropylene, PVC                  |
| Elastomers:  | fiberglass, TEFLON                     |
| Not Suitable | •                                      |
| Metals:      | -aluminum, mild steel                  |
| Plastics:    |  |
| Elastomers:  | Buna N (rubber), neoprene,             |
|              | HYPALON, VITON                         |

#### SAFETY AND HANDLING:

Before handling, storage or use, see the Material Safety Data Sheet (MSDS) for details.

Baker Petrolite 24 Hour Emergency Hotline: 1-800-424-9300 (CHEMTREC) U.S.A. 1-613-996-6666 (CANUTEC) Canada Baker Petrolite Customer Care Hotline: 1-800-872-1916 (8 a.m. to 5 p.m. CST)

Disclaimer of Liability: Baker Petrolite Corporation (BPC) warrants to purchaser, but no third parties or others, the specifications for the product shall fall within a generally recognized range for typical physical properties established by BPC when the product departs BPC's point of origin and that any services shall only be performed in accordance with applicable written work documents. BPC MAKES NO OTHER WARRANTY OR GUARANTEE OF ANY KIND, EXPRESS OR IMPLIED, INCLUDING NO IMPLIED WARRANTY OF MERCHANTABILITY OR FITNESS FOR A PARTICULAR PURPOSE, REGARDING ANY SERVICES PERFORMED OR PRODUCT SUPPLIED. BPC will give purchase, but does not guarantee the accuracy or correctness of such interpretations. BPC's recommendations contained herein are advisory only and without representations as to the results. BPC shall not be liable for any indirect, special, punitive, exemplary or consequential damages or losses from any cause whatsoever including but not limited to its negligence.





# CRW9070 Corrosion Inhibitor

#### **DESCRIPTION:**

회사님 소문 경험 것이

CRW9070 corrosion inhibitor is an amine based corrosion inhibitor which can be used to treat oil wells, water injection systems and packer fluids. It is soluble in fresh water and brines up to 12.0 pounds per gallon. CRW9070 provides protection from corrosion caused by both  $CO_2$  and  $H_2S$ .

#### APPLICATION:

CRW9070 corrosion inhibitor should be applied via continuous injection. A concentration of 10-50 ppm in the produced fluids is sufficient in most applications. The optimum rate needed should be based on the data obtained from the monitoring program.

For packer fluids, 0.5-2.0% should be mixed into the brine prior to injection into the annulus.

Your Baker Petrolite representative can evaluate your system's performance, specify the appropriate treatment and equipment, and design a comprehensive application program.

#### TYPICAL PROPERTIES:

| Form                    | Liquid          |
|-------------------------|-----------------|
| Specific Gravity @ 72°F | 0.924           |
| Specific Weight @ 72°F  | 7.70 lbs/US gal |
| Flash Point             | 62°F            |
| Pour Point              | -35°F           |
| Solubility              | Water soluble   |
| pH                      | 9.0-12.0        |

#### FEATURES AND BENEFITS:

#### Feature:

- Thermally stable
- Benefit:
  - Effective in hot wells

#### Feature:

- · Excellent brine solubility
- Benefit:
  - Mixes easily with packer fluids

#### Feature:

• Very water soluble

Benefit:

• Particularly effective in high fluid wells

#### Feature:

· Detergent properties

#### **Benefit:**

· Helps prevent under deposit corrosion

#### MATERIAL COMPATIBILITY:

#### Suitable:

| Metals:     | admiralty brass, aluminum, copper, |
|-------------|------------------------------------|
|             | mild steel, 304 stainless steel,   |
|             | 316 stainless steel                |
| Plastics:   | HD polyethylene                    |
| Elastomers: | TEFLON, VITON                      |

#### Not Suitable:

| Metals:     |                                       |
|-------------|---------------------------------------|
| Plastics:   | HD polypropylene, fiberglass          |
| Elastomers: | BUNA N (rubber), neoprene,<br>HYPALON |

#### SAFETY AND HANDLING:

Before handling, storage or use, see the Material Safety Data Sheet (MSDS) for details.

Baker Petrolite 24 Hour Emergency Hotline: 1-800-424-9300 (CHEMTREC) U.S.A. 1-613-996-6666 (CANUTEC) Canada Baker Petrolite Customer Care Hotline: 1-800-872-1916 (8 a.m. to 5 p.m. CST)

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| To: | SCOTIA INSTRUMENTS<br>0D 1 281 873 5344 | Attention: Robert MacLeod |               |
|-----|---|---------------------------|---------------|
| Fac |   | Pages: 18                 | Ref: sc327/sc |
| CC: |   | Fax                       | ·             |
| Rei | Gulf of Mexico                          | Date: August 31           | , 1999        |
|     |   |                           |               |

Further to our recent telephone conversation, I can now confirm the following information.

#### Environmental Information

Champion Technologies has reviewed the regulations for the use of chemicals in the gulf of Mexico, and can now confirm the following.

A number of U.S. operators have been previously contacted and it has been confirmed that there are no specific regulations relating to the use and discharge of chemicals in the Gulf of Mexico for 'federal' waters. A governing board or body empowered with approval and testing does not exist to directly intervene or regulate chemical use and subsequent discharge. However, there are regulation pertaining to state water (coastal waters) which are highly relevant to this project.

The Gulf of Mexico has a similar environmental position to that of the North Sea oilfield sector albeit ten years ago. It is believed that although the regulations have not been formalised, there are some general guidelines that Champion has learned about from our recent discussions with environmental regulatory bodies.

Champion Technologies has been advised to ensure that all products possess a full environmental data set relating to the Bioavailability, Bioaccumulation, Biodegradation and the toxicity information for Skeletonema, Corophium and Acartia. This is essentially a product having a full "HOCNF Format".

All products that have been quoted within this document have a full HOCNF Format data set.

It is known that oil soluble corrosion inhibitors are not allowed except where a corrosion inhibitor maybe be considered based on the pipeline protection period required.

The UK North Sea, Norwegian North Sea, Danish and Dutch oilfield sectors have the most sophisticated and structured environmental regulatory body in the world. Champion Technologies design, manufacture and supply in accordance with these guidelines and therefore has a full database of all the necessary environmental information for all its products. This position has ensured that Champion has the environmental technology and knowledge in order to supply to all oilfield sectors on a global basis. <u>Champion will revert with full information as to whether the chemical speicfications detailed below will be required to be revised, once clarification is sought for chemical discharge in to federal waters.</u>

Champion is very aware of the effect that chemicals can have on marine life and the environment as a whole. We take our responsibility as a chemical supplier very seriously and for a number of years an Environmental Policy has been an integral part of our corporate ideology.

The Blacksmith range of hydrotesting chemicals has both fully toxicity tested by the UK authorities under the Offshore Chemical Notification Scheme (OCNS). This category system reflects the toxicological properties, environmental impact and usage volume of chemicals offshore in the UK sector.

In January 1996, the new system of Harmonised Offshore Chemical Notification Format came into force. This scheme standardises the methods for testing, evaluation and approval of offshore exploration and production chemicals throughout the entire North East Atlantic Sector. This revised notification scheme supercedes the voluntary OCNS but is sufficiently similar that the old classifications will remain valid for a given period. The periods for which these will remain valid are as follows:

Category 4 : Until 1<sup>st</sup> June 1997

Category 3 : Until 1<sup>st</sup> January 1998

Category 2 : Until 1\* January 1998

Category 1 : Until 1\* January 1999

Category 0 : Until 1<sup>e</sup> January 2000

In addition, the revised prior notification tonnage triggers have also been amended. These are now set for the cumulative quantity of all chemicals <u>used</u> within each group at individual installations. This differs from the old OCNS for which the tonnage triggers were based upon the <u>discharge</u> of individual chemicals.

The UK Department and Trade and Industry (DTI) regulate the environmental classification of chemicals for offshore use. In doing so, each chemical is awarded a category based on it's toxicity profile. Summarised below are a selection of low toxicity packages.

| Product                          | Application         | HOCNF Category*           |
|----------------------------------|---------------------|---------------------------|
| Champion B1150<br>(Bactron K-54) | Biocide             | C                         |
| Champion B1710                   | Biocide             | D                         |
| Champion OS2<br>(Cortron RU-206) | Oxygen Scavenger    | E<br>(OS Parcom List 'A') |
| Champion CP1900                  | Corrosion Inhibitor | B                         |
| Champion Fluorescein Dye         | Leak Detection Dye  | D                         |
| Champion Cleardye*               | Leak Detection Dye  | E                         |

Page 2





## BLACKSMITH B1150 Product Data Sheet

#### Product Description

Blacksmith B1150 is a highly effective blocide used to control microbiological problems in both land based and offshore systems. Chemically, it comprises of a 50% solution of Glutaraldehyde.

#### **Product Application**

Blacksmith B1150 is an excellent non-selective biocide for the destruction of the major bacterial strains and is also effective against some fungi and algae.

Blacksmith B1150 is water miscible liquid which forms clear solutions in both fresh water and concentrated brines.

This product is one of the most environmentally acceptable blocides on the market.

#### Chemical & Physical Properties

| Form:                      | Lîquid                         |
|----------------------------|--------------------------------|
| Colour.                    | Clear, colourless/light yellow |
| Odour                      | Pungent                        |
| pH (20degC):               | 3-4                            |
| Boiling Water:             | 95degC approx.                 |
| Water Solubility;          | Completely miscible            |
| Relative Density (20degC): | 1.113                          |
| Viscosity (20degC):        | 20mPa.s                        |

#### Dosage

Blacksmith B1150 should be injected neat into the system at a dosage rate of 75-200ppm, for batch fill testing operations. Champion will be able to advise on the optimum concentration subject to systems conditions.

#### Environmental Information

Blacksmith B1150 is an environmentally friendly combined product and has been awarded an HOCNF Category of 1[C].

This product is also approved by the relevant authorities for use in the Danish, Dutch and Norwegian Sectors of the North Sea.

Champion Technologies Abbotavell Road, West Tullos, Aberdean AB12 3AD. Tel: 01224 879022 Fax: 01224 876022 Telex 73171 Registered Office: Abbotswell Road, West Tullos, Aberdeen, AB12 3AD. Registered in Scotland No. 165529



### BLACKSMITH OS2 Product Data Sheet

#### Product Description

Blacksmith OS2 is an aqueous solution of Ammonium Bisulphite (63-65%). This product has been designed to rapidly scavenge dissolved oxygen from seawater at normal temperatures for pipeline operations and water injection systems.

This product should be injected neat into the treatment solution with the minimum exposure to air.

### Product Application

Blacksmith OS2 should be injected into the treatment solution with the minimum exposure to air. For hydrotest applications it is important that Blacksmith OS2 is added to the test medium before the injection point of other hydrotesting chemicals. This product is one of the most economical methods for reducing the dissolved oxygen content to below 10ppb.

#### Chemical & Physical Properties

| Form:                    | Liquid              |  |
|--------------------------|---------------------|--|
| Colour.                  | Clear, light yellow |  |
| Odour:                   | Pungent sulphurous  |  |
| рН (20⁰С):               | 4.8-5.6             |  |
| Boiling Point            | 105°C approx.       |  |
| Water Solubility:        | Completely miscible |  |
| Relative Density (20°C): | : 1.32-1.40         |  |

### Dosage

Blacksmith OS2 should be injected at 155ppm as this dosage level will ensure a rapid rate of oxygen depletion.

### Environmental Information

Blacksmith OS2 is an environmentally friendly combined product and has been awarded an HOCNF Category of E (OSPARCOM List A).

Champion Technologies Abbotswell Road, West Tulloa, Aberteen AB12 3AD. Tel: 01224 879022 Fax: 01224 876022 Telax 73171 Registered Office: Abbotswell Road, West Tulloa, Aberteen, AB12 3AD. Registered in Scotland No. 166529



### BLACKSMITH CP1900 Product Data Sheet

#### Product Description

Blacksmith CP1900 is a highly formulated film-forming imadazoline salt based corrosion inhibitor designed to provide corrosion protection for hydrotest operations.

This is achieved by either a continuous injection operation during pipeline flooding or as a batch treatment program, prior to pipeline start-up.

#### Product Application

Blacksmith CP1900 is a highly active corrosion inhibitor which effectively forms a protective barrier between the test medium and the walls of the pipeline. This product was formulated in order to have a greater environmental acceptability whilst providing enhanced corrosion protection at a low dosage level. Blacksmith CP1900 is completely miscible in fresh water and salt water mediums and is active over a wide pH range.

#### Chemical & Physical Properties

| Fornt.                   | Liquid      |
|--------------------------|-------------|
| Colour                   | Clear amber |
| Flash Point              | >65*C       |
| Relative Density (20°C): | 0.996       |
| Viscosity (25°C);        | 5cP,        |

#### Dosage

Blacksmith CP1900 should be inject neat into the system, if used for continuous inject at a dosage rate of 100-200ppm. Dosage levels for batch treatment programmes are subject to system conditions and therefore a Champion representative can assist with dosage recommendations.

#### Environmental Information

Blacksmith CP1900 has been awarded an HOCNF Category B.



## BLACKSMITH FLUORESCEIN DYE

Product Data Sheet

### Product Description

Blacksmith Fluorescein Dye can be supplied in solid or liquid form although for hydrotesting operations the liquid form is generally favoured. Chemically, it is the sodium salt of hydroxy-o-carbonyl phenyl fluorene and has a dark orange appearance in the concentrate form.

### Product Application

Blacksmith Fluorescein Dye exhibits an intense green colour upon dilution and is generally detected by UV light at 491nm making it an excellent tracer dye for use in leak detection. Blacksmith Fluorescein Dye is most commonly used for hydrotest and comparisons. This product is generally regarded as the full strength industry standard.

### Chemical & Physical Properties

26

| Form:                    | Liquid                                     |  |
|--------------------------|--|--|
| Colour:                  | Dark orange                                |  |
| Odour:                   | None                                       |  |
| рН (@2% in water):       | 12   |  |
| Relative Density (20°C): | 1.0 - 1.1                                  |  |
| Solubility:              | Completely soluble in fresh and sea water. |  |

### Dosage

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Blacksmith Fluorescein Dye is typically dosed in the range of 25-40ppm for hydrotest applications and 1000-4000ppm for cementing applications.



### BLACKSMITH CLEAR DYE Product Data Sheet

### Product Description

Blacksmith Clear Dye is an optically sensitive tracer dye for hydrostatic testing and can be supplied in both the solid or liquid form, although for hydrotesting operations the liquid form is generally favoured. Chemically, it is an anionic fluorescent compound which offers substantial advantages over existing dye-based tracer systems.

### Product Application

Blacksmith Clear Dye has been formulated to be used for hydrotest leak and pressure test operations for both fresh and saline water. This product operates as a readily traceable detector when activated by UV light although it is invisible under white light. Blacksmith Clear Dye offers advantages over more familiar leak test tracers such as Blacksmith Fluorescein Dye as the visible discharge can not be observed.

### Chemical & Physical Properties

| Form:               | Liquid              |
|---------------------|---------------------|
| Colour:             | Yellow Solution     |
| Odour:              | None                |
| pH:                 | 8-9.5               |
| Solubility (@20°C): | completely miscible |

### Dosage

A Champion representative can advise of the required dosage levels based on the system conditions.

Attached is the Absorption spectra for Blacksmith Clear Dye (Ref. Figure 1).

Champion Technologies Abbotsweil Rosd, West Tulics, Aberdeen AB12 3AD, Tel: 01224 879022 Fax: 01224 876022 Telex 73171 Registered Office: Abbotswell Road, West Tulios, Aberdeen, AB12 3AD. Registered in Scotland No. 198529

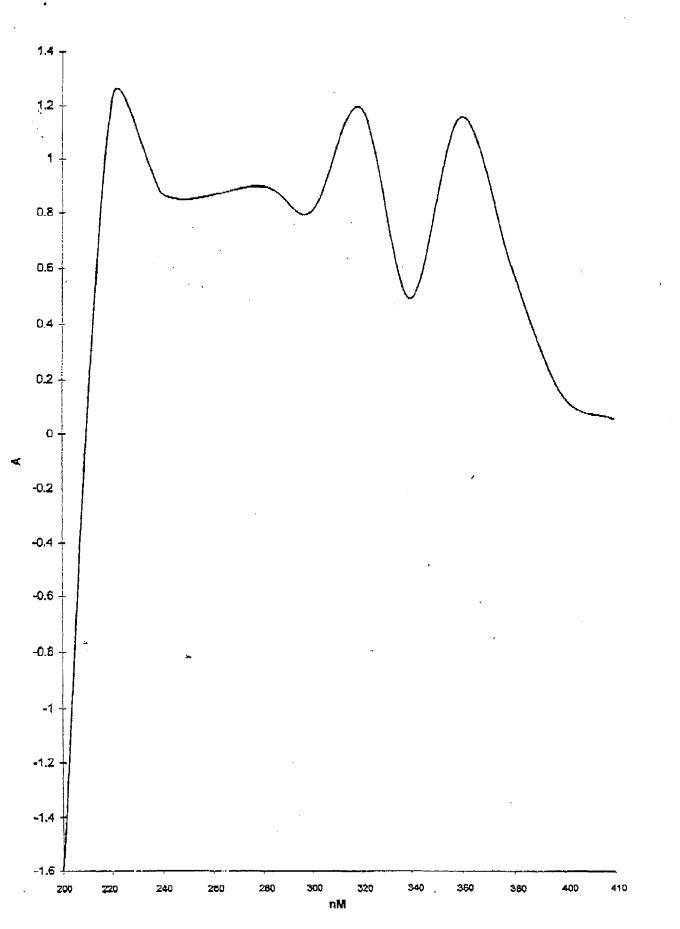


Figure 1: Absorption spectre profile for Blacksmith Clear Dye



# SUBSEA SPECIAL PRODUCTS

### **Innovative Chemical Solutions**

- Hydrotesting Corrosion Inhibition
   Leak Detection
   Wax Removal
   Solid State Chemistries
   Pipeline Bundle Protection
   Pipeline Swabbing
  - Free-flooding Pipelay Inhibition
     Hydrocarbon Decontamination
    - Subsea Tie-in ROV/Diver Intervention Descaling

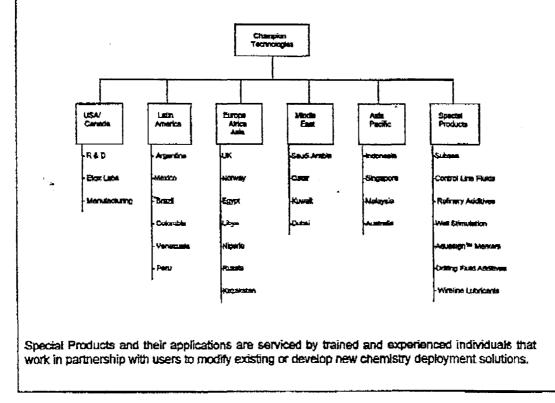
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### Company Overview

For over a decade Blacksmith has maintained its position as a leading North Sea oilfield chemicals service company whilst also developing a significant international portfolio through its Special Products range.

Following Blacksmith's acquisition in 1996 by Champion Technologies Inc, an International leader In speciality oilfield chemicals, Champion Technologies Special Products operates as a specialist division dealing primarity within the global service industry sector. Through Champion's established world wide infrastructure this group is able to service an increasing customer base with its Special Products and application technologies.

These Special Products are internationally recognised and respected and carry unparalleled track records. Many are industry standards and extensively used by the service industries leading contractors. Any Special Products material is available globally from any company location.



### Subsea Special Products

For over a decade Champion Technologies (formerly Blacksmith) has been the leader in the supply of chemical related products, services and consultancy to the North Sea Subsea and pipeline service industry.

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This experience is also exported widely and in particular Champion's strategic position in the South East Asian and Latin American subsea markets is comparable to that held in the North Sea.

A range of low toxicity chemistries available for deployment in several physical forms are customised to meet specific operational conditions has secured Champion's involvement in over 100 major pipeline commissioning and decommissioning projects achieving several industry records and notable firsts. A full track record is detailed on the back page.

Blacksmith's and now Champion's unique position of dedicating an experienced project team to the subsea service industry sector allows expert technical, operational and environmental advise on the selection/application of chemicals. Cooperation with locally experienced Champion operations means a rapid high level of assistance is always available to overseas customers.

The core application of Champion's Subsea Special Products range is corresion inhibition chemistry. Corrosion is such a vital consideration in subsea operations in that every activity where metallurgy is exposed to seswater, the potential for corrosion should be evaluated and prevented. A basic overview of the common corrosion process is enclosed within this document. In addition, throughout the past 2 years Champion has worked extensively on deepwater corrosion prevention techniques which allow chemistries to be deployed under difficult engineering and operational circumstances where the standard approaches to corrosion inhibition are not possible. However, deepwater corrosion follows a unique set of pathways, whereby an illustration of this process has been attached.

An application guide for Champion Subsea Special Products is included over. However, final selection of a Special Product and application technique usually occurs in consultation with the user and evaluates various factors, i.e. environmental, operational time and costs.

Full technical details on any application or product shall be provided on request.

| Oxygan Scavengers    | Remove dissolved oxygen from water to prevent oxygen induced corrosion.<br>Oxygen Scavengers are designed to provide a rapid rate of oxygen depiction, i.e. at 5°C, to reduce the oxygen content by 85%, well within the first minute of injection.                       | Blacksmith OS2<br>Blacksmith OS3   |
|----------------------|---|--|
| Biockles             | Remove bacteria from water to prevent microbiological induced conosion. Blocides are designed to be effective on a wide range of strains, e.g. Aerobic, Anaerobic and Sulphate Reducing Bacteria. Blofilm panetration is also adventegeous to eliminate blomass build up. | Blacksmith B1150<br>Blacksmith B1880<br>Blacksmith B1370<br>Blacksmith B1710<br>Blacksmith B1710<br>Blacksmith B1200 |
| Corrosion Inhibitors | Provide tenacious film through adsorption of the inhibitor molecules onto the metal surface.<br>The inhibitor film protects the metal from corrosion and prevents any bacterial fouling becoming directly attached to the surface.  | Blacksmith CP1300/E<br>Blacksmith CP1900<br>Blacksmith CP1620  |
| Cocktell Products    | Combination chemicals that contain one or more of the individual blocke, corrosion inhibitor and oxygen scavenger components.   | Blacksmäh O-3670R  |

Page 1 of 2

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### Novel Corresion Inhibition

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| Hydrotesting Sticks<br>for frae-flooding equipment<br><u>or</u> 'online' chemical<br>Insertion whilst subses | <ul> <li>Specific water treatment chemistries custom developed in solid form with dissolution rates designed to introduce chemistries under given operational conditions and time scales.</li> <li>Inserted into various equipment (e.g. spools/bundles/risere) prior to load out.</li> <li>Inserted into gaskets prior to seat plate changes/jumper installation.</li> <li>ROV/diver insertion for spool places/flanges and during hook up operations.</li> <li>Eliminates divers exposure to chemicals.</li> <li>Bonded for temporary fixing to equipment by water soluble adhesives providing delayed chemical deployment.</li> </ul> | Biocides Sticks<br>Oxygen Scavenger Slicks<br>Corrosion Inhibitor Sticks<br>Cocktell Sticks<br>Dye Sticks |
|--|--|---|
| Water Treatment Gels<br>for free-flooding equipment  | Specific water treatment chemistries custom developed in soluble gel form designed to introduce active<br>'corrosion inhibition' chemistry under operational conditions and time scales. Provides high level<br>localised corrosion protection.  | Biocides<br>Oxygen Scavengers<br>Carrosion Inhibitors<br>Cockialis  |
| Water Treatment Coatings<br>for pipelay applications   | Liquid inhibitor package designed to set with hard smooth finish. Used for 'online' seawater inhibition treatment of pipelines during free flooding operations. Contain water soluble corresion inhibitor, blocide, oxygen scavenger and can be applied by brush/spray onto internal pipe auriace prior to lay.  | Blocides<br>Oxygen Scavengers<br>Corrosion Inhibitors<br>Cocktails  |
| Vapour Phase Inhibitors  | Products designed to be added to the liquid test medium. On dewatering the VPI will be retained on the internal surfaces and release a corrosion inhibitor vapour which costs the metal and provides protection.   | Blacksmith VPI series   |

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| Decontamination Gels    | Remove hydrocarbons and retain contamination absorbed within structure of the moving gel. Decontaminates pipework to less than 40ppm oil in weter post flush.     | Blacksmith Musol & Gel                              |
|-------------------------|---|---|
| Swabbing Gels           | Contain dehydration solvent to act hydroscopicity on the pipeline internals. Lift and retain water<br>containingtion absorbed within structure of the moving get. | Blacksmith Melhanol Solget<br>Blacksmith MEG Solget |
| Pick Up Gels            | Water or hydrocarbon based gets designed to remove pipeline debris leaving the internal surface in<br>either a hydrophilic or hydrophobic state.                  | Blacksmith Aqua Solget<br>Blacksmith Gel Oli 10/11  |
| tsolation Gels/Gel Pigs | Highly viscous or pre-formed water or hydrocarbon based gets used in pigging and isolation operations.  | Blackemith SolgeVGefoll Series                      |

### Leak Detection

| Leak Delection Dyes        | Leak detection via visual and optical fluorescence.<br>Leak datection via optical fluorescence (chemical is colourless - no seawater colourisation). Very Low<br>Toxicity. | Blacksmith Fluoreace'n L.T<br>Blacksmith Clear Dye |
|----------------------------|--|--|
| Leak Delection Sticks/Gels | As above in stick or gel form  | As above in stick/gal                              |

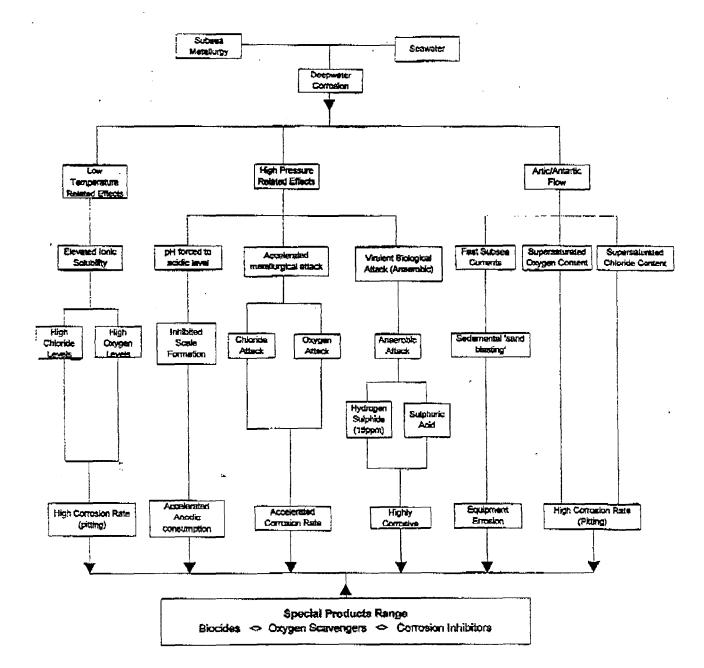
### Pipeline Blockage Removal

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### Pipeline RFO Conditioning

| Swabbing Solvenia | Glycole or methanol used in pipeline drying operations where advantageous over Nitrogen or vacuum.  | Methanol and Glycols |
|-------------------|---|----------------------|
|                   | Post dewatering an inhibitor get pig or alug of oil/gas phase correston inhibitor can be run. This conditions the pipe internal surfaces to receive fluids/gas pre-protected. | Biacksmith CP1315    |

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### OVERVIEW OF DEEPWATER CORROSION PROCESS

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